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Chitin and Chitosan Isolation, Properties, and Applications

Edited by Brajesh Kumar



Chitin and Chitosan -Isolation, Properties, and Applications

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IntechOpen Book Series Biochemistry

Volume 45

Aims and Scope of the Series

Biochemistry, the study of chemical transformations occurring within living organisms, impacts all of the life sciences, from molecular crystallography and genetics, to ecology, medicine and population biology. Biochemistry studies macromolecules - proteins, nucleic acids, carbohydrates and lipids -their building blocks, structures, functions and interactions. Much of biochemistry is devoted to enzymes, proteins that catalyze chemical reactions, enzyme structures, mechanisms of action and their roles within cells. Biochemistry also studies small signaling molecules, coenzymes, inhibitors, vitamins and hormones, which play roles in the life process. Biochemical experimentation, besides coopting the methods of classical chemistry, e.g., chromatography, adopted new techniques, e.g., X-ray diffraction, electron microscopy, NMR, radioisotopes, and developed sophisticated microbial genetic tools, e.g., auxotroph mutants and their revertants, fermentation, etc. More recently, biochemistry embraced the 'big data' omics systems. Initial biochemical studies have been exclusively analytic: dissecting, purifying and examining individual components of a biological system; in exemplary words of Efraim Racker, (1913-1991) "Don't waste clean thinking on dirty enzymes." Today, however, biochemistry is becoming more agglomerative and comprehensive, setting out to integrate and describe fully a particular biological system. The 'big data' metabolomics can define the complement of small molecules, e.g., in a soil or biofilm sample; proteomics can distinguish all the proteins comprising e.g., serum; metagenomics can identify all the genes in a complex environment e.g., the bovine rumen.

This Biochemistry Series will address both the current research on biomolecules, and the emerging trends with great promise.

Meet the Series Editor



Miroslav Blumenberg, Ph.D., was born in Subotica and received his BSc in Belgrade, Yugoslavia. He completed his Ph.D. at MIT in Organic Chemistry; he followed up his Ph.D. with two postdoctoral study periods at Stanford University. Since 1983, he has been a faculty member of the RO Perelman Department of Dermatology, NYU School of Medicine, where he is codirector of a training grant in cutaneous biology. Dr. Blumenberg's research is focused

on the epidermis, expression of keratin genes, transcription profiling, keratinocyte differentiation, inflammatory diseases and cancers, and most recently the effects of the microbiome on the skin. He has published more than 100 peer-reviewed research articles and graduated numerous Ph.D. and postdoctoral students.

Meet the Volume Editor



Dr. Brajesh Kumar is currently an assistant professor and head of the Department of Chemistry, TATA College, Kolhan University, Chaibasa, India. He received a Ph.D. in Chemistry from the University of Delhi, India, in 2009. He is a pioneering researcher in the fields of green chemistry, nano synthesis, environmental remediation, photocatalysis, nanomedicine, and polymer and natural product extraction. Dr. Kumar has several national and interna-

tional fellowships to his credit and has worked as a faculty member in various universities throughout India, Ecuador, and South Korea. He is a holder of two registered patents as well as a member of the American Chemical Society and the Indian Society of Chemists and Biologists (ISCB). He has also published more than eighty research articles and is an active reviewer of more than seventy journals. He was listed among the top 2% of scientists worldwide by Stanford University, USA, in 2021 and 2022.

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Preface

Currently, numerous biomaterials-based studies are being conducted, including research into chitin and chitosan, the second most abundant polysaccharide after cellulose. Chitin is obtained at an industrial scale from a variety of natural sources, including crustacean and insect exoskeletons, fungi cell walls, squid pens, and others. Chitosan is biodegradable, biocompatible, non-toxic, and water-soluble under acidic conditions and linear cationic aminopolysaccharide derived from the deacetylation of chitin. It contains free amino and hydroxyl groups that can be functionalized by binding with the cationic and anionic groups. It has numerous applications, especially in environmental remediation and the biomedical, pharmaceutical, agriculture, and food industries. This book addresses isolation, properties, and certain applications of chitin and chitosan, including films, fibers, nanoparticles, composite materials, hydrogels, polymeric complexes, water purification, antimicrobials, textile, cosmetics, biosensors, nanoporous scaffolds, and membranes. It is written by researchers from industry, academia, government, and private research institutions.

This book highlights the isolation, properties, and applications of chitin and chitosan. In Chapter 1, Paul Edgardo Regalado-Infante et al. cover recent studies on the biological functions of chito-oligosaccharides and their impact on a priority area such as agriculture, where these compounds could be used to substitute for chemical compounds that have generated serious health issues as well as environmental pollution. In Chapter 2, Renuka Vinothkumar and Janet Paterson focus on a new way of developing water-soluble colloidal chitin (WSCC) from prawn waste and investigate its fundamental rheological and antibacterial properties. WSCC films studied during this research may be used in food packaging or in medical applications. The use of WSCC biodegradable films will protect the environment in the future and will be an effective alternative to plastics. In Chapter 3, Mohsin Mohammed and Nadia Haj loaded chitosan with 5-fluorouracil via amide-mediated binding and evaluated the resulting product as a potential 5-fluorouracil delivery agent. The authors use 1H-13C nuclear magnetic resonance and Fourier-transform infrared and ultraviolet spectroscopy to characterize the product. The three cancer cell lines MCF-7, MDA-MB-231, and MDA-MB-453 showed a dose-dependent reduction in viability compared to the original medicine and these findings suggest that chitosan-methotrexate as a prodrug could be helpful in the treatment of colon cancer.

Nanoscience will continue to grow due to its numerous benefits, including those on human health, food processing, environmental safety, and device engineering. In recent years, several findings have been gathered indicating nano-chitosan as a potential plant health material. In Chapter 4, Pranab Dutta, Arti Kumari, and Madhusmita Mahanta outline the methods of green synthesis and characterization of chitosan nanoparticles, their utilization in plant protection and growth promotion, and the underlying mechanisms. In the coming years, the use of nano-chitosan for combating biotic and abiotic stresses and transport of agrochemicals will be a promising discipline for utility in sustainable agriculture. In Chapter 5, Magdi A.E. Abdellatef et al. highlight that chitosan is the ideal resource for plant disease management under sustainable agriculture. Chitosan is biodegradable, biocompatible, non-toxic, and has antimicrobial activity, which confers it with dual synergetic effects during host–pathogen interaction. Leishmaniasis, an infectious disease that affects humans, domesticated dogs, and wild animals, is caused by 20 of the 53 Leishmania genus species and is transmitted by sandflies. Finally, in Chapter 6, Felipe Trovalim Jordão et al. investigate the potential of the chitinase-encoding gene as a molecular diagnostic tool and a phylogenetic marker for studying basal trypanosomatid groups worldwide.

I am grateful to my loving wife Kumari Smita for her helpful comments on several chapters and excellent support in the conceptualization of the book's content.

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Chapter 1

New Perspectives on the Application of Chito-Oligosaccharides Derived from Chitin and Chitosan: A Review

Paul Edgardo Regalado-Infante, Norma Gabriela Rojas-Avelizapa, Rosalía Núñez-Pastrana, Daniel Tapia-Maruri, Andrea Margarita Rivas-Castillo, Régulo Carlos Llarena-Hernández and Luz Irene Rojas-Avelizapa

Abstract

The study of chitin and chitosan has stood out for many years due to their potential application in various areas such as the food industry, where they are either used as additives, prebiotics, or bio-conservatives; as to biomedical and pharmaceutical industries, where they function to treat diseases. Besides, in the agriculture field, it is known that they can cause a positive effect on the development of plants and optimize nitrogen fixation. In recent years, attention has been paid to their derivatives, chito-oligosaccharides which, unlike chitin and chitosan, they have different chemical characteristics, like their solubility, a characteristic that facilitates their use, contrary to chitin and chitosan. Moreover, the small size of chito-oligosaccharides can facilitate their entry into the cell. This review covers recent studies on the biological functions of chito-oligosaccharides and their impact on a priority area such as agriculture, where these compounds could be used to substitute the demand for chemical compounds that, until now, have generated serious health issues as well as environmental pollution.

Keywords: natural products, shrimp wastes, sustainable agriculture, chito-oligosaccharides, chitin and chitosan

1. Introduction

A world-class level concerning aspect is the accelerated population growth as well as an increased demand for goods and services. Against this background, it is important to define strategies that allow the supply of enough food for a population in continuous growth, through the implementation of efficient agricultural systems. Agriculture plays a key role in the development of any society; however, the various agrochemicals employed, such as biocides, growth stimulants, and fertilizers, among others, lead to several pollution issues that not only affect groundwater and soils, but also microbiota and surrounding wildlife, and of course, human health.

Therefore, it is essential to consider the implementation of products based on natural substances, whose characteristics may allow the substitution of chemical compounds in the agricultural sector. Biopolymers are one such possible solution to the problem because they are typically biodegradable materials obtained from renewable raw materials. Natural biopolymers include starch, cellulose, pectin, chitin, and chitosan, and have been part of humanity since its existence, being part of basic daily needs as fundamental as food and clothing, as well as medical materials, packaging, food additives, engineering plastics, chemicals for water treatment, among many others [1, 2]. The biopolymers known as chitin (poly-N-acetylglucosamine), and its derivative chitosan (obtained by deacetylation of chitin) have achieved a prominent place in the development of applications related to the treatment of water [3], soil moisture control [4], the intelligent release of fertilizers [5], growth stimulant [6], an inducer of defense mechanisms in plants [7], antimicrobial [8–10]. Their usefulness in this fight is such that their applications related to the control of population growth should also be considered due to the potential applications they have for the controlled release of contraceptives [11] and spermicides [12], including the expectations of creating a non-hormonal female contraceptive with no side effects [13]. One of the main biological applications of chitosan, which was recently discovered, is in the field of gene delivery, due to its ability to interact with anionic DNA [10, 14, 15].

Thus, the aim of this document is to present an exhaustive revision of the chitooligosaccharides derived mainly from chitosan and their potential application in agriculture and other areas. In recent years, many studies have investigated the effects of chito-oligosacharides on human health, for example, as immunological modulators, [16, 17] anticancer [18, 19], antidiabetic [20, 21], and antimicrobial compounds [22, 23]. Some factors that potentiate the effect of chito-oligosacharides are their lower molecular weight and higher degree of deacetylation [24, 25]. The chitooligosacharides can have antimicrobial properties, can induce plant resistance, and can stimulate plant growth as well, which makes them a promising alternative for the agricultural sector.

2. Shrimp species as a source of waste materials of industrial interest

Shrimp is one of the most important fishery resources worldwide, due to its nutritional value and high demand, especially in developed countries such as the United States of America, Japan, and the European Union [26].

Shrimp (*Penaeus* sp.) belongs to the animal kingdom, Phylum *Arthropoda*, class Crustacea, order Decapoda, and genus *Penaeus*. Like all arthropods, shrimp body is divided into three big main regions: cephalothorax, abdomen, and telson. It possesses various appendages such as antennules, antennae, mandibles, maxillae, maxillipeds, and pereiopods. This crustacean is essentially constituted by two parts: the crustacean muscular part, which corresponds to 50% of total mass, and the cephalothorax exoskeleton, including the tail, which is equivalent to the other 50% [27]. The cephalothorax is not fully exploited by man, so it is separated and broadly known as

the shrimp waste. In Mexico, until recently, these residues were thrown back into the ocean or used as a source of proteins for fattening foods. However, in countries such as Japan, Thailand, and Korea, these shrimp wastes have acquired a very important commercial relevance [28].

Shrimp represented the 2.4% of the worldwide fishery production in 2018, with 3.75 million metric tons in live weight, and an establishedpotential of 4.0 million metric tons for the year 2021 [29]. China is the leading country in shrimp production, with about 1.5 million metric tons of the total fishing (**Figure 1**).

In Latin America, Ecuador is the leading country in shrimp production, and it was expected that the production in this country reached 700,000 metric tons in the year 2021, making Ecuador the third main worldwide shrimp producer, only after China and Vietnam (**Figure 2**). Although Mexico suffered severe losses in 2013, the Mexican industry was able to recover its production in 2015. Besides, it was expected a higher growth, where 180,000 metric tons were expected to be reached by 2021 [30]. In Mexico, in terms of catch volumes, the shrimp occupied second place in the national fishery production, with a live weight volume of 155,281 tons in 2018 (**Figure 3**).

In recent years there has occurred overexploitation in all the commercial fishing species, including shrimp, because of uncontrolled growth fishery activities, either artisanal or industrial. Due to this problem, preventive measures have been implemented, including a temporally closed season in 2003, in which it was prohibited any fishing of shrimp species; this closed season was extended until 2017. Besides, it was also implemented a specific fishing season under the Mexican Standard NOM-009-PESC-1993. The productions of aquaculture farms show an exponential growth tendency, while open sea mats and bay fishery have kept a constant behavior (**Figure 4**).



Figure 1. Aquaculture shrimp production of the main producing countries. Sources: FAO [29] and Anderson et al. [30].



Figure 2.

Aquaculture shrimp production in the main producing countries of Latin America during 2013–2017 and projection for 2018–2021. Source: FAO [29] and Anderson et al. [30].



Figure 3.

Contribution of the main commercial species in the volume of national aquaculture production live weight in 2018. Adapted from the Mexican yearbook of aquaculture and fisheries, CONAPESCA [31].



Figure 4.

Mexican live weight shrimp production by origin 2009–2018. Adapted from the Mexican yearbook of aquaculture and fisheries, CONAPESCA [31].

2.1 Current situation: the birth of Mexican industries based on the harnessing of shrimp wastes

Chitin and its derivates have commercial use in different countries, mainly Japan, China, and the USA. All these countries already have a consolidated industry around this polymer, being China the main producer and exporter of chitin. As to Chile and Spain, both countries are in an early stage, like the rest of Latin America. In Mexico, the interest in the chitin sector, as well as in chitosan, and their derivatives, is relatively new; therefore, both academic institutions and industries have manifested their interest in this topic. For example, scientists from Centro de Investigación en Alimentación y Desarrollo AC (CIAD) in Sonora, México [32], have informed that the head and exoskeleton of shrimp should be consider not as a waste but as source for chitin and chitosan production. Additionally, there are enterprises such as "Neptuno", where researchers and businessmen collaborate to produce these polysaccharides and their derivates. Until now, they have actively participated in conferences and presentations aiming to awaken the interest in this field. Another enterprise called "Polímeros acuícolas", from Guasave, Sinaloa, pretends to exploit shrimp wastes to produce substances of interest to satisfy the global need. Furthermore, some well-known Mexican enterprises, like Resistol and Comex, have already created links with the former company.

2.2 Chemical composition of shrimp wastes

Shrimp obtained from fishery is destined for direct human consumption; it can be found in different presentations that require diverse industrial processes, or also can

Author	[35]	[36]		
% Water	7.87	4		
% Dry matter	92.13	96		
% Crude fiber	26.89	N/D		
% Crude protein	34.5	46.3		
& Fat	5.14	9.04		
% Ashes	25.60	17.04		
% Calcium	16.69	7.0		
% Phosphates	0.85	3.03		
Chitin energetic content	938*	2500		
Chitin	18.7	9.82		
*Calculated by the method of Schaibel (1980).				

Table 1.

Proximal analysis of the shrimp waste showing its content of chitin and energy (kcal/kg).

be consumed fresh or frozen. Considering that almost 50% of the crustacean weight is thrown before their consumption, it was calculated that the production of shrimp wastes in México will reach 79,138 tons in 2018 [31].

The crustacean exoskeletons are formed by successive protein-chitinous layers, with a high calcium carbonate content. Depending on the species, the chitin content can vary from 0.01 to 40% on a dry base, while protein content can fluctuate between 50 and 80%. Through X-ray diffraction studies, it has been determined that the polysaccharide chains of chitin in crustaceans, specifically in shrimp species, are tied in an antiparallel way, given a crystalline structure called type β [33]. Protein molecules adopt an antiparallel conformation of folded chains, while chitin molecules are arranged in a perpendicular way with respect to the protein chains, resulting in a tridimensional reticular structure, organized in layers with high mechanic strength. Therefore, such different protein-chitinous layers can adopt a "sandwich" structure, in whose center would be a nucleus made by chitin molecules, surrounded by layers of fibrous proteins arranged transversally [34]. In addition, it is important to emphasize that there are factors that can produce significant changes in the percentual composition of shrimp wastes, such as the crustacean species, season and geographical region of capture, and the storage of samplesas well. In general, there can be considered that such wastes contain an average of 34.5% protein, 26.89% crude fiber, and 25.60% ashes. More detailed data is shown in **Table 1**, where can also be seen that they possess a high calcium carbonate content and various phosphates. Therefore, these kinds of wastes may represent one of the possible non-conventional protein sources for animals, with a good nutritional potential, because such proteins are especially rich in lysine, which could equilibrate diets based on cereals [37].

3. Chemical structure of chitin, chitosan, and their derivative chito-oligosaccharides

3.1 Chemical structure and properties

Chitin is a non-linear polymer constituted by units of N-acetyl-2-amino-2-deoxy-D-glucose (N-acetyl-glucosamine or NAG), joined by glycosidic bonds β (1 \rightarrow 4) [38, 39].

The conformation of such glycosidic bonds produces an alternative spatial location of the N-acetyl groups along the polymer chain, where the N-acetyl-glucosamine dimer chitobiose can be considered as the minimal structural repeating unit of chitin (**Figure 5**) [34]. On the other hand, chitin is one of the most abundant natural polymers, like cellulose and hemicellulose [40]. In nature, chitin is forming cover structures of arthropods, insects, arachnids, mussels, fungi, and some algae [41].

Concerning chitosan, this is also a lineal polysaccharide that could be obtained after the extensive deacetylation of chitin and, therefore, it is composed of two different aminated monosaccharides, which are randomly placed along the polymeric chain. Such monosaccharides are amino sugars NAG and D-glucosamine (GA), which are linked, likewise, by glycosidic bonds β (1 \rightarrow 4), (**Figure 6**). It is important to point out that the total deacetylation of chitin is a quite complicated process, and therefore is possible to generally obtain mixtures of chitosans with different degrees of deacetylation (generally higher than 45%); thus, the criteria used to differentiate them is mainly their solubility in aqueous diluted acidic solutions [38].

3.2 Chitosan depolymerization to produce chito-oligosaccharides

Due to the very high molecular weight of chitosan, and its high viscosity as well, the use of this polymer becomes difficult for some applications. Therefore, this problem was solved by using the resulting products of its hydrolysis. However, such substances could be harder to obtain in the amounts required for large-scale industrial processes. In this regard, it has been reported that the hydrolysis or depolymerization of chitosan can be done through different methods: physical (ultrasound), chemical (hydrolytic reactions), or biological (using hydrolases). Among them, chemical hydrolysis is the most used at an industrial scale.

3.2.1 Chemical obtention of chito-oligosaccharides from chitosan

Chitosan can be hydrolyzed chemically either through acidic depolymerization or by an oxidative-reductive treatment. The acidic depolymerization is carried out by using a variety of chemicals such as hydrochloric, hydrofluoric, nitrous, sulfuric, and acetic acids. However, the use of such chemicals brings disadvantages, as their low yields obtained. Although they are relatively fast and cheap processes, they are inconvenient for their commercialization due to the production of toxic compounds and their considerable risk for the environment, since the materials used are highly residual [38]. There are multiple reports of chitosan oligomers obtained in this way and used alternatively in agriculture [43–46].



Figure 5. *Chemical structure of chitin.*



Figure 6. Comparative chemical structures of chitin and chitosan [42].

3.2.2 Enzymatic obtention of chitosan oligosaccharides

Such enzymatic processes are carried out generally in discontinuous reactors and are preferred over the chemical methods, due to the reduction of adverse factors. These processes require specific enzymes like chitosanases or some less specific ones like cellulases, lipases, hemicellulases, and pectinases [47]. Chitosanases are enzymes broadly distributed in nature, which are capable of degrading chitosan into low molecular weight oligomers. These enzymes have been found in bacteria, viruses, fungi, and plants [48, 49]. However, there is a limitation in the use of these specific enzymes, because of their high cost and low availability in high amounts [50]. Due to this, some researchers have explored the use of non-specific commercial enzymes, which showed to be able to degrade chitosan, almost with the same efficiency than chitosanases, but cheaper [51]. As occurred with specific chitosanases, hydrolases are also able to catalyze the breakdown of β - (1,4)-glycosidic bonds present in chitosan. These enzymes have been found in microorganisms like viruses, bacteria, and fungi. During a recent research performed by Olicón-Hernández et al. [52], the extracellular chitosanase from Bacillus thuringiensis, grown in a chitosan containing medium, was used as a crude enzyme previously sterilized by filtration to produce a mixture of mono-, di-, tri-, and tetra-saccharides using colloidized chitosan as the substrate. These results are encouraging because they show that it is possible to transform chitosan obtained from shrimp wastes, through a microbial process, into products with biotechnological importance, avoiding the traditional use of chemical substances, as different acids.

4. Chitin and chitosan uses

Chitin can be used in a variety of fields. For instance, it was studied as a wound healing and blood thinner for medical purposes. It was also used as stationary

support for the enzyme immobilization in column chromatography, due to its gelling and adhesive properties. In pharmacy is used as an excipient and dispenser of drugs. Besides, chitin has applications as an adhesive in textile and paper industries, and in agriculture for soil improvement. Due to its chelating properties, chitosan is used in water treatment and also for the decontamination of effluents. On the other hand, chitosan possesses physicochemical, functional, and biological properties, being useful in different fields such as medicine, pharmacy, agriculture, and food industry, among others. Chitosan has a high capacity for the sequestration of metallic ions, which is useful for the decontamination of industrial wastewater. Its polycationic nature grants a flocculant action, being also a good support for enzyme and cell immobilization, both in biotechnology and food industry [53].

Chitosan is also an excellent former of fibers, films, and membranes [54], and can be used to prepare microspheres or microcapsules; these capabilities, along with its biocompatibility and biodegradability, allow its use in both biomedical and pharmaceutic industries [55]. Also, it has been studied the use of chitosan as an excipient, to propitiate the controlled release of drugs and to reduce the cholesterol levels [56] and as a boost for the immune system and for the elaboration of the gels used in cosmetology [57]. Likewise, it has been described its antimicrobial action against pathogens and microorganisms that damage fruits and vegetables; this activity has been explained by supposing changes in the permeability of cells, due to the interactions between chitosan (a polycation) with the electronegative charges placed on the cell surface. Other therapeutical use concerns weight, high cholesterol, and burns controls [58]. Other applications, based on the polar capability of the hydroxyl and carboxyl groups of chitosan, have been proposed to make bio-electro sensors able to detect cancerogenic cells and at the same time being useful to administer antitumor agents to specific cells [59].

5. Some uses of chito-oligosaccharides

As indicated above, it is possible to obtain low-molecular-weight derivates of the above-cited polymers, by chemical or enzymatic treatments, which are known as chito-oligosaccharides, whose structure main contain scarcely 3 to 10 monosaccharide units. However, some authors consider that such chito-oligosaccharides may contain until 20 monosaccharide units in their chains. These compounds, particularly those derived from chitosan hydrolysis, can be used in a variety of biotechnological areas, standing out in medicine and pharmacy, due to their beneficial effects on human health, as is described hereafter [52].

5.1 Medical uses

Chito-oligosaccharides from chitosan confer an immunological modular effect, because theyit boost the immune system through cellular proliferation, besides possessing other stimulative immunological effects [57, 60–63]. In this particular instance, it is known that chito-oligosaccharides accelerate theformation of antibodies and also induce the cellular differentiation of leukocytes. Oligosaccharides from chitosan also possess an accelerating effect on intestinal transit since they help the proliferation of *Bifidobacterium* and *Lactobacillus* cells present in the intestinal flora [64].

Another important finding refers to their anti-tumoral activity, as it is believed that they can suppress and prevent cancer [65]. As metabolic emulators chitooligosaccharides have anti-cholesterolemic effects, because they are able to reduce cholesterol, triglycerides and glucose levels in the blood; besides they reduce blood pressure and have anti-obesity effects.

As food additives, chito-oligosaccharides are used as dietary fiber, and dietary supplement for poultry species and livestock [63]. Other relevant applications for these compounds involve their use in arthritis control and as an antidiabetic; also in the treatment of gastric ulcers, as antimutages, anti-inflammatories and as low caloric sweeteners [66, 67].

On the other hand, it is important to point out that there are two factors that should be considered among the most relevant, related to the biological activity of chito-oligosaccharides; one is the length of their chains and the other is their deacetylation degree [68]. In general, the longer chito-oligosaccharides are, the stronger effects they may have. However, this does not imply that the shortest oligosaccharides do not possess similar (or other different) biological effects. For example, it has been observed that penta-, hexaand hepta-glucosamines present the strongest and more varied biological effects [69].

5.2 Agricultural applications

It is known that chitin, and its derivates, have a broad range of biological activities which include antioxidant and, antimicrobial effects, and other properties that can be used on an industrial scale as well. Chitosan has been used for seed coverings, with the objective of controlling plagues and improving plant defense system against microorganisms [70]. It has been demonstrated that chito-oligosaccharides have diverse effects on plant cultivation and, on the enhancement of plant growth and development, besides improving both the quality and yield of the vegetable products [71]. In a study made by Mahdavi & Rahimi [70], it was tested the effect of chito-oligosaccharides in stimulation, specifically for germination and growth, of *Trachyspermum ammi*, observing that its growth was accompanied by a decrease in the damaging impact caused by abiotic stress like high salinity. Likewise, a study using three fractions of chito-oligosaccharides of different molecular weights, obtained from the same initial chitosan sample, clearly demonstrated that the fraction with the lowest molecular weight produced a higher acceleration in the germination of zucchini seeds covered with these compounds. In another study reported by Zou *et al.* [72], it was observed the significant benefits for soy yields, seed germination, and plant growth. Other uses of chito-oligosaccharides in the agricultural sector are as agents to conserve seeds [73], as plant defense enhancers and for the protection against plagues and diseases [74].

5.2.1 Effects on plant resistance to diseases

Jia *et al.* [75] used *Arabidopsis* plants which were pre-treated with 50 mg/L of chito-oligosaccharides per day, before their inoculation with the tobacco mosaic virus (TMV); it was found that the expression of defenses, associated with genes related to pathogenicity, resulted strengthened.

In another study, the efficiency of chito-oligosaccharides to prevent and control the southern rice striped black dwarf virus was demonstrated, evidencing that these compounds regulate the increase of proteins related to plant defenses. Also, a field test has been carried out after treatment with chito-oligosaccharides as an antifungal agent in grape plantations, where results showed that the mortality and infection

rates were reduced significatively on inoculated plants with pathogens such as *Diplodiaseriata* y *Phaeomoniella chlamydospora* [76].

5.2.2 Effects on growth and plant development

The use of oligosaccharides can help to improve plant growth, seed germination, chlorophyll content, nitrogen fixation, and nutrient absorption. Oligomers from chitosan not only have the property of inducing resistance, but they can promote plant growth and development. During their interaction with tobacco cells, these compounds regulate concentrations of indole-3-acetic acid (IAA) and their related peroxidases, which indicates a growth accelerator effect. Also, field tests have been done to evaluate oregano growth using different oligosaccharide doses (50–1000 ppm),

Species	Properties of QOS*	Concentration used	Mode of use	Effects/ observations	Reference
Tomato	_	25, 50, 75, 100 mg/L	Foliar	Plants height, leaves number, fruit yield	[77]
Wheat	93% GD	15 mg/L	Foliar	Chlorophyll, activity PEPC, saccharose content, TCA cycle	[78]
Mint	90% GD	40, 60, 120, 160 mg/L	Foliar spray	Longer root length	[79]
Lemmon grass	90% GD	40, 60, 80, 100 mg/L	Foliar spray	Root length	[80]
Chili	91.4% GD; PM 8 kDa	50 mg/L	Foliar	Increase on chlorophyll content	[81]
Wheat	95% GD	0.01%	Foliar	Sprout length, root length, chlorophyll content, sugar content	[72]
Bean	85% GD	100 mg/L	Foliar	Plant height, more vanilla pods, seeds yield	[82]
Barley	Different PM	_	Plant tissue culture	Higher growth and seed yield	[83]
Lavender	80% GD; PM 16 kDa	30, 40, 100 mg/L	Plant tissue culture	Higher plant length, multiplication, and sprout weight.	[84]
Curly kale	_	10, 50, 100 mg/L	Hydroponic cultivation	Enhance of growth and reduction for harvesting time	[84]
Soy, wheat and rice	_	10–100 mg/L	Hydroponic cultivation	They promote growth under stress	[85]

Table 2.

Some reports of the effects of oligosaccharides on the physiological attributes of plants.

where results indicated that, from 200 to 500 ppm chito-oligosaccharides, there was an increase in plant height, while doses from 50 to 200 ppm significantly regulated the concentration of polyphenols.

5.2.3 Effects on the quality of vegetable products

Studies carried out in this regard showed that treatment with chito-oligosaccharides can significantly improve the quality of strawberry plants; in such research a treatment with 50 mg/L of these chito-oligosaccharides, applied to the fruits prior to harvest them, increased their pulp viscosity, lignin content, sugar, protein, and titratable acidity; besides, they strengthen the strawberry antioxidant capability due to a higher production of components such as anthocyanins, total phenols, flavonoids, and vitamin C. **Table 2** shows some works concerning the effects of chito-oligosaccharides on diverse physiological attributes of plants.

6. Conclusions

Sustainable agriculture is a relevant topic considering the growing population worldwide and the severe damage that chemicals have caused to the environment due to their use for agricultural purposes. Many agrochemicals used nowadays are costly in the global market; however, they promote agricultural production. There is well known that many of these compounds used to protect crops against diseases and to increase yields, are considered pollutants of soil, crops, biological diversity, microbiota and, in addition, may cause diseases in animals and humans. Thus, there is currently an enormous need to promote healthier ecosystems and support sustainable soil management to minimize the use of these harmful synthetic agrochemicals, while promoting the development of methods and products for the control of pests and diseases, that could be more respectful to the environment. The utilization of protein-chitinous residues, which are abundant in our country (and without the correct management, may also generate pollution to the environment), for the obtention of derivates like chitin, chitosan, and chito-oligosaccharides from chitosan, among others, are considered valuable in many fields, due to their properties like biocompatibility, biodegradability, and null toxicity. Therefore, they are an alternative to conventional agriculture, being a feasible alternative to enhance productivity and crop protection. To that effect, chito-oligosaccharides may fulfill environmental and health requirements to help meet the needs of a constantly growing population, which involve the production of high-quality food along with low environmental impact. Unlike chitin and chitosan, chito-oligosaccharides are soluble in water, and due to their low molecular weight, they can enter the cell and have a higher biological effectivity, not to mention their relevant capability to stimulate seed germination, plant growth, to activate resistance mechanisms in crops, antimicrobial activity, and more important, they can be obtained by microbial methods that are harmless to the environment and at a low cost.

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Conflict of interest

The authors declare no conflict of interest.

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Chapter 2

Preparation of Water-Soluble Colloidal Chitin (WSCC) from Prawn Waste and Its Characterization

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Abstract

Chitin, the shell material of prawn, is a biodegradable polymer and environmentally biocompatible with low toxicity. Chitosan is the deacetylated form of chitin, which consists of poly-D-glucosamine units with no or few N-acetyl-Dglucosamine units. Commercial applications of these natural polymers are increasing in various sectors. Therefore, in addition to the environmental benefit, it may be economical to recover chitin from prawn waste. Chitosan is soluble in various organic acids, solvents and water. The poor solubility of chitin is the major limiting factor in its use in industrial applications. Number of studies have investigated to overcome the solubility problem of chitin. This research focuses on a new way of developing watersoluble colloidal chitin (WSCC) from prawn waste and investigates its fundamental rheological and antibacterial properties. WSCC films studied during this research may be used in food packaging or in medical applications. The use of WSCC biodegradable films will protect the environment in the future and will be an effective alternative to plastics that threatens the environment. The antibacterial study may be applied in pharmaceutical, medical and food packaging and coating applications. This research was conducted at the University of New South Wales, Australia in 2008.

Keywords: prawn waste, water-soluble chitin, chitin characterization, chitin rheology, antibacterial

1. Introduction

Pollution of soil, air and water contributed to environmental deterioration; its control is necessary. Plastics have become part of our lives; the treatment of waste plastics has become a serious problem because of the difficulty of land reclamation and disposal by incineration [1]. Recent interests have focused mainly on biodegrad-able plastics that are biocompatible to the environment [2]. Chitin and chitosan are examples of biodegradable, biorenewable and biofunctional polymers derived from seafood processing waste [3–8]. Therefore, in addition to the environmental benefit, it may be economical to recover chitin from prawn waste. The poor solubility of chitin is

the major limiting factor in its use in industrial applications [9]. But chitosan is soluble in various organic acids, solvents and water [10]. Several studies have investigated the solubility problem of chitin. This research focuses on a new way of developing watersoluble colloidal chitin (WSCC) from prawn waste and investigates its fundamental rheological and antibacterial properties.

2. Literature review

2.1 Chitin and chitosan

Chitin is derived from the Greek word *chiton*, which means a coat of nail. Chitin is the major component of the exoskeleton of invertebrates and the cell wall of fungi and yeast [11, 12], from mushrooms [13]. Chitosan is usually obtained by alkaline or enzymatic deacetylation of chitin. The importance of chitin and chitosan has grown partly because they represent a renewable and biodegradable source of materials, and partly because of the recent increased understanding of their functionality in various applications [2, 14–25]. Usually, chitin is prepared from crustacean waste through deproteinization using alkali or enzyme, demineralization or decalcification using acid followed by decolourization using decolouring agents in order to remove the proteins, calcium and colour, respectively. The chitin thus obtained can then be deacetylated either by alkali or enzyme to produce chitosan [26]. The properties of chitin and chitosan depend on the processing conditions. Chitosan prepared from chemical and enzymatic deacetylation of chitin differs in their degree of deacetylation (%), distribution of acetyl groups, chain length and conformational structure of chitin and chitosan molecules. These factors affect the characteristics of chitin and chitosan [27].

2.2 Solubility of chitin and its derivatives in water

The most remarkable difference between chitin and chitosan is their solubility. Chitin is insoluble in almost all solvents; chitosan dissolves in almost all aqueous acids. The insolubility of chitin is the major disadvantage to its use. Solvents of chitosan are generally safe to consume, thus allowing its use in various industries including the food industry. Most solvents used for the dissolution of chitin are toxic, hence, they cannot be used in food processing applications [28]. This research investigates the preparation of a water-soluble chitin derivative rather than chitosan because there are many studies on the preparation of chitosan water-soluble derivatives.

The solubility of chitin is achieved by the destruction of the strong hydrogen bonds in chitin molecules and their reorganization form a chitin gel [29]. Various procedures to make chitin water-soluble and its characterization are reported: [17, 30–53]. The water-solubility of chitin can be obtained by structural modification and by controlling the degree of deacetylation. Such modifications affect the properties of watersoluble chitin derivatives.

The degree of deacetylation of chitin/ chitosan plays a major role in their solubility in water. Modified chitins having 50% degree of deacetylation become soluble in water [54–57]. Deacetylated chitin has 50% degree of deacetylation with tosyl, iodo, trimethylsilyl and glucosyl groups soluble in water as well as in organic solvents [58]. This research mainly investigates on preparing water-soluble chitin derivatives having the degree of deacetylation similar to that of natural chitin.

2.3 Characterization of chitin and its derivatives

Characterization of chitin derivatives is by solubility, crystallinity, viscosity, degree of deacetylation, molecular weight, mechanical, thermal and moisture retention properties and antimicrobial properties and is helpful to determine their suitability in specific applications.

2.3.1 Solubility and crystallinity

Solubility is an important parameter for the use of chitin and its derivatives in a wide range of industrial applications. The sorption ability of chitin increases as the number of amino groups grows as the degree of deacetylation increases chitin is being converted into chitosan. The solubility characteristics of chitin/ chitosan are governed mostly by the extent of degree of deacetylation, the distribution of acetyl groups, degree of dissociation, processing methods, pH and the ionic strength [28, 59, 60]. The solubility of chitin and its derivatives can be demonstrated using chitosan in a dilute acidic medium. In this system, chitosan tends to be at equilibrium Eq. (1).

$$Chitosan - NH_2 + HOH \rightleftharpoons Chitosan - NH_3^+ + OH^-$$
(1)

Chitosan is soluble when pH is lower than 6 or 5.5. At lower pH, the amino groups in chitosan are fully protonated and the positively charged polymer chains will repel each other and fall apart in solution thus resulting in its dissolution. At pH above 6.5, chitosan will precipitate [54].

Polymer swelling reduces the crystallinity of chitin derivatives in a solution [61]. The crystallinity of chitin derivatives varies with the substitution of other functional groups onto the polymer chains [54]. The crystallinity index of chitin (85%) is higher than that of the water-soluble chitin derivative (48 to 57%) [42]. Because the acid or alkali treatments depolymerize the polymer chains during processing [62, 63]. Chitin shows a crystalline structure whereas water-soluble chitin derivatives show an amorphous structure due to structural modification [32, 54, 64].

2.3.2 Viscosity

Viscosity refers to the resistance to flow in liquids while elasticity refers to energy recovery in solids. Polymeric material may be time-dependent, acting more like a solid during short processing time (rapid movement) or acting more like a fluid during long processing time (slow movement). When a polymeric material has both fluid as well as solid behaviours, it is called viscoelastic. Like solubility, viscosity of chitin derivatives is also an important property in processing these polymers. The viscosity data also provides the information on the structure and properties of chitin-derived polymers [47]. This data will be helpful while designing new and innovative chitin-based films for various industrial applications. The dibutyryl chitin with an intrinsic viscosity of greater than 1 dl/g had a good spinnability and film-forming ability [65]. Apparent viscosity [η] or just viscosity, commonly used in place of dynamic viscosity, is defined as the ratio of the imposed shear stress [τ] to the shear rate [$\dot{\gamma}$]. See Eq. (2).

$$\eta = \tau / \dot{\gamma} \tag{2}$$

In a Newtonian fluid, the shear stress is proportional to shear rate, viscosity is therefore constant. If the viscosity of a fluid varies with respect to shear rate or shear stress, then it is termed a non-Newtonian fluid. A liquid such as water, alcohol etc, which is composed of a single substance is usually a Newtonian fluid. On the other hand, a polymer solution or a colloidal solution containing high molecular weight compounds and/or suspended solids is generally a non-Newtonian fluid.

Mathematical equations have been derived by many researchers to form flow models for describing the rheological behaviour of a material in terms of shear rate and shear stress. The simplest model that is used to describe the flow behaviour of a Newtonian fluid is Eq. (3). In case of non-Newtonian fluid, a power law model is usually used to describe the flow characteristics Eqs. (3) and (4) [66].

$$\tau = \eta \dot{\gamma} \tag{3}$$

$$\eta = K \left[\dot{\gamma} \right]^{n-1} \tag{4}$$

$$\tau = K \left[\dot{\gamma} \right]^n \tag{5}$$

where,

 τ —Shear stress (Pa)

 $\dot{\gamma}$ —Shear rate (s⁻¹)

 η —Apparent viscosity (Pa. s)

K—Consistency coefficient (Pa. sⁿ)

n—Flow behaviour index, dimensionless number (n = 1 for Newtonian fluids; n <1 for shear thinning or pseudoplastic fluids and n >1 for shear thickening fluids).

Rheological study of both chitin, as well as water-soluble O-carboxymethylated chitin derivatives in N,N-dimethyl acetamide/ lithium chloride solvent system, exhibits non-Newtonian shear-thinning behaviour. The power law indices of these solutions increase along with the temperature [67]. Therefore, a solution containing chitin and/or its derivatives is a non-Newtonian fluid.

2.3.3 Degree of deacetylation (%)

The degree of deacetylation has an influence on all physiochemical properties such as molecular weight, viscosity and solubility. The presence of 50% amine groups defines the boundary between chitin and chitosan; chitin has less than 50% deacetylation and chitosan has more [68]. The degree of deacetylation is affected by the concentration of alkali, processing temperature, reaction time, previous treatment of chitin, particle size and chitin concentration. The degree of deacetylation and the distribution of the acetyl groups influence the solubility [69]. This research used the colloidal titration method to determine the degree of deacetylation of chitin and its derivatives [27, 32, 70].

2.3.4 Thermal analysis

A 'glass' can be defined as a solid, brittle material that has an amorphous liquid-like structure with very little flexibility or any obvious fluidity. Glass can be achieved by melting the ordered form of the material and then by rapid cooling or supercooling. A perfectly crystalline polymer will melt at a well-defined temperature; this melting transition is defined as a first-order transition. Melting causes discontinuous changes in volume, enthalpy and primary thermodynamic variables [71]. In an amorphous

polymer, the molecular motion of the polymer chains is immobile at low temperatures. The state of the polymer is glassy. When the polymer is heated, the molecules obtain sufficient energy to slide over one another. The polymer becomes viscous, flexible or rubbery at the glass transition [72]. The glass transition temperature is highly specific to each anhydrous amorphous material and depends on experimental conditions, moisture content and molecular weight [73, 74].

The final glass transition temperature has significant impact on the final texture, diffusivity and the rate of deterioration. Stiffness or brittleness of the polymers is lost by the reduction in the glass transition temperature owing to the plasticizing effect of water. This will make the polymer unsuitable for making films [74–76]. Thermal behaviour of chitin derivatives is conducted by thermomechanical analysis (TMA), thermogravimetric analysis (TGA) and differential scanning calorimetry (DSC). DSC is an effective technique to evaluate the thermal behaviour as well as to determine the degree of deacetylation of chitin derivatives [64, 77].

Figure 1 shows the typical chromatogram of a chitin derivative during DSC analysis. In the DSC chromatogram, the first endothermic peak indicates the loss of absorbed moisture by the films. The second exothermic peak(s) indicates the degradation of chitin derivatives. Finally, the phase transition of the chitin polymer occurs. Various studies have been performed to analyze chitin and its derivatives using DSC [58, 77–80]. Both chitin and water-soluble carboxymethyl-chitin exhibit the endothermic peak that relates to the loss of water during DSC [64]. The decomposition of chitin during DSC is obtained in two stages: one peak at a temperature range of 200 to 260°C and the second peak at 300°C to 360°C [79]. In contrast, a single-stage decomposition of chitin is obtained at around 400°C [77]. Chitin shows better thermal stability than chitosan it contains fewer amine groups [77]. The increase in molecular weight causes a proportional increase in glass transition temperature. DSC analysis of



Figure 1. *DSC chromatogram of a chitin derivative.*

high molecular weight chitin and chitosan shows no glass transition even up to a temperature of 550°C [64, 77, 79].

2.3.5 Moisture absorption

Food packaging technology requires the use of low oxygen and carbon dioxide permeable materials. The presence of water in the polymer influences the way in which these gases are sorbed and diffused [81]. Biomedical or pharmaceutical activity depends on how the water molecules are associated with the polymer. Moreover, the swelling characteristics of polymer gels are dominated by the nature of the polymer and the state of water [82]. Moisture absorption in polymeric films is important for a variety of industries ranging from microelectronics to adhesives and coatings. In many applications, water absorption leads to reliability problems such as the degradation of dielectric properties, corrosion or delamination. A significant number of studies covering many polymer systems have focused on characterizing the absorption and diffusion properties of water in polymer films [83]. It is very interesting to obtain a better understanding of the water sorption phenomena and the mechanical strength of chitin-based films prior to their use in food, medical and pharmaceutical applications. The solubility and strong swelling of the finished films in water decrease the stiffness of the films [47].

Better understanding of moisture absorption mechanisms and controlling steps may help not only in optimizing the use of chitin films but also in designing new chitin-based polymers. Equilibrium moisture absorption properties are frequently controlled by diffusion properties [especially intra-particle diffusion], degree of deacetylation, chemical structure and physical modification of the polymer; when the size of sorbent particles increases, moisture absorption performance may drastically decrease, the time required to reach equilibrium exponentially increases and sometimes the sorption capacity at equilibrium diminishes [64, 81, 84]. The moisture absorption ability of water-soluble chitin derivatives depends on their chain conformation in solution and their molecular weight [39].

2.3.6 Antimicrobial properties of chitin

Although there are many studies about the antimicrobial activity of chitosan, few studies have been performed to analyze the antimicrobial activity of natural chitin or water-soluble chitin. This is because chitosan has higher degree of deacetylation. An increase in the degree of deacetylation of chitin and hence the number of amino groups increases the antimicrobial activity [26, 85–87]. Therefore, there is a relationship between the antimicrobial activity and the degree of deacetylation of chitin and its derivatives. This research mainly focuses on the antibacterial property of water-soluble chitin derivatives against *Bacillus cereus*.

3. Materials and methods

3.1 Materials used

Due to the difficulty in sourcing adequate commercial prawn waste, raw eastern school prawns (*Metapenaeus macleayi*, approximately 9 cm body length) were obtained from Department of Primary Industries, Fisheries Conservation Technology

Unit, NSW, Australia and hand-peeled to obtain prawn waste to conduct this research (moisture content 74%; Ash 23% and 11% chitin dry basis). This prawn waste was stored at –22°C until used for research. All solvents used were HPLC grade supplied by Lab Scan Analytical Sciences. All chemicals were AR grade.

3.2 Preparation of water-soluble colloidal chitin (WSCC)

Step 1: Natural chitin was recovered from prawn waste before it was converted into WSCC. First, prawn pigment, astaxanthin complex was extracted from prawn waste followed by deproteinization using 10% sodium hydroxide (1:2, w/v) at 100°C for 6 hours and then demineralization using 2M hydrochloric acid (1:3, w/v) for 48 hours at ambient temperature. The residue was natural chitin [88]. The moisture content chitin and ash content were measured using Equation 2.8 [89]. Care was taken that the ash content of chitin was less than 1%.

Step 2: This natural chitin was freeze-dried at ambient temperature for 24 hours at 0.4 mbar and stored in the desiccator. Finely ground freeze-dried natural chitin (100 to 125 mesh) (5%, w/v) was digested in sodium hydroxide (50%, w/v) for 6 hours at 87°C to prepare chitosan. The chitosan was thoroughly washed to neutral pH and freeze-dried [88].

Step 3: Chitosan hydrochloride was prepared by dissolving 2.5 g of freeze-dried chitosan in 100 mL of 10% acetic acid followed by precipitation using concentrated hydrochloric acid. The precipitate was thoroughly washed using methanol several times to get chitosan hydrochloride free of chloride. The presence of Cl⁻ was tested by adding 1% silver nitrate to the filtrate; a white precipitate indicated presence of Cl⁻. The chloride-free precipitate called chitosan hydrochloride was dried in the oven at 50°C and moisture content was determined.

Step 4: Two methods were studied to prepare WSCC.

Method 1: Oven-dried chitosan hydrochloride (0.5 g) was dissolved in 2 mL of distilled water and then re-acetylated by adding a mixture of acetic anhydride (5 mL) and pyridine (2.5 mL). The reaction mixture was stirred in a magnetic stirrer overnight at ambient temperature to evaporate the solvent. This WSCC was dried in the oven at 50°C and the moisture content was measured. Method 2: Oven-dried chitosan hydrochloride (0.5 g) was re-acetylated by mixing with an equal volume of acetone (5 mL) and acetic anhydride (5 mL). The mixture was stirred in a magnetic stirrer overnight at ambient temperature to evaporate the solvent. It was then dried in the oven at 50°C. The moisture content of the oven-dried WSCC was measured.

3.3 Preparation of WSCC film for antibacterial study

WSCC film of 0.22 mm thickness was prepared by pouring 15 mL of WSCC solution in water (5%, w/v) on a glass petri dish underlined with a layer of microwave-safe all-purpose food packaging film without imperfections. WSCC film of 0.52 mm thickness was prepared by pouring 100 mL of WSCC solution in water (5%, w/v) on a glass plate (16 cm \times 24 cm) underlined with a layer of GLAD wrap. These two samples were then oven-dried at 40°C. Careful attention was given that the prepared film was free from air bubbles and physical imperfections. The thickness of the dry WSCC films was measured using a vernier calliper at five random positions and averaged.

3.4 Characterization of WSCC

3.4.1 Viscosity of WSCC using different viscometers

In this study, the viscosity of WSCC dissolved in distilled water was evaluated using two viscometers. They are rotational Haake viscotester® VT550 and SV-10 AND Vibro viscometer (SV series 300). The results obtained from these two viscometers were then compared. Rotational Haake viscometer measures the viscosity by measuring the running torque of the cylindrical rotors immersed in a sample because viscosity is directly proportional to a running torque required to develop steady rotating motion. Its temperature is controlled by a re-circulating water bath and water jacket. SV-10 AND Vibro viscometer measures the viscosity by controlling the amplitude of the sensor plates immersed in a sample and measuring the electric current to drive the sensor plates. SV-10 AND viscometer vibrates with sine-wave of frequency about 30 Hz and amplitude of approximately 0.2 mm. The temperature at the geometric centre is measured, not controlled. These two viscometers have different shear rates. The shear rate of the Haake viscometer are shown in **Table 1**.

3.4.2 Viscosity testing using Haake viscometer

In the Haake viscometer, a sample solution (9 mL) of WSCC (5%, w/v) in distilled water was taken in a concentric cylinder (NV type, system no: 8, radius: 20.5 mm) together with an NV rotor (system no: 8, radius: 17.85 mm, height: 60 mm). Prior to the analysis, the sample solution and sensor system were placed in the water bath at the respective temperature for 15 minutes for equilibration. The flow curves were observed at shear rates 0.13 to 300 s⁻¹ and at temperatures 20°C, 30°C and 40°C. The parameters for power-law models were noted. The operating parameters used during viscosity analysis are given below. Ramp1 eliminated start-up effects and was not taken into account while analyzing the results.

Ramp1: CR 0.13 s⁻¹; t = 10 s; #5Ramp2: CR lin, 0.13–300 s⁻¹; t = 120 s; #100Ramp3: CR lin, 300–0.13 s⁻¹; t = 120 s; #100

Power-law models of flow curves were calculated for each temperature Eq. (4). The apparent viscosity at shear rates 5, 10, 100 s^{-1} was calculated from each model Eq. (5) [66]. The temperature dependence of viscosity follows the Arrhenius exponential relationship Eq. (6).

Viscosity coefficient (mPa.s)	Maximum shearing rate (s $^{-1}$)	Effective shearing rate (s ⁻¹)
1	590	420
10	130	92
100	42	30
1000	17	12
10000	10	7

Table 1.

Nominal shear rate values of a viscosity standard Newtonian fluid using SV-10 AND Vibro viscometer.

$$\eta = \mathrm{Ae}^{E/\mathrm{RT}} \tag{6}$$

where,

E—Activation energy of the sample

A—Empirical constant

T—Temperature of reaction mixture

R—Universal gas constant

η—Apparent viscosity

This exponential relationship of Arrhenius applies to a polymer system of low molecular weight and low viscosity [90]. From the results obtained, WSCC was a low molecular weight polymer with low viscosity. Therefore, Arrhenius models were constructed for WSCC solutions using Eq. (7). Arrhenius models were obtained by plotting the natural logarithmic of apparent viscosity (η) on Y-axis and the inverse of temperature (K^{-1}) on X-axis for each shear rate. Arrhenius model Eq. (7). Activation energy values (E) at each shear rate were calculated by multiplying the respective slope of the Arrhenius equation with the gas constant value (R). The intercept was the natural logarithmic of the empirical constant, A.

3.4.3 Viscosity testing using SV-10 AND viscometer

Sample solution (10 mL) of WSCC (5%, w/v) in distilled water was analyzed using SV-10 AND Vibro viscometer, which delivers a single shear rate, and the temperature and viscosity profile were recorded over time. An Arrhenius plot was constructed and Arrhenius constants were determined.

Apparent viscosity values (η_S) at 20°C, 30°C and 40°C were interpolated using the Arrhenius model, and the corresponding shear rates ($\dot{\gamma}$) were calculated from the effective shear rate *vs* viscosity co-efficient standard curve. The calculated shear rates were then substituted in the power-law equations of Haake viscotester Eqs. (4) and (5). The apparent viscosity values (η_H) corresponding to the SV-10 AND vibratory viscometer shear rates at 20°C, 30°C and 40°C were calculated. Finally, the apparent viscosities of WSCC in distilled water were obtained using both the viscometers and were compared.

3.4.4 Degree of deacetylation determination

Degree of deacetylation of chitin samples was measured by the colloidal titration method. Oven-dried chitosan hydrochloride (1 g) samples prepared from methods 1 and 2 were titrated against 0.1M sodium hydroxide using phenolphthalein indicator. The degree of deacetylation of the sample was calculated Eq. (7) [91]. Standard deviation and significant differences were calculated.

$$\%DD = \frac{N1V1}{1000} x \frac{V0}{V2} x (MWCTS - Cl) x \left\{ \frac{100}{W4x (1 - \%MC/100)} \right\}$$
(7)

where,

%DD—Degree of deacetylation of chitin sample (%)

N₁—Concentration of sodium hydroxide (M)

V₁—Volume of sodium hydroxide used (mL)

V₀—Total volume of chitosan chloride solution (mL)
V₂—Volume of chitosan chloride solution used for titration (mL)
MW_{CTS-Cl}—Monomer weight of chitosan chloride
W₄—Weight of chitosan chloride taken for titration (g)
%MC—Moisture content of chitosan chloride (%)

3.4.5 Differential Scanning Calorimetric [DSC] analysis of WSCC film

Differential scanning calorimetric measurements were performed using Universal V4.3A TA Instruments. WSCC films [sample weight 3 mg, 0.22 mm thickness] were equilibrated in a desiccator or in the relative humidity chamber (92.5%) for a week and the DSC analysis was performed under a dynamic nitrogen atmosphere (50 mL/minute) at a heating rate of 5°C/minute. Samples equilibrated to a range of relative vapour pressures (36.1%) between these two were also tested. Intermediate relative humidity (36.1%) of WSCC film was obtained by equilibrating the film at ambient conditions. Accurately weighed sample (± 0.1 mg) was placed into a covered aluminium sample holder. An empty sample holder was used as reference and two runs were performed for each sample by heating the sample from 25°C up to 450°C.

In another study, two samples treated at the ambient relative humidity were tested individually using DSC. The DSC curves were performed under a dynamic nitrogen atmosphere (50 mL/minute) at a heating rate of 5°C/minute. Accurately weighed samples (± 0.1 mg) were placed into an aluminium sample holder and sealed. An empty sample holder was used as reference and the runs were performed by heating the samples from 25°C up to 110°C with an isothermal for 15 minutes to remove the moisture present in the samples. The samples were then reweighed and reheated from 25°C up to 480°C [77]. The results are then compared with the phase behaviour of the moist WSCC films.

3.4.6 Moisture absorption isotherm of WSCC film

In this study, moisture absorption behaviour of WSCC was investigated at different relative humidities. Moisture absorption isotherm of WSCC film of 0.22 mm thickness was prepared using saturated salt solutions of different relative humidities (**Table 2**). WSCC films (2.5×2.0 cm) were placed in each of the relative humidity chambers and the samples were kept at a controlled temperature of 25°C for equilibration. The moisture gained by each of the samples was measured after a week and

No.	Saturated Solutions	Equilibrium relative humidity (%) at 25°C
1	Potassium acetate [CH ₃ COOK]	22.5
2	Magnesium chloride [MgCl ₂ . 6H ₂ O]	32.7
3	Potassium carbonate [K ₂ CO ₃ . 2H ₂ O]	43
4	Sodium nitrite [NaNO ₂]	64
5	Sodium chloride [NaCl]	75.1
6	Potassium chloride [KCl]	84.2
7	Potassium nitrate [KNO ₃]	92.5

 Table 2.

 Relative humidity standards used for the experiment [83, 92].

the moisture absorption isotherm was prepared by plotting the relative humidity on X-axis and the moisture content of the film on Y-axis.

3.4.7 Antibacterial activity of WSCC film

Bacillus cereus 043800 was supplied by the culture collection of Department of Biological Sciences, UNSW and stored at -80° C freezer. The inoculum suspension (10 µL) was spread on WSCC film (2 cm × 1.5 cm). A control WSCC film was made without inoculum. Each treatment was carried out in duplicate. The films were then placed in the aseptic plastic Petri plates and autoclaved glass Petri plates separately. Petri dishes containing the films were then sealed using adhesive tape and incubated at 30°C for a day. Then inoculated films were taken out and the films were placed into a 'stomacher' bag with 10 mL of peptone/water (0.1%, w/v). The stomacher bags were initially massaged by hand to loosen the adhesion of cells to the film followed by 'stomaching' using stomacher for 5 minutes and then allowed to sit for 5 to 10 minutes. The bag was again massaged by hand before plating to ensure homogeneous distribution of the suspension. 50 µL of each of these samples was spread-plated onto a brain heart infusion agar plates. The plates were incubated at 30°C, and colonies were counted 1 day later [93].

4. Results and discussion

4.1 Solubility of WSCC

The WSCC prepared by Method 1 was soluble in water and the oven-dried material was not readily soluble in water. Moreover, this method was not reproducible, and the strong odour of pyridine was other major concern. When the processing conditions of chitin to chitosan were changed in Method 2, the degree of deacetylation of the prepared chitosan was less and the oven-dried WSCC readily formed a colloidal suspension in water. Method 2 was reproducible in preparing WSCC and this method did not use pyridine. Therefore, Method 2 was chosen to produce WSCC from prawn waste.

4.2 Characterization of WSCC

4.2.1 Viscosity of WSCC using different viscometers

4.2.1.1 Viscosity testing using Haake viscometer

Flow curves obtained for WSCC dissolved in distilled water (5%, w/v) using Haake viscometer at different temperatures and different shear rates were modelled by power-law equations. The value of the power law exponent, n was less than 1 (ranging between 0.8 and 0.9) at all temperatures Thus, WSCC dissolved in distilled water was a shear-thinning, non-Newtonian fluid. As would be expected, the apparent viscosities generally decreased with an increase in temperature and shear rate during the experiment. A flow curve obtained for WSCC dissolved in distilled water (5%, w/v) at 30°C is shown in **Figure 2**.

The Arrhenius model for WSCC in distilled water using Haake viscometer at different shear rates and at different temperatures is shown in **Figure 3**. Arrhenius



Figure 2. Flow curve obtained for WSCC dissolved in distilled water (5%, w/v) at 30°C using Haake viscometer.



Figure 3.

Arrhenius models of WSCC in distilled water (5%, w/v) at different shear rates and at different temperatures: Haake viscometer.

models at different shear rates were straight lines with negative slope, hence fit into Arrhenius exponential relationship. The R² values were between 0.94 and 1.00. An increase in temperature decreased the apparent viscosities at different shear rates $(5 \text{ s}^{-1}, 10 \text{ s}^{-1} \text{ and } 100 \text{ s}^{-1})$. Therefore, temperature dependence of apparent viscosity of WSCC in distilled water obeyed the Arrhenius exponential relationship. The

Arrhenius model constants at different shear rates for this particular sample system are shown in **Table 3**. In this sample system, the activation energy values increased with an increase in temperature (positive activation energy values) and shear rate.

4.2.1.2 Viscosity testing using SV-10 AND vibratory viscometer

The relationship between time, temperature and apparent viscosity profile of WSCC in distilled water (5%, w/v) using SV-10 AND vibratory viscometer is shown in **Figure 4**. As expected, the apparent viscosity decreased with increase in temperature.

The Arrhenius model of natural logarithmic values of apparent viscosity against the reciprocal of absolute temperature for WSCC in distilled water (5%, w/v) using SV-10 AND viscometer is shown in **Figure 5**. The relationship was linear with negative slope ($R^2 = 0.99$), thus followed the Arrhenius exponential relationship. The Arrhenius model for this system obtained using SV-10 AND viscometer was y = -8252.4x + 32.297. The intercept, which is the natural logarithmic value of the empirical constant, was -32.297. The activation energy for this sample system calculated

Shear rate (s $^{-1}$)	Activation energy, E (KJ/mol.K)	Empirical constant, ln(A)		
5	0.0005	-4.6568		
10	0.0006	-4.7567		
100	0.001	-5.0889		

Table 3.

Arrhenius model constants of WSCC in distilled water (5%, w/v) at different shear rates: Haake viscometer.



Figure 4.

Temperature and apparent viscosity profile of WSCC in distilled water (5%, w/v) using SV-10 AND vibratory viscometer.



Figure 5. Arrhenius model of WSCC in distilled water (5%, w/v) at a single shear rate: SV-10 AND vibratory viscometer.

from the slope of the Arrhenius model was 68.61 KJ/mol.K, which is a positive value. Positive activation energy for the same sample system was obtained in the Haake viscometer as well (**Table 3**).

The Arrhenius models constructed for WSCC in distilled water (5%, w/v) using both viscometers showed negative slope. The Arrhenius models obtained from both viscometers confirmed that the apparent viscosity decreased with an increase in temperature. As mentioned in 3.4.1, the apparent viscosities (η_S) of WSCC dissolved in distilled water (5%, w/v) at 20°C, 30°C and 40°C were calculated from the Arrhenius model of the SV-10 AND vibratory viscometer, which are shown in **Table 4**. The corresponding shear rate values for the calculated apparent viscosities at 20°C, 30°C and 40°C were obtained from the effective shear rate *vs* viscosity standard curve of the SV-10 AND vibratory viscometer. These values are listed in **Table 4**. The apparent viscosities were not constant with varying shear rates. Thus, WSCC in distilled water (5%, w/v) behaved like a non-Newtonian fluid. A similar result was obtained when analysed using the Haake viscometer.

The obtained shear rate values from the SV-10 AND vibratory viscometer values at 20°C, 30°C and 40°C were then substituted into the power-law model of Haake viscometer. The apparent viscosities (η_H) for those shear rates were then worked out using Haake viscometer power-law equations. The obtained apparent viscosities (η_H) are given in **Table 4**.

Temperature (°C)	Apparent viscosity (mPa.s)	Shear rate (s^{-1})	Apparent viscosity $\eta_{\rm H}$ (mPa.s)
20	15.82	48.55	9.43
30	6.25	32.14	8.75
40	2.62	21.85	8.32

Table 4.

Apparent viscosities and shear rates of WSCC in distilled water (5%, w/v) at different temperatures: SV-10 AND vibratory viscometer.

The apparent viscosities for the respective shear rates obtained using both viscometers were different (**Table 4**). The differences in the calculated apparent viscosities by these two viscometers are due to the different working principles. In Haake viscometer, shear rate can be controlled whereas the shear rate cannot be controlled in SV-10 AND Vibro viscometer. Both these viscometers showed that the system containing WSCC in distilled water (5%, w/v) was a non-Newtonian fluid with shear thinning behaviour and an increase in temperature reduced the apparent viscosities. Similar results are obtained for chitin and water-soluble *o*-carboxymethylated chitin dissolved in *N*,*N*-dimethyl acetamide/ lithium chloride solvent system [67]. Thus, the apparent viscosity of WSCC dissolved in distilled water is dependent on the temperature and the shear rate.

4.2.2 Degree of deacetylation of WSCC

There was no significant difference between the degree of deacetylation of chitin and WSCC (**Table 5**). This is because acetylation with a mixture of acetic anhydride-acetone gives rise to the complete acetylation of amino groups [94, 95]. Thereby the original degree of deacetylation of chitin was restored by WSCC. The yield of WSCC was 82 g/kg of prawn waste (dry basis).

4.2.3 Differential Scanning Calorimetric (DSC) analysis of WSCC film

The DSC curves obtained for WSCC films equilibrated in high relative humidity (92.5%) (moisture: 16.4%) and in a desiccator (moisture: 6.52%) are shown in **Figures 6** and 7 with exothermic peaks facing up. A generic DSC curve obtained for the WSCC film equilibrated in high relative humidity is also shown in **Figure 1**. Samples equilibrated to a range of relative vapour pressures between these two were also tested. The DSC thermogram for both these samples showed an endothermic peak (Peak 1) after 70°C followed by an exothermic peak around 180°C (Peak 2).

Various studies indicate that the endothermic peak at 100°C is attributed to the evaporation of absorbed water and the first exothermic peak is probably due to the degradation of chitosan studied during DSC [64, 82, 96–98]. In another study, the endothermic peak corresponds to the loss of moisture seen at a lower temperature (70°C), which is similar to this experiment [77]. The onset of the endothermic peak is related to pressure build-up because of water evaporation inside the sealed sample cups during DSC. The pressure at which the seam of the cups started to leak corresponds to approximate vapour pressure of water. The leaking of sealed sample pans during DSC also relates to the sample weight loss after the run [64].

Peak 1 in this study is due to loss of absorbed moisture by the WSCC film. As expected, a smaller peak of water loss (Peak 1) for a low moist sample was obtained

Sample	Degree of deacetylation (%)
Chitin	$\textbf{27.64} \pm \textbf{0.61}$
Chitosan (Method 1)	88.22 ± 0.46
Chitosan (Method 2)	54.08 ± 0.69
WSCC	28.46 ± 1.08

Table 5.

Degree of deacetylation of chitin and its derivatives.



Figure 6. DSC curve for high-moisture WSCC film.



Figure 7. DSC curve for dry WSCC film.

compared to a high moist sample. It was also expected that the sample equilibrated in the desiccator would be truly anhydrous. However, sample treated in desiccator showed endothermic peak for water loss. Some water in the film was not completely removed during sample drying in the desiccator. Therefore, it is suggested that desiccators be evacuated to accomplish faster and complete removal of water. Moreover, totally anhydrous chitin samples are difficult to obtain because of the high water affinity presented by these polymers [77].

In this study, Peak 2 is due to the degradation of chitin derivatives. The first exothermic peak for degradation of chitin derivatives is obtained at around 180°C in

this study. The exothermic peak for the decomposition of chitin derivatives depends on the molecular weight and the presence of hydrophilic groups [64]. Viscometric analysis of WSCC shows that the molecular weight of WSCC is lower than chitin. Because of this reason, low molecular weight WSCC derivatives degrade faster than the high molecular weight chitin. The degradation of WSCC film occurred at relatively lower temperature than the published data of chitin and water-soluble carboxymethyl-chitin run [64]. The thermal stability of WSCC is poorer compared to chitin and water-soluble carboxymethyl-chitin.

The degradation temperature (T_d) of high-moist WSCC film was low compared to low-moist WSCC film. However, noticeable differences in the degradation temperature of both low-moisture and high-moisture samples were hard to see (**Table 6**). WSCC film did not show a considerable increase in the absorption of moisture over a wide range of relative humidities (**Figure 8**). Therefore, moisture content of WSCC film slightly influences the degradation of WSCC derivatives in this study.

The glass transition temperature (T_g) for WSCC film was observed at around 360° C. The glass transition temperature (T_g) of high-moisture WSCC film was lower than that of low-moisture WSCC film (**Table 6**). Higher molecular mobility accelerates reactions limited by diffusion and decreases stiffness [84]. The thermal degradation of the material is sensitive to moisture content confirms that degradation reactions were

Samples	T _m (°C)		T _d	(°C)	T _g (°C)	
	Run 1 Run 2		Run 1	Run 2	Run 1	Run 2
Low-moisture WSCC film	72.82	70.32	185.97	186.91	364.05	364.45
Standard Deviation	1.41		0.67		0.28	
High-moisture WSCC film	76.96	77.69	181.23	185.10	360.07	363.48
Standard Deviation	0.52		2.74		2.41	

Table 6.

Glass transition temperature (T_g) of WSCC films determined using DSC.



Figure 8. Moisture absorption isotherm of WSCC film.

diffusion-limited. Therefore, moisture content did influence the phase behaviour of WSCC film. There was, however, sample weight loss in both high-moisture (57%) and low-moisture (50%) WSCC films after DSC run. The difference in the sample weight loss is due to the difference in the moisture content. The sample weight-loss (15%) of water-soluble carboxymethyl-chitin film after DSC run is also reported by Ref. [64]. It was beyond the scope of this project to exhaustively test the rheological and thermophysical properties of chitin and chitosan films against a range of other polymers. Such tests would be more profitably carried out at a larger scale of production, which would enable rolled or extruded films to be made. Nevertheless, the tests done show that the recovery of prawn chitin and its processing can lead to films of high and reproducible quality.

4.2.4 Moisture absorption isotherm of WSCC film

The specific moisture absorption isotherm of WSCC film (0.22 mm thickness) appeared to follow the expected sigmoid curve usually obtained over the whole range (0 to 1) of relative vapour pressure (**Figure 8**). This kind of isotherm is also called a type II isotherm in reference to the Brunauer–Emmett–Teller (BET) model [99]. A similar result was obtained for chitosan films by Ref. [99].

The specific moisture absorption of WSCC film increased with increasing relative humidity. The absorbed moisture by WSCC films varied from 40% to 55% between low (22%) and high relative humidity (92.5%). In this case, there was not much difference obtained in terms of the absorption of specific moisture over a wide range of relative humidities. The other study shows that the absorbed moisture content by chitosan films varies from 5 to 45% over relative humidities from 20% to 80% [99]. This indicated the moisture absorption of WSCC films was less compared to chitosan films over a wide range of relative humidities. The solubility and strong swelling of the finished film in water decreases the stiffness of the films which in turn decreases the suitability of the film in various applications [47]. The specific moisture absorption behaviour of WSCC film may not vary considerably over a range of relative humidities. This property will increase the selectivity of the finished films when they are used as membranes, packaging films or coating materials. In-depth study on the moisture absorption mechanism of WSCC films will be carried out during the scale-up of this process. Moisture absorption isotherm of WSCC films will be useful to determine the suitability of the film for maintaining proper moisture content of a particular product.

4.2.5 Antibacterial activity of WSCC film

The antimicrobial activity of WSCC film against the gram-positive bacterium *Bacillus cereus 043800* was investigated. The average values of the number of colonies of *B. cereus* during the antibacterial study are presented in **Table 7**. The empty glass and plastic Petri plates incubated alone for sterility check did not show the presence of *B. cereus* on brain heart infusion agar. WSCC film inhibited around 90% of the growth of *B. cereus 043800*. This shows that films made by WSCC have excellent antibacterial activity against *B. cereus*. The interesting finding in this study is that although the degree of deacetylation of WSCC was less (28.5%), WSCC film inhibited almost 90% of the growth of *Bacillus cereus 043800*. The extension of this work would test the overall antimicrobial activity of WSCC film against different microorganisms. Recently, water-soluble chitin derivation was investigated for the film coating of

Antibacterial study details	Plastic Petri plate	Glass Petri plate
(A) Cell concentration in <i>stock</i> suspension (cfu/mL)	8.7×10^7	8.7×10^7
(B) Cell concentration in <i>inoculum</i> suspension (cfu/mL)	2.98×10^6	2.63×10^6
(C) Average number of bacteria in the control (WSCC film without bacterial <i>inoculum</i>) (cfu)	1.98×10^4	$2.13 imes 10^4$
(D) Average no. of bacteria inoculated onto film (cfu)	2.98×10^4	2.63×10^4
(E) Average no. of bacteria present in the film after incubation (cfu)	3×10^3	3.1×10^3
Bacterial colonies inhibited (100–(E/D \times 100)) (%)	89.2	88.2

Table 7.

Concentration of Bacillus cereus 043800: An antibacterial study of WSCC film.

Ricotta cheese, and it proved to be efficient in prolonging the shelf life of Ricotta cheese [17].

5. Conclusions

In conclusion, this study has clearly demonstrated the fundamental characteristics of WSCC. The study of WSCC will be helpful in evaluating its suitability in various industrial sectors including food and pharmaceuticals. To effectively use WSCC as a functional ingredient, relationships between the functional properties and characteristics of WSCC must be constantly monitored for proper quality control. In the current study, limited relevant information on aspects of such relationships was obtained. More extensive investigations are needed for a better understanding of the relationships reported in the present research, especially in view of current worldwide interest in commercial use of water-soluble chitin derivatives. This will be best done during the scale-up of this current project.

The biodegradable water-soluble colloidal chitin recovered from prawn waste offers exciting possibilities because of its water solubility. These chitin derivatives can be obtained from the prawn waste changing the waste stream into a valuable resource that is commercially viable. The antibacterial property of water-soluble colloidal chitin may be applied in pharmaceutical, medical and food packaging and coating applications. Biodegradable films made from these natural polymers obtained from renewable sources will protect the environment in the future and will be an effective alternative to plastics that threatens the environment. However, vegetarians may object to the use of animal polymers.

6. Recommendations

The study of the preparation of water-soluble colloidal chitin from prawn waste recommends scale-up to pilot scale and reexamination of the stability of water-soluble colloidal chitin. Further microbial investigation is required to explore the water-soluble colloidal chitin film's antimicrobial activity on a wide range of microorganisms. An appropriate method needs to be developed to determine the molecular weight and physio–chemistry of the water-soluble colloidal chitin following scaled-up production. Detailed study about the toxicity of water-soluble colloidal chitin is necessary prior to its use in industrial applications, especially in food applications because these compounds may contain chemical/solvent residues. Water-soluble colloidal chitin can be tested as an antimicrobial coating in fruits and vegetables. The application of water-soluble colloidal chitin in food packaging applications can be studied in the future.

Further, the use of sodium hydroxide and hydrochloric acid during preparation of water-soluble colloidal chitin generates chemical waste. World over efforts is on to find out an alternative to chemical method of preparation of chitin/ chitosan. Therefore, an alternative way of preparing water-soluble chitin derivatives using enzymes or simply by adding or substituting functional group to chitin molecule is recommended. Such material can be used for preparing films and the related studies.

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Chapter 3

Preparation and Bioactivity Applications of Novel Chitosan Derivatives

Mohsin Mohammed and Nadia Haj

Abstract

Chitosan (CS) is a substance abundant in nature. It is a biopolymer consisting of repetitive components of glucose and *N*-acetyl-glucose amine connected by (1,4)-gly-cosidic bonds. It has so many applications that are biodegradable, non-toxic, and biocompatible. The CS was loaded with 5-fluorouracil (5FU) *via* amide-mediated bind-ing, and the resulting CSFUAC product was evaluated as a potential 5FU delivery agent. A new CS-Schiff base derivative was created using CS extracted from local fish scales by combining CS with another aromatic aldehyde. The antimicrobial effectiveness of the new product was evaluated. It includes two fungi and four strains of pathogenic bacteria. The MTT assay is employed to determine the cytotoxicity of the newly synthesized compounds. Finally, CS was used to synthesize a prodrug for colon cancer. As a colon cancer prodrug, methotrexate (MTX) was converted to the combined (methotrexate-imidazole) and linked with the CS to produce the CSMTX conjugate. Additionally, the compound's hemolytic action and chemical stabilities were evaluated. In the MTT, three types of cancer cell lines (MDAMB231, MCF7, and MDAMB453) were utilized to test how toxic the compounds made in the lab were to cancer cells.

Keywords: chitosan, chitosan-Schiff base, prodrug, cytotoxicity, antibacterial activity

1. Introduction

The chemical structure of chitin (SH)-derived chitosan (CS) has the chemical structure of " $(1 \rightarrow 4)$ 2-amino-2-deoxy—D-glucopyranose." This is a typical co-biopolymer origin in the shells of cockroaches, the shells of crustaceans, and fungal cellular walls (**Figure 1**). The primary sources of CS and SH are crustaceans-like crabs, shrimp, and fish scales. CS is the most abundant and superior natural substance in nature, and second only to cellulose. Due to its excellent quality and adaptability, CS is in a league. In addition, they possess unique properties, for example, non-toxic, mucosal adhesion, biodegradability, biocompatibility, antimicrobial activities, and hydrophilicity. However, it is cholesterol lowering. These characteristics make CS useful in medicine, horticulture, stabilizers for staple foods, biocatalysts, and biology [2–7].

The beneficial biological effects of CS include antitumor, antibacterial, and hemostatic effects and wound healing. Applications include biomedicine design,



Figure 1. Chitin, chitosan, and cellulose's molecular structures [1].



Figure 2.

Publications indexed by Scopus concerning chitosan and its derivatives.

pharmaceuticals, drug delivery, restoration materials, chelation of metal particles, water absorption processing, and plant safety [8–10]. From 1985 to June 2015, there was a significant increase in the research on chitosan and its derivatives; **Figure 2** depicts the number of Scopus-indexed publications on chitosan and its derivatives [11].

Figure 3 shows some chemical modifications of chitosan. **Table 1** illustrates some applications of chitosan derivatives.

Below is a summary of some applications of CS:

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Figure 3. Some chemical modifications of chitosan [12].

Fields	Applications	Descriptions/references		
Food industry	Food packaging preservation	Film-forming property and antibacterial property [13].		
	Food preservative	Antibacterial properties [14].		
	Beverage clarifier	Flocculation [15].		
	Wastewater treatment	Flocculation and chelating and adsorbing property [16].		
Chemical	Chemical industry	Moisture absorption and water retention properties [17].		
industry	Mouthwash	Antibacterial activity [18].		
	CS fiber	Antibacterial and anti-wrinkle [19].		
Textile industry	Dyeing and fixing	Physical adsorption and film-forming property [20].		
	Coatings	Film-forming property and antibacterial property [21].		
	Drug carrier	Film-forming, antioxidant, antitumor, and biocompatibility [22].		
Medicine	Wound dressings	Anti-inflammatory and antibacterial properties [23].		
	Tissue engineering	Proliferative, hemostatic, and antibacterial properties [24].		
	Artificially simulate enzyme	Catalysis [25].		
Functional materials	Liquid crystal materials	Optical and film-forming property [26].		
Agriculture	Protect seed	Film-forming property and antibacterial property [27].		
	Improve soil	Adsorption and bacteriostatic activity [28].		
	Improve crops	Immune and bacteriostatic properties [29].		

Table 1.

Examples of chitosan derivative applications.

2. Chitosan as a prodrugs for cancer

Worldwide, the leading cause of death is cancer. In 2008, cancer was the cause of death for about 13% of all people who died, or 7.6 million people. Numerous kinds of cancer are known, such as "prostate, colon, lung, and breast cancers," but colon cancer is the most lethal [30]. Numerous medications, such as Bevacizumab, (Avastin)Oxaliplatin, Folinic acid, 5-fluorouracil (5-FU), Methotrexate, and Celecoxib, are used to treat cancer and are used to treat colorectal cancer (Celecoxib). However, most drugs are consumed with food, restricting the higher gastrointestinal tract (GIT). They are not stable inside the body, which prevents the drug from concentrating effectively at the desired tumor site. Due to their lack of specificity, some medicines can also have the opposite effect [31]. 5-FU is a specific anticancer drug that is frequently employed in treatment. 5-FU has disadvantages, including rapid ingestion in the body, a small half-life, a propensity for activating cancer cell resistance and cytotoxicity. These facets necessitate higher medication dosages, which raises the danger of adverse properties. So, a perfect system for delivering 5-FU would send the drug in tiny amounts and let it out quickly at the target site [32–34]. A prodrug for colon cancer must meet several requirements, such as not being toxic, compatible with the body, and stable in the GIT [35–38]. This section designates the synthesis of a prodrug for the colon by binding CS with 5-FU. The drug was transformed to "5-fluorouracil-1-acetic acid (FUAC)," which was then conjugated to CS to produce "chitosan-1-acetic acid-5-fluorouracil (CS-FUAC)," which has been validated for colon cancer treatment. The CS was extracted from fish scales using a chemical process. Infrared Fourier transform and ultraviolet spectrophotometry was used to characterize the product. It was looked at as a possible delivery agent after a covalent bond was made between CS-FUAC and 5-FU.

2.1 SH and CS extraction

In Kirkuk, Iraq, fish scales were collected at a local fish market. Based on a documented procedure, for demineralization and deproteinization, a 1% sodium hydroxide and hydrogen chloride solution were prepared with the receptivity of 40 g/mol and 36.5 g/mol [39].

2.2 Preparation of CS-FUAC

As published in the literature, FUAC was constructed with minor variations. 5-FU and aq. of KOH are reacted and heated at 100°C for 90 minutes. Then, for over 6 hours, chloroacetic acid was added regularly using a water bath at 60°C while stirring. The result was acetified to produce needle crystals of FUAC with a yield of 60% [40]. To achieve "1-acetic acid-5-fluorouracilimidazoline," a solution of FUAC in DMSO and "1,1-carbonyldiimidazole" was combined with CS in a glacial acetic acid (GAA) aqueous solution (**Figure 4**). "FT-IR spectroscopy" was utilized to examine the CS conformation, and synthesized FUAC and the CS-FUAC. Using a UV spectrophotometer, the produced chemicals were analyzed. Experimentation was conducted at a wavelength of 273 nm. Using a "Bruker Avance (500) spectrometer", ¹H- and ¹³C-NMR spectra were collected from DMSO-d6 and 1% CF₃COOD/D₂O solutions (**Figure 5**).

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Figure 4. *Method for preparing "CS-FUAC."*



Figure 5. ¹*H*-*NMR spectra of CS-FUAC.*

2.3 The drug content percentage of CS-FUAC conjugates

UV-visible spectroscopy established the fraction of FUAC conjugated to CS. The amide link in the CS-FUAC conjugates was initially hydrolyzed in the primary medium. As predicted, the concentration of FUAC was raised by raising the molar fraction of CS to FUAC. **Table 2** displays the FUAC concentration in conjugates where the molar ratios of "CS: FUAC were 1:1 and 1:2," respectively.

2.4 The stability test of conjugates of CS-FUAC

During transit through GIT, the drug used in the treatment of colon cancer must be stable in various pH ranges. Therefore, two buffers with pH of 1.2 and 7.4 were used to examine the prodrug stability in acidic and basic environments to determine its performance in acidic and basic environments. Based on the way conjugates are released in an acidic buffer, only 3–5% could be released. In contrast, it was 3–4% in



Figure 6.

Stability evaluation of CS-FUAC and 5-FU at pH 1.2 (A) and a pH of 7.4 (B).

Sample ⁱ	Drug loading (wt%) ⁱⁱ	Yield (%)
CS-FUAC (1:1)	0.16	60
CS-FUAC (1:2)	0.75	68
ⁱ All reaction was carried out at 75°C for 24 h. ⁱⁱ Drug loading determined by UV.		

Table 2.

Reaction data for CS-FUAC.

the primary buffer (**Figure 6**). In acidic environments, drug release is more significant than in neutral conditions. This could be because of the hydrolysis of an amide linkage in the acidic condition.

2.5 Conclusion

CS-FUAC, a potential colon-prodrug, was synthesized. The created conjugates appeared more stable under the initial conditions. In addition, CS-FUAC was more durable than the original medication in different pH conditions. An *in vitro* cytotox-icity investigation revealed that those synthesized derivatives are more active than free drugs. The cytotoxicity test, *in vitro*, demonstrates that these prodrugs are significantly more effective against "human colorectal cancer cell lines (HT-29)" than free medicine. In addition, it was nearly twice as cytotoxic to colon cancer cells compared with normal cells. Considering these outcomes, CS-FUAC as a prodrug appears to be an excellent method for colon delivery.

3. Synthesis and evaluation of Schiff base chitosan in biological systems

The Schiff base reactions produce the essential CS derivatives because of their application in the organic field. Furthermore, the response of CS with aromatic rings or heterocyclic aldehydes has led to the formation of stable Schiff bases (SBs), which are excellent molecules with uses in pharmacology, medicine, and other fields. Antimicrobial and cancer-prevention medications, for example [41, 42]. Interaction between CS amine sites and aldehydes or ketones, then removing water molecules, produces CSSBs [43]. Quinoline and quinazoline derivatives can also be found in

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various natural products, producing heterocyclic molecules with critical pharmacological applications. Quinoline and its results have antimalarial, antiviral, antibacterial, analgesic, anti-hepatoma, and anti-inflammatory properties. Oxazole derivatives are considered essential heterocyclic molecules in medicinal chemistry [44–56]. CH is extracted from a fish scale described previously and deacetylated to produce CS. This study's objective is to estimate the DD percentage of CS, which is carried out by acid-base titration for samples of CS collected at various stages of the manufacturing process. In this investigation, FTIR and TGA will be used to characterize and validate the physicochemical parameters of the products. This investigation seeks to develop a synthetic method for three new CS Schiff bases (CSSB)s compounds by combining CS with "2-chloroquinoline-3-carbaldehyde, quinazoline-6carbaldehyde, and oxazole-4-carbaldehyde." Furthermore, the antibacterial possible of CS and its novel derivatives were investigated using two types of "fungi, C. Albicans and A. fumigate." Utilizing FT-IR, (¹H, and ¹³C) NMR spectroscopy, the structures of the manufactured products were established. The cytotoxicity of newly synthesized derivatives was determined using the MTT experiment.

3.1 Method of acid-base titration

The acid-base titration technique was used to calculate a CS sample's degree of deacylation (DD). An indicator was used to control the endpoint, which turned to a blue-green color. The endpoint of the acid-base titration was used to calculate the DD percentage. The below equations were used to calculate the DD% of the deacetylated SH [57].

$$(-\text{NH2\%}) = \left(\frac{0.016[(\text{C1V1}) - (\text{C2V2})]}{\text{Wx100}}\right)$$
(1)

$$(DD\%) = 203(NH2\%/(16 + 42(-NH2\%))x100$$
 (2)

where the concentration and volume of HCl used are C1 and V1, the concentration and volume of NaOH used for titration are C2 and V2, and the weight of samples used for acid-base titration is W.

3.2 Synthesis of CSSBs

CSSBs were created by dissolving extracted CS in 2.0 percent aq. acetic acid, and carbonyl compounds dissolved in ethanol and added to the same amount of "CS. CS-P1, CSP2, and CS-P3" were the products of reactions between CS and "2-chloroquinoline-3-carbaldehyde, quinazoline-6-carbaldehyde, and oxazole-4-carbaldehyde", respectively. Three CSSBs were synthesized, as exposed in **Figure 7**.

The configuration of the synthesized derivatives was established with "FT-IR spectroscopy, ¹H NMR, and ¹³C NMR." All FT-IR spectra of the newly prepared derivatives presented a band between 1633 and 1655 cm⁻¹, related to the (-C=N) bond. Bands between 1400 and 1500 cm⁻¹ and 1057 cm⁻¹ correspond to (C-C) and inplane (C-H) bonds. The absence of a band in the region between 1660 and 1730 cm⁻¹ showed that the carbonyl group was absent, denoting that no free carbonyl remained. The (C-H) stretching of (CH₃- and -CH₂-) is represented by the vibrational bands at 2921 and 2883 cm⁻¹. Bands at 1155 and 900 cm⁻¹ were found in the glycosidic bonds. The glycosidic bonds and the (C-O, C-C, and C-O-C) stretching of the glycan ring



Figure 7. Development of CSSB derivatives.



Figure 8. ¹*H NMR form CS-P1 compound.*

were linked to the 1205–975 cm⁻¹ bands [58, 59]. "Using ¹H and ¹³C NMR, the structure of the prepared CSSBs was confirmed." The ¹H NMR spectra of the synthesized derivatives CS-P1 and CS-P2 are depicted in **Figures 8** and **9**, respectively. The ¹³C NMR spectra for CSP1 is illustrated in **Figure 10**.

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Figure 9. ¹*H NMR spectra of CS-P2.*



Figure 10. ¹³C NMR spectra CS-P1.

3.3 Examine the solubility

The solubility of the synthesized products was investigated using a variety of organic solvents. DMSO and mixtures of DMSO-CF₃COOH in equal proportions were used. The results are shown in **Table 3**. Certain solvents, such as diluted HCl and

Solvents										
	CH ₃ C	соон	CF ₃ C	юон	DMSO	HCl	NaOH	H ₂ O	кон	DMSO+ CF ₃ COOH
Comp. codes	25°C	70°C	25°C	70°C	25°C	25°C	25°C	25°C	25°C	25°C
CS-P1	S+	S*	S**	S**	S	S*	S+	S+	S**	S
CS-P2	S+	S*	S**	S**	S	S*	S+	S+	S**	S
CS-P3	S+	S*	S**	S**	S	S*	S+	S+	S**	S
S = soluble, S+ = insoluble, S* = partially soluble and swelling, S^{**} = partially soluble.										

Table 3.

CSSB solubility characteristics in a range of solvents.

CH₃COOH, showed incomplete dissolution or swelling at 70°C. Most inorganic solvents are insoluble in the products.

3.4 In vitro cytotoxicity examination

The MTT test is a colorimetric test used to measure cytotoxicity and cell viability. Based on MTT, the cytotoxicity of the synthesized derivatives was assessed, as depicted in **Table 4**.

Compared to the control, the outcome of the verified derivatives 'CS-P1, CS-P2, and CS-P3' exhibit the minimal difference between them. Several previous types of research have confirmed that CS and CSSB derivatives are non-toxic to cells. Consequently, CS has numerous medical applications [60–62].

3.5 Antimicrobial assessment

The CSSB derivatives' antibacterial activity was assessed using the inhibition zone technique. **Table 5** illustrates the results. The outcomes show that CS and all CSSBs affect *E. coli* and *K. pneumonia* strains as CS. According to the study, CS can prevent *S. aureus* and *E. coli* from forming new cells [63]. CSSBs derivatives presented antimicrobial activity against *S. aureus*, with inhibition zones of "22 \pm 0.3, 20 \pm 1.2, and 19 \pm 0.62 mm" for "CS-P1, CS-P2, and CS-P3," respectively. The synthesized compounds also have antibacterial activity against *S. mutans*, with inhibition zones of "15 \pm 0.89, 17 \pm 0.50, and 18 \pm 1.20 mm," respectively. Two fungal strains were used to test the CSSBs' antifungal activity, and all the CSSBs tested showed positive results.

Comp. conc. (mg)	Viable cells in the presence of CS	Viable cells in the presence of CS-P1	Viable cells in the presence of CS-P2	Viable cells in the presence of CS-P3				
25	99 ± 0.83	99 ± 0.73	99 ± 0.60	99 ± 0.75				
50	97 ± 0.73	99 ± 0.91	98 ± 1.3	99 ± 0.89				
100	94 ± 0.63	98 ± 0.74	98 ± 1.2	98 ± 0.50				
150	93 ± 0.88	97 ± 0.65	98 ± 0.62	96 ± 1.2				
200	89 ± 0.53	90 ± 0.72	90 ± 0.74	91 ± 1.2				
The experiment was repeated three times and mean was calculated.								

Table 4.

The CS and SB derivative cytoxicity assessment.
	Gram-negative bacteria		Gram-positive bacteria		Fungi	
Comp. codes	E. coli	K. pneumonia	S. aureus	S. mutans	C. albicans	Ammophilus fumigatus
CS	24 ± 0.63	26 ± 0.73	NA	NA	26 ± 0.79	16 ± 0.83
CS-P1	22 ± 0.73	28 ± 0.91	22 ± 0.3	15 ± 0.89	34 ± 0.99	26 ± 0.91
CS-P2	27 ± 0.83	27 ± 0.72	20 ± 1.2	17 ± 0.5	31 ± 1.29	25 ± 0.72
CS-P3	22 ± 0.98	26 ± 0.65	19 ± 0.62	18 ± 1.2	26 ± 0.49	21 ± 0.65
NA means not detected.						

Table 5.

Results of antibacterial and antifungal action of CS and CSSBs.

According to the published studies, numerous mechanisms are expected to explain how CS acts on bacteria, which vary depending on the metabolic process, the type of microorganism, and the cell wall composition. The initial recommendation is to disrupt the organism's cell wall electrostatic attraction between the positively charged amine in CS and the negatively charged residue group in bacterial nucleic acid cellular components such as COO^- or $PO4^{-2}$. The interface of bacterial DNA with CS is proposed as the second procedure. Protein and messenger RNA perversion into bacterial cells is induced by CS, followed by nuclei. Another theory relies on the ability of CS to form metal complexes. Metal complexes such as Zn^{2+} , Mg^{2+} , and Ca^{2+} are examples. Metals are required for bacterial metabolic and growth processes [63].

3.6 Conclusion

The hunt for new antibiotics has risen in tandem with the rise in antibioticresistant microorganisms. CS may be good material in this field. Cyprinus scales extracted 89% of the CS from local market-purchased Carpio fish. Acid-base titration was used to determine the DD % and describe how CS is put together. Three new CSSB derivatives that included the branches' distinct parts were synthesized, and their formations were confirmed using "FT-IR and ¹H and ¹³C NMR spectroscopy." The new SB configuration was antibacterial against many bacteria and fungi tested. CSSBs had practically no effect on cytotoxic mouse fibroblast cell lines after being prepared. As a result of the findings above, it is reasonable to believe that the prepared CSSBs could be utilized with high efficiency, care, and performance in numerous biomedical fields.

4. A novel prodrug of methotrexate based on chitosan and evaluation of their bioactivity

Methotrexate (MTX) is the most effective medicine for treating several types of cancer, including colon cancer. Nevertheless, this medication can reduce the bioavailability of the goal material. It is administered orally and rapidly digested. MTX is an antimetabolic agent that inhibits folic acid metabolism. 1948 marked the beginning of clinical use of the drug as an anticancer agent, following the finding of its abnormal effects on DNA combination [64]. Due to the drug's physical and chemical characteristics, oral administration results in sedate retention at the beginning of the gastrointestinal tract. The drug's numerous significant limitations and disadvantages were demonstrated once aimed directly at a specific position of absorption or a particular portion of the alimentary channel, for instance, the colon. As a result, it is critical to explain how the medicine arrived at a specific treatment location [65]. Recent preparation of the prodrug involved covalently attaching the colon drug to the carrier. These prodrugs frequently alter their physical and chemical characteristics to improve infatuation at the activity site, increase the duration of action, and reduce toxicity and side effects [66]. This project is designed to develop a prodrug for colon cancer treatment by incorporating MTX into a biopolymer. The MTX was then converted to "methotrexate – imidazole" and loaded into CS to create colon cancer prodrugs containing CS-MTX conjugates. The structure of the synthesized derivatives was confirmed using spectroscopic analysis. The compound's chemical stability and hemolytic activity were also investigated. The drug concentration percentages were calculated. The "MDA-MB-231, MCF-7, and MDA-MB-453" cell lines were used to test the cytotoxicity of the prepared derivatives in a dish with MTT.

4.1 Synthesis of CS-MTX

The CS-MTX compound was developed as a possible treatment for MTX-induced colon cancer. As previously described, the CS was extracted and used to prepare the CS-MTX. As shown in **Figure 11**, methotrexate and imidazole were used to kickstart the process. To make MTX-imidazole *in situ*, equal quantities of "MTX and imidazole" were added to the mixture. A small quantity of "*N*, *N*-carbonyl diimidazole (CDI)" was added. At a concentration of 2%, CS was dissolved in GAA. The catalyst, triethylamine (TEA), was added in a few drops [67].

The confirmation of the CS-MTX compounds was established by "NMR and FTIR spectroscopy." A downfield signal at 6.02 Hz on the ¹HNMR chart indicated the proton was linked to C1 of the sugar (**Figure 12**). The proton signal was observed at 3.60, which is connected to C1. Since the two protons were in diverse environments, the methylene group C5 showed signals at 2.53, and C6 led signals at 2.08 and 2.28. The carbonyl signals were shown in the ¹³C NMR spectrum (**Figure 13**) at 173.2 and 173.4. The value of the anomeric carbon was 101.5. The IR spectrum of the conjugate (**Figure 14**) revealed the CS and MTX vibrations. The N-H group had peaks at 3064 cm⁻¹, while the C-H group had 3032–2834 cm⁻¹. At 1660 and 1720 cm⁻¹, COOH group C-O vibration-related signals were detected. The (C-C, C-N, and CO) vibrations were found in signals between 1499 and 1183 cm⁻¹. A new peak at 1700 cm⁻¹ confirmed the development of an amide bond between MTX and CS. This bond was vulnerable to acid hydrolysis or the amidase enzyme, allowing the drug to be released.

4.2 Determination of drug contains

The percentage of MTX-CS was calculated using UV-visible spectroscopy. The initial conditions for this experiment were based on the amide bond hydrolysis between MTX and CS. As expected, increasing the amounts of CS and MTX in the conjugates improved the results. The CS-MTX (1:1) contained 0.60% w MTX, and the yield was 65.0%.CS-MTX (1:2), on the other hand, had a content of 0.72% w and produced 70%. All reactions were carried out at 50°C for 20 hours. The proportion of medicine left was determined using this equation [68]:

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$$\binom{w}{w}$$
 of MTX loading = $\frac{MTX \text{ amount}}{CS - MTX \text{ conjugates amount}} \times 100$ (3)

4.3 Hemolytic exercise

The percentage of hemolysis produced by CS-MTX derivatives was investigated at various concentrations. The proportion of hemolysis increased as the concentration of CS-MTXs increased, as shown in **Figure 15**. According to the test results, the CSMTXs'



Figure 11. Synthesis of CS-MTX.



Figure 12. The ¹H NMR of "CS-MTX."



The ¹³C NMR of "CS-MTX."



Figure 14. The FT-IR of the "CS-MTX."

hemolysis rate was less than 4.5 percent, below the international standard of less than 5% [69]. This assessment was conducted depending on the method described [70]. The rate of hemolysis increases as the level of CS-MTX rises. The hemolysis rate in the CSMTXs was less than 4.5 percent, less than the international standard of less than 5 percent. White rabbit red blood cell (RBC) samples were used in this experiment. A 0.9 percent of NaCl solution was added after 3 mL of blood was extracted and centrifuged for 20 minutes at 4000 rpm. The controls were made by mixing 1 mL of red blood cells and distilled water with 5 mL of normal saline. A sequence of solutions containing both derivatives was carried out by adding 6,2,0.6,0.3,0.7 g of MTX and the prepared combination to tubes, followed by 2 mL of normal saline and 1 mL of red blood cells. The resulting solutions were kept for 2 hours at 37°C in a water bath. Separately, the tube was centrifuged at 4000 rpm for 20 minutes. At a wavelength of 541 nm, UV absorbance was measured. The following equation used to calculate the percentage of hemolysis was as follows:

$$%Hemolysis = \frac{A \text{ sample} - A \text{ negative control}}{A \text{ positive control} - A \text{ negative control}} \times 100$$
(4)

4.4 Intracellular cytotoxicity

Three human breast cancer cell lines were used in this project "MDA-MB-231, MCF-7, and MDA-MB-453." The cell lines' viability was decreased by CSMTX

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Figure 15. *The hemolysis caused by 'CS-MTX' treatment.*

conjugates in a dose-dependent manner, according to MTT analyses. CS-MTX cytotoxicity was variable across all cell lines tested. The IC50 values for "MCF-7, MDA-MB-231, and MDA-MB-453 were 363.5±31.2, 198.8±20.4, and 163.4±10.8 g of CS-MTX/mL," respectively, over 24 hours. However, the cell "MDA-MB-453" used CS-MTX more precisely than the other cell lines in **Figure 16**. The cytotoxicity of CS MTX and the free medicine MTX was studied using "MCF-7, MDA-MB-231, and MDA-MB-453 cells," and the effect was calculated using the MTT assay [71]. A 96-well plate was used to test a range of CS-MTX and MTX concentrations (1–10 M). These concentrations were added before coating the plate with roughly 3 x 103 cells. After that, the plate was kept for 12, 24, and 48 hours. After removing the previous media, new



Figure 16.

MTT viability tests cell lines after mixing 200 g/mL in DMSO as a control for 24 hours. "Data were calculated as mean SD (n = 3) with a P-value of 0.01 compared to the control." Each solution's final exhibition was equal to 0.5 percent.

media were added, including (DMEM/F12) supplemented with 15 L of (MTT, concentration—500 g/mL). The cells were then grown at 37°C for 3–4 hours. The darkblue formazan crystals were dissolved in DMSO. A Cytation three multimode plate reader manufactured in the United States was used to calculate their absorptivity at 574 nm for each well. The equation below was used to convert the obtained absorbance values into applied rate cells for unprocessed control cells:

Relative cell viabillity =
$$\frac{\text{Absorbance of the sample}}{\text{Absorbance of the control}} \times 100$$
 (5)

4.5 Conclusion

The synthesized CS-MTX derivatives are most likely human colon cancer prodrugs. Under acidic conditions, the prepared derivatives were satisfactorily stable at a pH of 1.2. The synthesized compounds had a long half-life value of 4.52 in acidic conditions and 16.01 in basic media than the original drug. The CS-MTX conjugate produced significant results in the MTT assay. The three "cancer cell lines MCF-7, MDA-MB-231, and MDA-MB-453" showed a dose-dependent reduction in viability compared to the origin medicine. The IC50 values were "363.53 \pm 1.2, 198.82 \pm 0.4, and 163.41 \pm 0.8 g of CS-MTX/mL after 24 hours." These findings suggest that CS-MTX as a prodrug could be helpful in the treatment of colon cancer. Additional tests are being conducted to estimate the synthesized prodrug's various biological activities.

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Chapter 4

Chitosan Nanoparticle: Synthesis, Characterization, and Use as Plant Health Materials

Pranab Dutta, Arti Kumari and Madhusmita Mahanta

Abstract

Chitosan is a naturally occurring biopolymer having multifaceted applications in agriculture, medicine, food industry, and cosmetics. The association of this natural biopolymer with nanotechnology can produce revolutionary effects in plant protection and agriculture. Nano-chitosan can be fabricated using various methods. However, the green synthesis approach has gained attention in recent years. The green engineered nanoparticles are economical, energetically feasible, and environmentally benign. The biosynthesized nano-chitosan has evolved as a potential plant protection agent. Chitosan nanoparticles possess antifungal, antibacterial, and antiviral properties, and are found to be effective against seed-borne and soil-borne pathogens. Nano-chitosan also behaves as an effector molecule and induces local and systemic defense responses in plants. The mode of action of nano-chitosan involves alterations in membrane permeability, replication, cytoplasmic alterations, induction of defenserelated genes, and cell lysis. Furthermore, chitosan nanoparticles can be used for soil improvement and can reduce pest and pathogen attacks, thereby promoting the growth of plants. The authors outline the methods of synthesis and characterization of chitosan nanoparticles, their utilization in plant protection and growth promotion, along with the underlying mechanisms.

Keywords: chitosan nanoparticles, biopolymer, green synthesis, characterization, plant protection

1. Introduction

Agriculture is a primary activity upon which the economic status of a country relies. The produce obtained from agricultural activity serves the purpose of mitigating domestic hunger as well as earning foreign currency. This demands the high production and productivity of crops. However, crop health, quality of production, and productivity are attributed to different biotic and abiotic factors associated with it. It has been already said that 20–40% of crop loss occurs around the globe due to attack of pest and diseases [1] that limits the yield of crops. The detrimental effect of chemical pesticides on Earth has resulted in the development of green strategies, such as sustainable farming, with the use of resistant varieties, biopesticides, nano bioformulations, integrated disease, and pest management.

As the interest of the scientific community shifted from chemocentric to sustainable agriculture, it opened up a vast possibility of exploiting nature-based biodegradable materials with potential biocontrol efficacy for plant disease management. With more intense research in this field led the scientists to speculate that the nano-sized materials may perform excellent activity as compared to the base source used for their synthesis. Therefore, the development of nano bioformulations and their use is encouraged to achieve the goal of sustainable farming. Different biopolymers viz., cellulose, starch, alginate, chitin, and chitosan, are used for the development of new materials with noble functionality and environmental sustainability. Among these, chitosan is the second most abundant biopolymer found in nature which is used widely due to its unique characteristics, such as abundance, large surface-to-volume ratio, biodegradability, biocompatibility, pH sensitivity, non-toxicity, and a safe alternative [2, 3]. Apart from that, nano-chitosan possesses antifungal and antibacterial activity along with other plant growth-promoting traits [4, 5] which makes it promising in several aspects of plant growth and development.

This chapter aims to briefly review the importance, green synthesis, characterization, mode of action, and successful use of chitosan nanoparticles (ChNPs) for the effective management of plant diseases.

2. Importance of chitosan nanoparticles

Chitin and chitosan are the primary components of crustacean shells, such as shrimp, squid, crab, and lobster, the exoskeleton of terrestrial insects viz., honeybees and silkworm, and cell walls of fungi like molds, yeast, and ray fungi, such as *Streptomyces* [6, 7]. Chitosan is a cationic biopolymer obtained by the whole or partial deacetylation of chitin. It is a linear polysaccharide consisting of (1–4)-linked 2-amino-2-deoxy-β-D-glucopyranose obtained after deacetylation of N-acetyl-D-glucosamine [8, 9]. The term chitosan does not specifically indicate a unique compound but a group of co-polymers owing to their degree of deacetylation, polymerization, molecular mass, viscosity, and acid dissociation constant, that is, pka [10]. The chitosan derived from microbial sources is considered as promising as the process underlying can be manipulated to prepare a pure and uniform product with desired specific characteristics [11]. It is a very versatile biopolymer having multifaceted activity in the field of medicine, agriculture, food industry, cosmetics, and sewage treatment (**Figure 1**) [12]. ChNPs are widely used due to their unique polymeric cationic character, absorption enhancing effects, mucoadhesive nature, biocompatibility, and biodegradability. Being a modified linear polysaccharide with varying numbers of free amino groups in their polymeric chain with cationic property, chitosan offers ionic cross-linking of multivalent anions, which is making it a significant biopolymer for the synthesis of nanoparticles [6, 13]. The positively charged ChNPs have more affinity toward the negatively charged biological membrane and site-specific targeting *in vivo* [14]. Chitosan when applied to foliage or soil, can elicit innate defense response within the plant to resist insect and pathogen attack [5] by the production of antifungal hydrolases, phytoalexins, or by inducing structural barriers via the synthesis of lignin-like material [15, 16]. Further, ChNPs significantly have a positive impact on the biophysical properties of the plant. The application of nano-chitosan increases the rate of photosynthesis, induces root nodulation, upregulates nutrient uptake, enhances the rate of germination of seed, and boosts plant vigor [5, 17]. It is widely used in the delivery of fertilizers, micro-nutrients, and pesticides as it ensures



Figure 1. Applications of nano-chitosan.

slow release of the drugs with enhanced solubility [10]. Moreover, chitosan-mediated genetic transformation is also a successful one as it forms a complex via electrostatic interaction and protects the nucleic acid from nuclease degradation. It gives rise to stably transformed plants as compared to those developed via traditional methods of gene delivery [18, 19].

3. Green synthesis of chitosan nanoparticles

Green synthesis is a novel method of synthesis of NPs using microorganisms, such as bacteria, fungi, actinomycetes, and botanicals [20–25]. It is a bottom-up approach to nanoparticle synthesis which interconnects two disciplines of science, that is, nanotechnology and biotechnology. Biogenic synthesis of nanoparticles is preferred over chemical or physical synthesis methods as it is a safe, environmentally benign, economically and energetically feasible, less time-consuming process, and it makes optimum use of the redox potential of metabolites produced by biological entities to convert the macromolecules to nano form [9, 21, 26–30]. The main principle of the biogenic synthesis of the nanoparticle is based on the redox reactions that occur when the microorganisms/biological entity grabs the metal ion and detoxifies it to element metal through the enzymatic activity of the cell. It can be categorized into intracellular and extracellular synthesis. In intracellular synthesis, the metal ions are transported into the cell and the reduction reaction occurs within the cell cytoplasm, cell wall, and/or periplasmatic space, therefore, the resultant nanoparticles form inside the cell. However, the latter involves the synthesis of nanoparticles on the cell surface via the catalytic activity of reductase enzyme upon the trapped foreign entity [31]. Harvesting of nanoparticles from the cell matrix is a tedious process in intracellular synthesis which is why extracellular synthesis of nanoparticles is preferred mostly. The microbes are often regarded as eco-friendly green nano-factories due to their large-scale production ability of nanoparticles with relative control over their size and shape (regulated by surrounding environmental conditions) with a simpler process of production [26].

In phytofabrication of nanoparticles, generally, the metal ions are added to the plant part extract and then continuously stirred in a magnetic stirrer. The formation of nanoparticles can be confirmed visually when the resultant solution changes its color [30, 32]. It is due to the activity of the antioxidant, such as polyphenol, flavonoids, and phytoalexins, that act upon the metal ion and convert it to nontoxic element metal in nanoform. Therefore, a plant with a higher percentage of natural reducing constituents is desirable for phytofabrication of nanoparticles. Pelargonium graveolens L'Her commonly known as the rose geranium plant is a rich source of natural antioxidants. Its leaves and essential oil have several therapeutic applications in the field of pharmacology [33, 34]. El-Naggar et al. [9] mixed an equal volume of chitosan solution and phytoextract of *P. graveolens* and incubated the mixture at 50°C in a rotary shaker. The resultant turbid solution indicating the ongoing redox reaction is then centrifuged at $10,000 \times g$ for 10 min. It was further washed with an acetic acid solution to remove the unreacted chitosan and the ChNPs present in the solution are extracted by subjecting it to freeze drying. Similarly, Boruah and Dutta [5] biogenically synthesized ChNPs from fungal sources rich in chitosan. Isolation of chitosan was done by treating the fungal biomass with a series of alkali and acid treatments under controlled conditions. Initial treatment of fungal biomass with NaOH yields an alkali-insoluble material (AIM) that was further subjected to an acid treatment to extract the fungal chitosan. It was then converted into nano-chitosan by adding 1% TPP solution in a magnetic stirrer.

Guzman et al. [35] synthesized colloidal AgNps by a combination of ultrasonication and chemical reduction methods using gallic acid and chitosan, respectively, and during characterization, they found that as-synthesized gallic acid-chitosan modified silver nanoparticles (GC-AgNps) were monodispersed, spherical shape with an average size of 26.23 ± 9.92 nm, and stable for four weeks without any noticeable change in size. GC-AgNps were found highly effective against *Escherichia coli* even at 1μ g/mL after 120 min of exposure.

4. Characterization of chitosan nanoparticles

The nanoparticle formation is greatly affected by the processing conditions and time. They are characterized on the basis of their surface plasmon resonance, morphology, particle size distribution, zeta potential, functional group analysis, etc., using UV-vis spectrophotometer, electron microscope, dynamic light scattering, Fourier transform infrared (FTIR) spectroscopy [36]. Moreover, atomic absorption spectroscopy is done to study the release profile of Ch-encapsulated nanoparticles.

4.1 Characterization by UV-vis spectroscopy

UV-vis spectroscopy is the confirmatory analysis that ascertains the formation of nanoparticles by surface plasmon resonance. UV-vis absorption spectroscopy is used to examine the optical properties of ChNPs obtained from the commercial production of chitosan showed an absorption band at 330.25 nm, whereas those obtained from biogenic sources exhibited the absorption band at 310–342 nm [5]. Kain and Kumar [37] reported a band at 200–300 nm for the biogenically synthesized ChNPs from common yarrow. Moreover, AbdElhady [38] studied the UV-vis spectrophotometer reading of ChNPs prepared at two different temperatures, that is, 40°C, 60°C, and 80°C, and found that the absorption band was obtained at 356, 348, and 353 nm, respectively.

4.2 Characterization by dynamic light scattering

The main principle of DLS is based on the Brownian movement of particles/ molecules present in the solution that results from their collision with the randomly moving solvent particles. A laser beam is passed through the sample, and the fluctuation in scattered light due to the random motion of particles is detected by the photon detector. DLS is used for the measurement of average particle size, particle size distribution, polydispersity index (PDI), and zeta potential. PDI explains the polydispersity or monodispersity of particles in an aqueous medium. PDI value greater than 0.5 represents polydispersity and less than 0.5 normally shows the monodispersity of particles. Generally, monodisperse ChNPs exhibit the PDI value within the range of 0.2-0.4 [2]. Further, the surface charge of nanoparticles, also known as zeta potential, explains the stability of the nanoparticles, which is measured in the range of ±30 mV. The ChNPs show a positive zeta potential value that may vary from 11.2 \pm 1.2 mV to 18.7 \pm 0.4 mV [5, 39]. Further, the appropriate particle size determined by DLS showed that the ChNPs are nearly spherical in shape with size ranging from 150 to 350 nm [40]. However, Sivakami et al. [41] prepared ChNPs by cross-linking low molecular weighed chitosan with TPP and found a minimum particle size <100 nm. Similarly, Boruah and Dutta [5] reported that the biogenically synthesized ChNPs exhibited their size within the range from 78.36 to 300.1 nm. A mean particle size of 50 nm was obtained by Sahab et al. [42] for the chitosan poly acrylic acid nanoparticles.

4.3 Characterization by electron microscopy

The internal and external morphology of the nanoparticles is studied using a transmission electron microscope and scanning electron microscope. Transmission electron microscopy revealed that the ChNPs are often spherical shaped with an amorphous nature [5, 14, 43]. Similarly, scanning electron microscopy studies conducted by many researchers found that the ChNPs are nearly spherical shaped with the smooth external surface [44, 45]. Parida et al. [46] obtained round ChNPs with a 78 nm diameter. Similarly, Kain and Kumar [37] found that the green synthesized ChNPs of common yarrow (*Achillea millefolium*) are smooth surfaced, spherical with a diameter less than 100 nm with a smallest diameter of 4.15 nm.

4.4 Characterization by Fourier transform infrared spectroscopy

FTIR study is conducted to confirm the synthesis of nanoparticles by determining their functional groups. Sample preparation for FTIR is done by gently triturating it with KBr which is then compressed into disks. The compressed disks are scanned against a blank KBr pellet background at 25°C to obtain the FTIR results. For every spectrum, a 32 scan interferogram was collected at transmittance/absorbance mode in the 4000–400 cm⁻¹ region [2]. The functional groups of a chitosan nanoparticle consist of amide ($-NH_2$) and hydroxyl (-OH) group, C-H, C-N, C-O, and P=O stretching [5]. Generally, the FTIR peak at 3000–3500 cm⁻¹ attributed to (-OH) and ($-NH_2$) is the confirmatory peak for the formation of ChNPs [46]. Sharma et al. [45] obtained a wider peak of the hydroxyl group (3200–3600), which led them to conclude that hydrogen bonding is enhanced in ChNPs when analyzed by FTIR. Choudhary et al. [2] found a band at 3424 cm⁻¹ for the synthesized chitosan nanoparticles that represent the stretching vibration of the combined peaks of the amide

 $(-NH_2)$ and hydroxyl (-OH) group. Kain and Kumar [37] reported FTIR peaks at 3317.48, 2139.29, and 1638.46 as a confirmation of the formation of ChNPs. Similarly, Boruah and Dutta [5] obtained a strong and broadband at 3250 cm⁻¹ signifying the stretching between hydroxyl and amide groups. They have also found other peaks at 2865 cm⁻¹, 1182 cm⁻¹, 1642 cm⁻¹, and 1182 cm⁻¹ is attributed to C–H stretching, asymmetric C–O stretching, stretching between C=O and N–H banding, and P=O.

5. Application of chitosan nanoparticles in plant health management

ChNPs have emerged as a potential antimicrobial agent and found effective against numerous phytopathogens. Nano-chitosan is reported to be effective against seedborne as well as against soil-borne pathogens. It behaves as an elicitor of plant defense responses, inducing both local and systemic defense responses. Thus, nano-chitosan can be used as a plant health material by protecting the plant from biotic and abiotic stresses and by promoting the growth of the plants.

ChNPs possess a greater affinity toward the membrane of microorganisms and can easily penetrate the pathogen's cell [47]. The smaller size and greater surface area of ChNPs increase the antimicrobial efficiency of chitosan biopolymer. Several studies confirmed ChNPs as an effective plant health management agent due to their dual role as plant protection and plant growth stimulating agents.

ChNPs have been known to possess antifungal properties against numerous phytopathogenic fungi. Nano-chitosan exhibit greater efficacy as compared to its bulk counterpart as NPs can negotiate cell wall and cell membranes more effectively due to their unique physico-chemical properties. It can effectively inhibit the development of phytopathogenic fungi at any stage of their life cycle. Chitosan can completely inhibit spore germination, germ tube elongation, and mycelial growth of fungi [48], and it can penetrate the cell membrane by plasma membrane permeabilization and results in cell lysis [49]. Boruah and Dutta [5] biogenically synthesized chitosan nanoparticles using four different fungal sources viz., Beauveria bassiana, Fusarium oxysporum, Trichoderma viride, and Metarhizium anisopliae. In vitro assay suggested that synthesized ChNPs in combination with *T. asperellum* was effective in suppressing mycelial growth of soil-borne fungal pathogens viz., Rhizoctonia solani, Fusarium oxysporum, and Sclerotium rolfsii. Abdel-Rahman et al. [50] studied the efficacy of chitosan (Ch) (2 and 4 g/L) as well as ChNPs (0.2 and 0.4 g/L) against blue rot disease of apples caused by *Penicillium expansum*. They observed that ChNPs performed better than compared to their bulk counterpart for both natural and artificial infections. Also, the fruit quality parameters, such as firmness, titratable acidity, and total soluble solids, were kept intact. The expression of defense-related genes *viz.*, chitinase, β -1,3-glucanase, peroxidase, phenylalanine ammonia lyase-1 (PAL1), xyloglucan endotransglycosylase (XET), and pathogenesis-related protein (PR8), were also upregulated indicating the development of systemic acquired resistance in plants against the pathogen. Saharan et al. [51] synthesized ChNPs using the ionic gelation method and its efficacy was determined against phytopathogenic fungi viz., Macrophomina phaseolina, Alternaria alternata, and Rhizoctonia solani. They observed a decline in the radial growth of the fungi in a dose-dependent manner. Muthukrishnan and Ramalingam [52] synthesized ChNPs biogenically mediating Penicillium oxalicum. The nanomaterial was found effective against Fusarium oxysporum ciceri, Pyricularia grisea, and Alternaria solani with the rate of inhibition 87%, 92%, and 72%, respectively. Also, seed treatment with ChNPs exhibited positive

morphological effects, such as enhanced germination percentage, seed vigor index, and biomass content in chickpeas. The efficacy of nanomaterials depends on their size, charge, and permeability through biological membranes. Again, the *in vivo* assay conducted under detached leaf condition observed 100% suppression of blast disease symptoms when treated with ChNPs prepared using the ionic gelation method [53]. Kheiri et al. [54] synthesized ChNPs from chitosans of different molecular weight and observed their antifungal activity against *Fusarium graminearum* causing fusarium head blight in wheat. The dynamic light scattering analysis showed a variable size of synthesized nanomaterials (180.9, 339.4, 225.7, and 595.7 nm). The inhibitory effect of these NPs was tested at different concentrations and maximum mycelial growth reduction (77.5%) was observed at 5000 ppm. The results obtained from greenhouse trials indicated a decline in the area under the disease progress curve (AUDPC) in NP-treated plants.

The reports of ChNPs as an antibacterial agent against plant pathogenic bacteria is very scarce and need further thrust in this domain. ChNPs were found effective against Ralstonia solanacearum causing bacterial wilt of tomato and potato. In vitro experiment indicated an increase in the inhibition zone with increasing concentration of nano-chitosan and found highest at 200 µg/ml concentration. In vivo assay revealed foliar application of nano-chitosan led to a decline in the disease incidence and severity in bacterial wilt-infected tomato and potato plants [55]. ChNPs directly interact with the bacterial cell wall and may cause modification in the external shape, loss of flagella, and lysis of the cell. The RAPD-PCR results showed differences in the genotype of treated *Ralstonia solanancearum* as compared to the genotype of untreated isolates [55]. The antibacterial activity of Cu-chitosan nanoparticles against Pseudomonas syringae pv. glycinea causing bacterial blight of soybean was reported by Choudhary et al. [56]. Concentration of 1000 ppm was found most effective in controlling the bacterial pathogen. An in vitro assessment was conducted with ChNPs and chitosan nanocomposites with lime essential oil and thyme critical oil against Pectobacterium carotovorum. The results indicated that chitosan nanocomposites with thyme essential oil were effective in producing an inhibition zone [57]. ChNPs were also observed to possess potentially high antibacterial activity against P. fluorescens and Erwinia carotovora causing bacterial soft rot [58]. Oh et al. [8] reported the antibacterial activity of ChNPs against phytopathogenic bacteria viz., E. carotovora subsp. carotovora and X. campestris pv. vesicatoria. Santiago et al. [59] biogenically synthesized Ch-derived NPs containing AgNPs and observed its antibacterial efficacy against R. solanacearum causing bacterial wilt of tomato. They found Ag-NP entrapped chitosan as a suitable alternative to chemical bactericides. Cs/TiO2NPs were found effective against the most dreaded bacterial pathogen of rice viz., X. oryzae pv. oryzae [60].

The chitosan biopolymer was found to inhibit the systemic propagation of viruses and virus-like organisms in infected plants and induce a host hypersensitive response against the viral pathogen [61–63]. The molecular weight of chitosan affects the degree of suppression of viral infections [64]. However, none of the studies has practically proved the ability of the chitosan molecule to absolutely inactivate the virus particles. Most of the studies have reported the inactivation of viral replication that prevents the multiplication and subsequent spread of the virus particles systemically. It may be hypothesized that ChNPs having a smaller size and greater surface area can easily penetrate into host tissues and tightly binds with nucleic acid causing selective inhibition and ramification of virus particles. Lu et al. [65] reported inhibition of TMV in tobacco pants when oligochitosan (50 μ g ml⁻¹) was applied 24 h before inoculation. The epidermal cells of tobacco leave treated with oligochitosan showed an increase in

Chitosan nanomaterial	Application	Target pathogen	Effect	Reference
Chitosan nanocomposite with T. asperellum	Antifungal	Rhizoctonia solani, Fusarium oxysporum and Sclerotium rolfsii.	Reduction in mycelial growth	[5]
Chitosan NPs	Antifungal	A. alternata, M. phaseolina and R. solani	Reduction in mycelial growth	[51]
Chitosan	Antifungal	Alternaria kikuchiana Tanaka and Physalospora piricola	Inhibition of spore germination, germ tube elongation, and mycelial growth	[48]
Rhodamine-labeled Chitosan	Antifungal	F. oxysporum	Inhibition of mycelial growth, cell membrane permealization, and lysis	[49]
'hitosan NPs	Antifungal and plant growth promotion	Penicillium expansum	Reduction in natural and artificial infections. Induction of defense-related genes and resistance in plants	[50]
Chitosan NPs	Antifungal and plant growth promotion	Sclerospora graminicola	Promote seed germination and seedling vigor, induced systemic and durable resistance	[66]
Ag-chitosan nanocomposites	Antifungal	Seed-borne fungi	Inhibition of mycelial growth	[67]
Cu-chitosan nanocomposites	Antifungal	Fusarium graminearum	Inhibition of mycelial growth	[68]
Chitosan Janocomposites	Antibacterial	Ralstonia solanacearum	Inhibition zone production, cell wall modification, loss of flagella and cell lysis	[55]
Chitosan nanocomposites with thyme critical oil	Antibacterial	Pectobacterium carotovorum	Inhibition zone production	[57]
Cs/TiO2NPs	Antibacterial	X. oryzae pv. oryzae	Reduction in infection	[60]

Chitosan nanomaterial	Application	Target pathogen	Effect	Reference
Oligochitosan	Induction of resistance in plants against TMV	Tobacco Mosaic Virus	Increase in the levels of intracellular H ₂ O ₂ , NO, and phenylalanine ammonia-lyase (PAL)	[65]

Table 1.

Antimicrobial properties of chitosan-based nanomaterials.

the levels of intracellular H₂O₂, NO, and increased activity of phenylalanine ammonialyase (PAL) indicating induction of plant defense response against TMV (**Table 1**).

6. Mode of action of chitosan nanoparticles

6.1 Direct activity against pathogen

ChNPs can directly interact with the cellular membrane of microorganisms due to their unique physicochemical properties and can easily permeate into the cytoplasm. The direct mode of action of ChNPs against fungi includes inhibition of spore germination, germ tube elongation, mycelial growth, and cell lysis. Benhamou [69] conducted ultrastructural studies and reported that chitosan induces numerous structural and morphological changes leading to distorted hyphae. This can be explained as the chitosan particles are polycationic in nature, it allows alteration in membrane permeability and cytoplasmic aggregations. As a result, the activity of enzymes involved in the synthesis and assembly of cell wall polymers are dwindled. The antibacterial effect includes disruption of the bacterial cell wall, cellular membrane, loss of external appendages, such as flagella, finally, leading to cell lysis. None of the studies have proved ChNPs to inactivate viruses and viroids. Most of the studies reported inhibition of virus replication, multiplication, and spread by chitosan. However, against pests and pathogens, ChNPs operate via an indirect mechanism, such as induction of host resistance.

6.2 Indirect mechanism

Chitosan molecule is generally used as an elicitor rather than an antimicrobial agent in plant disease control. It can be recognized by the plant PRRs and can trigger a cascade of defense responses. Chitosan molecule behaves as MAMP/PAMP or general elicitor and induces nonhost resistance in plants along with priming systemic immunity [70]. The cascade of biochemical and molecular reactions induced by chitosan includes enhanced H⁺ and Ca²⁺ influx into the cytosol, callose apposition, activation of MAP-kinases, hypersensitive response, oxidative burst, synthesis of phytohormones *viz.*, jasmonates and abscisic acid, as well as phytoalexins and PR-proteins [71].

6.3 Physical barrier in pathogen penetration

ChNPs can agglutinate around the penetration sites of the pathogen after its application on plant tissues and has two major effects. The first effect includes



Figure 2. Schematic representation of the mechanism of action of chitosan nanoparticles on plants.

isolation of the penetration site from healthy tissues by forming a physical barrier that prevents further spread of the pathogen. Around the isolated zone, several biochemical changes occur that lead to the elicitation of hypersensitive response, accumulation of H_2O_2 and other free radicals, which lead to cell wall fortification and induction of systemic acquired resistance. The second effect includes the initiation of wound healing process by binding with various materials (**Figure 2**).

6.4 Plant growth promotion

Nano-chitosan imparts an eustress effect on seedling germination and plant growth parameters, such as plant height, shoot length, root length, and biomass content, which have been confirmed through a series of studies. The study conducted on the effect of nano-chitosan on *Phaseolus vulgaris* L. under salt stress conditions revealed that 0.3% nano-chitosan was the best treatment in terms of germination, growth parameters. Also, significant increase in M.S.I, Chl.a, Chl.b, proline, catalase, carotenoids, and antioxidant enzymes were observed [72].

7. Conclusion and future prospects

Chitosan is a naturally occurring miracle compound having enthralling antimicrobial and eliciting properties. Nano-chitosan is gaining attention nowadays due to its greater efficacy and biosafety. Chitosan nanomaterials can be used in varied ways for plant disease management, thereby preserving crop quality and yield. In recent years, several findings have been gathered indicating nano-chitosan as a potential plant health material. However, more studies need to be channelized to unveil the exact mode of action of nano-chitosan specific to the pathosystem. Incorporation of chitosan nanomaterials into integrated pest management practices by devising suitable incorporation techniques need to be pursued. The biopolymer-based nanomaterials need extensive exploration owing to their multifunctional properties and diverse mechanisms. In the coming years, the use of nano-chitosan for combating biotic and abiotic stresses and transport of agrochemicals would be a promising discipline for utility in sustainable agriculture.

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Chapter 5

Chitosan Is the Ideal Resource for Plant Disease Management under Sustainable Agriculture

Magdi A.E. Abdellatef, Eman Elagamey and Said M. Kamel

Abstract

In the spirit of returning to nature and using scientific applications to raise plant efficiency and reduce pathogen risk, scientists began searching for safe, natural alternatives to pesticides that are highly effective and low cost. On top of these alternatives, chitosan came with its biodegradability, biocompatibility, antimicrobial activity, and nontoxicity, which granted it dual energetic effects during the hostpathogen interaction. Chitosan promotes plant growth, regulates plant cell homeostasis and metabolic processes, and triggers plant defense mechanisms; on the other hand, it inhibits the ability of pathogens by disrupting pathogen growth and reducing reproduction, wherefore chitosan will become an increasingly prevalent and ideal resource for agricultural sustainability.

Keywords: chitosan, eco-friendly, antimicrobial activity, abiotic stress, defense responses

1. Introduction

Diverse plant species are adversely affected by phytopathogens on a large scale, which results in severe economic loss and microbial toxins in food and animal feed. The conventional approaches to plant disease control are no longer efficient and secure, especially in light of the occurrence of severe climatic changes. Researchers had to change their traditional thinking in the process of combating plant diseases, which depends mainly on the use of chemical pesticides. In line with the strategy of sustainable agriculture, natural compounds, especially chitosan, have been put into use as safe alternatives that are effective in control, improving plant properties, and keeping the biological balance between plants and beneficial microorganisms in the soil.

Chitosan is a natural, biodegradable, and environmentally eco-friendly polysaccharide obtained from the exoskeleton of crustaceans such as shrimp, shellfish, crabs, cuttlefish, squid pen, and crawfish through the deacetylation process of chitin [1]. Moreover, chitosan can be produced from fungal chitin [2]. Chitosan has a unique nature as it is a linear polymer consisting of two subunits linked together, namely D-glucosamine and N-acetyl-D-glucosamine [3]. Chitosan concentration, chemical modification, acetylation degree, and molecular weight have all been identified as critical factors in the suppression of plant pathogen infection in the host plant [4, 5]. In sustainable agriculture, chitosan can be used alone or in combination with other compounds in plant nutrition with the aim of withstanding abiotic stress, stimulating plant defense systems, combating plant diseases, promoting plant seed germination and seedling growth [6, 7], and increasing crop production and quality. In addition, chitosan can purify soil and agricultural wastewater from heavy metals such as mercury, copper, uranium, and lead and thus can be reused for irrigation [8].

2. Chitosan chemical, physical, and biological features

Chitosan has many properties that have captured the interest of researchers over the past 20 years to explore the prospect of using it in a variety of scientific and practical sectors. Chitosan properties depend on several very important factors, such as molecular weight, degree of deacetylation, and solubility. The molecular weight of chitosan might influence the crystal size and morphological character of chitosan-based thin-film composites (TFCs) and other products or membranes [9]. The chitosan molecular weight may range between 50 and 2000 kDa depending on the source of chitin, and it has numerous influences on the viscoelastic properties of solutions and hydrated colloidal forms [10]. The degree of deacetylation determines the content of free –NH₂ groups in the polysaccharide and it has an influence on all the functional properties of chitosan [9]. Chitosan can dissolve in aqueous solutions, but it does so more readily in acidic media than in neutral or basic media [11]. The solubility of chitosan varies according to several factors, including polymer molecular weight, degree of acetylation, pH, temperature, and polymer crystallinity. Chitosan has unique chemical properties, e.g., linear polyamine, reactive amino and hydroxyl groups, and chelated metal ions. In addition, the biological properties of chitosan are biocompatibility, biodegradability, antimicrobial activity, biosafety, and nontoxicity [12]. These biological properties vary in influence on plants and their fungal and fungal-like microorganisms, bacterial, viral, viroid, and nematode pathogens according to the physical and chemical features of chitosan. Thus, chitosan compounds are strongly recommended to be used in the management strategies against phytopathogens such as viruses [4], bacteria [13], and fungi [14].

3. Antimicrobial activity of chitosan against phytopathogens

Several hypotheses have been postulated to explain the mechanism by which chitosan affects several phytopathogens:

a. Chitosan has polycationic nature enables it to interfere with electronegative charges on the outer surface of the microbial cell. The external electrostatic interaction between the positive amino glucosamine groups –NH3⁺ of chitosan and the negative charge on the cell surface exists in teichoic acids in gram-positive bacteria, lipopolysaccharides in gram-negative bacteria, and phospholipids in the fungal cell membrane (**Figures 1** and **2**), leading to changes in cell permeability and leakage of intracellular electrolytes and proteinaceous constituents and cell death [15, 16]. Divya et al. [17] suggest that the antimicrobial properties of chitosan are due to the repeated amino groups on the backbone of the polymer structure. Chitosan Is the Ideal Resource for Plant Disease Management under Sustainable Agriculture DOI: http://dx.doi.org/10.5772/intechopen.107958

b. Chitosan functions as a chelating agent of metals and vital nutrients, causing microbial starvation and impairing microbial development [18] as the amine



Fungal Cell

Figure 1. *Potential antifungal activity of chitosan.*



Figure 2.

Antibacterial effects of chitosan against (a) gram-negative bacteria and (b) gram-positive bacteria.

groups in the chitosan molecules are in charge of the uptake of metal cations by chelation. Chitosan metal-binding capacity increases at high pH since the amine groups are not protonated and the electron pair on nitrogen in the amine group is available for donation to metal ions [19]. The high molecular weight of chitosan might cause a reduction in cell membrane permeability due to a polymer coating on the surface of the cells that blocks cell access to nutrients [20].

c. The internal electrostatic interactions between the positive amino groups on the polysaccharide chain of low molecular weight chitosan and the negative phosphate groups on the nucleic acid chain of microbial cells lead to the inhibition of the synthesis of DNA/mRNA and a decrease in the abundance level of protein and enzymes [21].

Not all forms of pathogens are equally sensitive to chitosan (**Table 1**); therefore, the degree to which pathogens react to chitosan depends on a number of factors:

- a. The solubility of chitosan increases in acidic solutions and becomes insoluble in solutions with a pH greater than 6.5, according to the pKa values of its amino groups, which range from 6.3 to 6.5. Many articles have shown that chitosan is a great antimicrobial agent under acidic conditions [21]. Meanwhile, the pathogen inhibitory effect of chitosan depends on the type of solvent. Chitosan dissolved in lactic acid shows the best inhibitory effect as compared to that dissolved in formic acid and acetic acid [22].
- b. The ability of chitosan to exhibit intracellular antimicrobial action depends on its molecular weight. Low molecular weight chitosan has the highest inhibitory effect on *Rhizopus stolonifera* [23], while high molecular weight chitosan shows better efficacy on *Fusarium oxysporum* f. sp. *vasinfectum*, and *Alternaria solani* [25].

Furthermore, the abundance of polysaccharides and proteins that make up the numerous layers of the cell wall in both fungi and bacteria has a major impact on the mechanical strength of interaction with chitosan [73].

- c. The degree of deacetylation affects the antimicrobial properties of chitosan. Chitosan with a high degree of deacetylation exhibits a more positive charge, which results in stronger electrostatic interactions with the microbial cell surface and higher antimicrobial activity [13].
- d.The inhibition rate of pathogen growth depends on the concentration of chitosan [74].

3.1 Antifungal activity

Chitosan is efficient in inhibiting hyphal growth, mycelial elongation, spore formation, spore germination, spore viability, germinal tube, and fungal virulence factor production of phytopathogenic fungi [14]. The ability of chitosan to penetrate the plasma membranes of phytopathogens depends on the degree of membrane fluidity. Chitosan-sensitive fungi possess polyunsaturated fatty acid-rich membranes such as linoleic acid (high fluidity membrane), while chitosan-resistant fungi possess saturated

Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
Tomato wilt	Fusarium oxysporum f. sp. radicis- lycopersici	Antifungal	Chitosan was bound to negatively charged phospholipids that alter plasma membrane fluidity and induced membrane permeabilization.	[15]
Potato dry rot	Fusarium sulphureum	Antifungal	Chitosan caused morphological changes such as intertwisting hyphal, distortion, and swelling with excessive branching, abnormal distribution of cytoplasm.	[22]
Post-harvest fungi of fruits and vegetables	Rhizopus stolonifer	Antifungal	Low molecular weight chitosan caused inhibition of mycelial growth, while the high molecular weight chitosan affected spore shape, sporulation, and germination.	[23, 24]
Early blight of potato, tomato, capsicum, and eggplant	Alternaria solani	Antifungal	Chitosan reduced hyphal growth, inhibited sporulation and spore germination, and induced morphological	[25, 26]
Cotton wilt	Fusarium oxysporium f. sp. vasinfectum	Antifungal	changes.	
Apple dieback	Valsa mali	Antifungal	-	
Banana wilt	Fusarium oxysporum f. sp. cubense	Antifungal	Chitosan formed abnormal shapes, vesicles, and empty cells devoid of cytoplasm in the mycelia and agglomeration of hyphae.	[27]
Soybean sudden death syndrome	Fusarium solani f. sp. glycines	Antifungal	Chitosan was able to induce the level of chitinase activity in soybean. Furthermore, chitosan interfered directly with the fungal membrane function and interacted with fungal DNA and mRNA.	[28]
Fruit rots in mango,	Alternaria alternata	Antifungal	Chitosan caused aggregation of mycelium and structural	[24, 29]
strawberry, –	Botrytis cinerea	Antifungal	changes such as excessive —	[24]
peach	Penicillium expansum	Antifungal	and hyphae size reduction.	[24]
Bayoud disease in date palm	Fusarium oxysporum f. sp. albedinis	Antifungal	Chitosan elicited a defense reaction against this fungus in date palm roots, by eliciting peroxidase and polyphenoloxidase activity and increasing the level of phenolic compounds.	[30]

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Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
Cucumber wilt	Fusarium oxysporum f. sp. cucumerinum	Antifungal	Chitosan caused hindering of the Fusarium cell wall, disrupting DNA, disrupting structural and functional protein biosynthesis, and influencing metabolic pathways.	[31]
Early blight disease	Alternaria solani	Antifungal	Chitosan covered the cell wall, causing membrane disruption	[29]
Root rot disease	Fusarium oxysporum	Antifungal	and cell leakage. Moreover, chitosan penetrated into fungal living cells. leading	
Damping off disease	Pythium debaryanum	Antifungal	to the inhibition of various enzymes and interference with the synthesis of mRNA and proteins.	
Postharvest decay of oranges	Penicillium citrinum, Penicillium mallochii	Antifungal	Chitosan inhibited the growth of two <i>Penicillium</i> species by compacting fungal spores and reducing the activity of plant cell walls by degrading enzymes produced by fungi. Additionally, chitosan increased the wound healing process in orange fruits.	[32]
Anthracnose and ripen fruit rot in chilli	Colletotrichum capsici	Antifungal	Chitosan completely inhibited the mycelial growth of <i>C. capsici</i> by forming physical barriers around the penetration sites of the pathogen, preventing them from spreading to healthy tissues. In addition, chitosan activated the defense-related antimicrobial compounds, induced a decline in malondialdehyde content, and increased the concentrations of soluble sugars and proline, as well as peroxidase and catalase activities.	[33]
Anthracnose of camellia	Colletotrichum camelliae	Antifungal	Chitosan enhanced the activity of H ₂ O ₂ , the defense-related enzymes, such as polyphenol oxidase, peroxidase, catalase, phenylalanine ammonia-lyase, and the concentration of soluble proteins inside the camelia plant.	[34]
Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
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Fasciation of Arabidopsis thaliana	Staphylococcus simulans	Antibacterial	Chitosan caused multiple changes in the expression profiles of <i>Staphylococcus</i> <i>aureus</i> SG511 genes involved in the regulation of stress and autolysis, as well as genes associated with energy metabolism were observed.	[35]
Common scab in potato tubers	Streptomyces scabies	Antibacterial	Chitosan reduced disease incidence and disease severity by eliciting plant defense mechanisms and systemic resistance.	[36]
Bacterial leaf blight and leaf streak in rice	Xanthomonas oryzae pv. oryzae, Xanthomonas oryzae pv. oryzicola	Antibacterial	Chitosan caused membrane lysis and the destruction of biofilm of bacteria and increased the activities of phenylalanine ammonia-lyase, peroxidase, and polyphenol oxidase in rice seedlings.	[37]
Bacterial leaf spot of tomato	Xanthomonas vesicatoria	Antibacterial	Chitosan induced systemic resistance mechanisms in tomato by activation of several defense enzymes and elicited expression of PIN II and ETR-1 genes from several molecular pathways involved in pathogen defense.	[26]
Bacterial head rot of broccoli	Pseudomonas fluorescens	Antibacterial	Chitosan reduced the disease incidence, the lesion diameter and induced systemic resistance mechanisms in broccoli.	[38]
Bacterial brown stripe of rice	Acidovorax avenae subsp. avenae	Antibacterial	Chitosan caused membrane disruption and lysis, reduction of biofilm formation, and gene expression change.	[39]
Bacterial fruit blotch of watermelon	Acidovorax citrulli	Antibacterial	Chitosan significantly inhibited the growth of <i>A</i> . <i>citrulli</i> by damaging and altering the cell membrane, separating the cytoplasmic membrane from the cell envelope, coagulating the cytosolic components, forming a vacuole-like structure, and broking of cell walls, leading to the leaching out of a mass of nutrient and nucleic materials.	[40]

Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
Bacteria of crown gall disease	Agrobacterium tumefaciens	Antibacterial	Chitosan was bound with negatively charged components on the bacterial surface, via	[29]
Soft mold disease	Erwinia carotovora	Antibacterial	electrostatic interactions, causing changes in the permeability of the bacterial wall and shutting down the cell division, leading to bacterial fatality.	
Bean mild mosaic virus	BMMV	Antiviral	Chitosan led to the inhibition of systemic propagation of viruses over bean plants and enhanced the hypersensitivity response of bean plants to viral infection.	[5]
Tobacco mosaic virus in Arabidopsis thaliana	TMV	Antiviral	Chitosan induced TMV resistance in wild-type and jasmonic acid pathway- deficient (jar1) Arabidopsis plants enhanced the expression of the defence- related gene PR1.	[41]
Alfalfa Mosaic Virus n <i>Nicotiana</i> glutinosa	AMV	Antiviral	Chitosan induced systemic acquired resistance, increased total carbohydrates and total phenolic contents, as well as triggered the transcriptional levels of peroxidase, pathogen-related protein-1, and phenylalanine ammonia-lyase.	[42]
3anana root- xnot nematode	Meloidogyne incognita	Antinematode	Chitosan suppressed root-knot nematodes by decreasing the total number of galls, immature stages, females with egg masses, and the count and rate of build-up of nematodes in the soil.	[43]
Tomato root- knot nematode	Meloidogyne javanica	Antinematode	Chitosan reduced the number of nematodes per plant and promoted plant growth.	[44]
Carrot root- cnot nematode	Meloidogyne incognita	Antinematode	Chitosan caused a synergistic effect against <i>M. incognita</i> on carrots and could be utilized to manage nematodes as a potential sustainable treatment.	[45]
Rice root-knot	Meloidogyne graminicola	Antinematode	Chitosan reduced root-galling and nematodes development and induced a plant defense mechanism against nematodes.	[46]

Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
Thymus daenensis	Abiotic stress	Drought stress	Chitosan stimulated osmotic adjustment through proline accumulation and reduction of lipid peroxidase levels, which increased the integrity of cell membranes of thyme leaves.	[47]
Potato plants	Abiotic stress	Drought stress	Chitosan reduced membrane relative permeability and malondialdehyde concentration of potato leaves; raised the concentration of proline and soluble proteins; enhanced the activities of superoxide dismutase, and enhanced their antioxidation ability; increased the activities of protective enzymes and regulated the content of osmotic regulatory substances like peroxidase.	[48]
Apple seedlings	Abiotic stress	Drought stress	Chitosan acted as an exogenous antioxidant that enhanced resistance to oxidative stress during drought in apple seedlings.	[49]
Cowpea plants	Abiotic stress	Drought stress	Chitosan increased the thickness of the midrib region, mesophyll tissue, and the midrib vascular bundle, therefore reduced water stress in cowpea.	[50]
Sweet basil (Ocimum ciliatum, Ocimum. basilicum)	Abiotic stress	Drought stress	Chitosan reduced the harmful effect of water stress, increased plant growth parameters, and significant effected on total phenol content and antioxidant activity of the extracts of the two species.	[51]
Coffee plants	Abiotic stress	Drought stress	Chitosan enhanced the content of chlorophyll and carotenoids in the leaves of coffee seedlings, increased mineral uptake and stimulated the growth of coffee seedlings.	[52]
Pepper plants	Abiotic stress	Antitranspirant	Chitosan decreased transpiration due to inducing closure of the plant's stomata by a decrease of K in the guard cells.	[53]
Safflower and Sunflower seedlings	Abiotic stress	Anti-salinity	Chitosan exhibited positive effects on salt stress alleviation through the reduction of MDA and increased proline contents and CAT and POX activities in both crops.	[54]

Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
Wheat seedlings	Abiotic stress	Anti-salinity	Chitosan increased antioxidant enzyme: superoxide dismutase (SOD), catalase (CAT) and peroxidase (POX) activities, reduced malondialdehyde (MDA) content in leaves, and accelerated the accumulation of proline.	[55]
Maize plants	Abiotic stress	Anti-salinity	Chitosan ameliorated the effects of salinity stress and improved plant growth.	[56]
Rice seedlings	Abiotic stress	Anti-salinity	Chitosan lowered malondialdehyde and increased proline levels and enhanced the activities of catalase and peroxidase enzymes.	[57]
Fenugreek (Trigonella foenum- graecum)	Abiotic stress	Anti-salinity	Chitosan increased plant dry weight, length of stem and roots, leaf relative water content, and amount of anthocyanin photosynthetic pigments, resulting in better growth and establishment of fenugreek plants.	[58]
Bean (<i>Phaseolus</i> vulgaris)	Abiotic stress	Anti-heat stress	Chitosan enhanced uptake of mineral nutrients, cell division, and chlorophyll accumulation in the leaves.	[59]
Cucumber seedlings	Abiotic stress	Anti-heat stress	Chitosan improved the cold resistance of cucumber seedlings, protected the membrane system, improved the capability of eliminating active oxygen species, and alleviated the damage to photosynthetic organization.	[60]
Maize plants	Abiotic stress	Anti-heat stress	Chitosan caused a decline in malondialdehyde (MDA) content and relative permeability of the plasma membrane. Also, increased concentrations of soluble sugars, proline, and peroxidase (POD) and catalase (CAT) activities led to an enhanced germination index and reduced the mean germination time (MGT). Additionally, it increased shoot height, root length, and shoot and root dry weights	[61]

Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References	
Bean (Phaseolus vulgaris)	Plant improvement	Promote plant growth	Chitosan enhanced shoot and root length, fresh and dry weights of shoots and roots, and leaf area in beans.	[62]	
Rice and soybean plants	Plant improvement	Promote plant growth	Chitosan promoted plant length and shoot growth in both rice and soybean.	[63]	
Cucumber plants	cumber Plant Promote plant Chitosan increased the unts improvement growth vegetative growth and increased the total yield of cucumber fruits in terms of quantity and quality.				
Okra plants	Plant improvement	Promote plant growth	Chitosan enhanced the biochemical activities in okra, resulting in an increase in growth and the fruit-bearing nodes, and improved the yield quality and quantity.	[65]	
Tomato and eggplant plants	Plant improvement	Promote plant growth	Chitosan enhanced the growth and functional components of tomatoes and eggplants by increasing plant height, branch, leaf, flower, and fruit number and size.	[66]	
Verbena bonariensis	Plant improvement	Promote plant growth	Chitosan enhanced the plant height, number of inflorescences and number of leaves.	[67]	
Eustoma grandflorum	Plant improvement	Promote plant growth	Chitosan soil treatment resulted in greater shoot length, stem diameter, weight of cut flowers, and an increase in the number of flowers.	[68]	
Soybean leaves	In the number of nowers. In leaves Plant Stimulate Chitosan increased activity of improvement plant defense phenylalanine ammonia-lyase responses and tyrosine ammonia-lyase enzymes which increased total phenolic content in plants.		[69]		
Chickpea plants	Plant improvement	Imparts immunity against Fusarium	Chitosan caused significant ECM and guard cell remodelling, and translated ECM cues into cell fate decisions during fusariosis.	[70]	
Several plants	Plant improvement	Eliciting plant immunity	Chitosan-induced defense responses such as increased cytosolic H* and Ca ²⁺ , MAP- kinase activation, callose apposition, oxidative burst, and hypersensitive response (HR) Moreover, it induced the synthesis of abscisic acid, jasmonate, phytoalexins, and pathogenesis-related (PR)	[71]	

Host/disease name	Pathogen/ stress	Chitosan effects	Chitosan mode of action	References
Soybean cells	Plant improvement	Elicited callose synthesis	Chitosan caused the rapid and quantitative formation of callose and elicited other metabolic changes that appeared to be important for resistance against pathogens in soybean cells.	[72]

Table 1.

Chitosan mechanisms in plant protection against biotic and abiotic stresses.

fatty acid-rich membranes such as palmitic acid or stearic acid (low fluidity membrane) [15]. The incubation period affects the antifungal activities of chitosan. The growth inhibition of *F. oxysporum* f. sp. *radicis-lycopersici* at a low concentration of chitosan increases with the long incubation period [74]. Chitosan causes excessive mycelial branching, abnormal shapes, swelling and hyphae size reduction in *F. oxysporum* f. sp. *cubese* [27], *F. solani* f. sp. *glycines* [28], *Botrytis cinerea*, and *A. alternate* [24]. Chitosan is also responsible for cytological alteration, protoplasm dissolution, and large vesicles of fungus [27]. Chitosan caused morphological changes such as large vesicles or empty cells devoid of cytoplasm in the mycelium of *B. cinerea* and *F. oxysporum* f. sp. *albedinis* [30, 75]. Fungi that have been exposed to chitosan make fewer spores than untreated fungi. In other instances, full sporulation suppression was observed following chitosan therapy. The length and shape of the conidia of *Ranunculus stolonifer*, *Penicillium digitatum*, and *F. oxysporum* f. sp. *cucumerinum* showed a significant decrease in wilt disease severity compared to chitosan-untreated Fusarium [31].

3.2 Antibacterial activity

Gram-positive and gram-negative bacteria have significantly different cell wall structures and surface polarity, which results in different sensitivity to chitosan [13]. The cell wall of gram-negative bacteria is distinguished by the presence of lipopoly-saccharides, which contain phosphate and pyrophosphate groups [77]. This provides it with a high negative charge, making it more bound to chitosan [78]. While the cell wall of gram-positive bacteria contains polysaccharides associated with lipoteichoic and teichoic acids. Teichoic acid is negatively charged due to the presence of phosphate groups in its structure which gives it a small negative charge that makes it less bound to chitosan. The stopping of the teichoic acid biosynthesis pathway in *Staphylococcus aureus*, led to an increase in chitosan resistance [35]. Gram-negative bacteria are more hydrophilic than gram-positive bacteria react to chitosan different gram-positive bacteria (**Figure 2**).

Chitosan has potent antibacterial activities against a variety of plant pathogenic bacteria like *S. aureus* [80], *Streptomyces scabies* [36] *Ralstonia solanacearum* [81], *Xanthomonas* spp. [26, 37], *Pseudomonas* spp. [38, 80], and *Acidovorax* spp. [39, 40]. The inhibitory activity of chitosan against bacteria varied with molecular weight [82], concentration [39], solvent type [29], bacterial type (gram-positive/gram-negative) [83], cell wall structure [84], period of incubation and abiotic factors [85]. Chitosan binds to the negatively charged surface of bacteria at low concentrations (less than

0.2 mg/ml) to cause agglutination, but at higher concentrations, the presence of more chitosan positive charges may have given the bacteria a net positive charge that keeps them suspended [86]. Moreover, Goy et al. [87] suggest that chitosan is responsible for the hydrolysis of peptidoglycans, the main component of the bacterial cell wall, increasing electrolyte leakage, and potentially causing the death of the plant pathogens. Additionally, Liang et al. [88] reported that chitosan is the responsible substance for the destruction of the bacterial cell membrane, which causes death due to the leakage of intracellular substances. Chitosan applied to tomato plants inhibited the growth of Xanthomonas vesicatoria [26]. Chitosan-protected cucumber from Pseudomonas syringae pv. lachrymans that causes bacterial angular leaf spot damage [89]. Chitosan has decreased the disease incidence of broccoli that was infected with Pseudomonas fluorescens [38]. The disease index of watermelon seedlings infected with Acidovorax citrulli was significantly reduced by chitosan at 0.4 mg/ml [40]. Chitosan solution at 0.10 mg/ml markedly decreased the surviving cell number of Xanthomonas pathogenic bacteria isolated from different geographical origins compared with the control after 6 h of incubation, regardless of the bacterial strain [90].

3.3 Antiviral activity

One of the most destructive plant diseases is the viral disease which causes serious damage to many plant species, affecting agroecosystems and food security. For that reason, searching for new eco-friendly application technologies to suppress the invasion of viral plant diseases is urgently needed to fulfill the nutrients required to feed the world's population [91]. Chitosan and its derivatives have been used as a promising and powerful tool against plant viruses. Chitosan has demonstrated antiviral activity against Potato virus X (PVX) through the possible mechanism of induced resistance and responsive defense mechanisms or the inhibition of systemic propagation of plant viruses in potato plants [4]. Complete inhibition or suppression of systemic virus multiplication in the host plant has not clearly proven the capability of chitosan to stop the virus activation. However, the multiplication block may be due to the binding of chitosan molecules with the nucleic acid of a targeted virus, causing serious damage to the viral genome [92]. Studies by Jia et al. [41] explained the role of chitosan in inducing systemic acquired resistance in Arabidopsis plants infected with Tobacco mosaic virus (TMV) and which signaling pathways are involved in the processes of defense mechanisms. Their obtained results revealed that chitosan application induced TMV resistance through specific pathways in the plant. The induction in the Arabidopsis plants happened through the jasmonic acid pathway-deficient (Arabidopsis plants jar1) and at the same time did not induce the salicylic acid pathway-deficient (Arabidopsis plants NahG). The application of chitosan as a protective and curative treatment against Alfalfa mosaic virus (AMV) on Nicotiana glutinosa plants under greenhouse conditions was studied by [42]. They proved that the AMV concentration was significantly reduced through both protective and curative treatment with 70.43% and 61.65%, respectively. On the other hand, possible ways of inducing systemic resistance and responsive defense mechanisms were measured, resulting in an increase of total phenols and carbohydrates as well as phenylalanine amonialyase (PAL) and peroxidase.

3.4 Antinematode activity

Serious, highly damaging, and economic losses to a wide range of plant hosts, including fruit trees, vegetables, agronomic crops, foliage crops, grasses, nuts, and

forest trees, were reported to be caused by several nematode species under different genera. Economic losses to more than 2000 kinds of higher plant species are often great when certain plant species are grown in warm regions around the world. The development of efficient new and eco-friendly nematode management strategies such as biological control, natural products, plant extracts, and botanical products is needed urgently to reduce the high toxicity of chemical nematicides [93]. Studies by El-Ansary et al. [43] revealed that chitosan significantly reduced the disease severity of root-knot nematode infection caused by Meloidogyne incognita in banana plants cv. Williams with an improvement in plant growth parameters and yield production. Chitosan-treated tomato plants produced less root-knot nematode reproduction, which enhanced the size, weight, and growth of the plant's roots and shoots [44]. Recently, Khan et al. [45] evaluated chitosan as a nematicide against the infestation caused by Megalaima incognita in carrot plants under in vivo and in vitro conditions. They reported that egg masses and second-stage juvenile (J2s) of *M. incognita* were affected by the usage of different concentrations of 500, 1000, 1500, 2000, and 2500 ppm of chitosan. Maximum mortality of J2s and the highest inhibition in egg hatching was observed at 2500 ppm of chitosan after 36 h incubation period.

4. Role of chitosan in plant protection against abiotic stress

Chitosan improves plant tolerance to drought, salinity, and high temperature [94] (**Table 1**).

4.1 Effect on drought stress

Agricultural productivity is limited by drought stress, which has a number of negative effects on plant health and lowers plant growth and yield. Chitosan is one of the effective solutions to mitigate the harmful effects of drought stress on plants (**Figure 3a**). Under extreme drought stress, the free proline content in leaves considerably increases [95]. In chitosan-treated plants, the accumulation of proline production in the absence of drought stress was enhanced [96]. After chitosan treatment, proline accumulation was enhanced in the thyme plants [47]. The accumulation of proline helps in reducing the leaf water potential, improves the turgor of leaves, and facilitates water delivery to them. In addition, proline is crucial for maintaining redox balance, quenching ROS, and osmotic adjustment [97].

On the other hand, a water deficit condition disturbs the plant's cell membrane integrity. The membrane permeability and malondialdehyde concentration are related to each other, indicating membrane stability. Malondialdehyde levels rise in conditions of water deficiency; this lipid peroxidation byproduct has the potential to lead to membrane leakage because of the accumulation of free radicals. Chitosan functions as a positive regulator in osmotic adjustment and can eliminate the adverse effects of drought stress symptoms by decreasing the production of lipid peroxidation. The pretreatment of thyme and potatoes with chitosan reduced lipid peroxidation, removed ROS, and improved cell membrane integrity [47, 48]. In apple seedling leaves exposed to drought stress, chitosan significantly improved the integrity of cell membranes and decreased the production of malondialdehyde and electrolyte leakage [49]. Plants attempt to reduce the harmful effect of drought by raising the level of soluble sugar in the cell by breaking down the polysaccharides that help in the preservation of turgor [98]. Chitosan is an important source of sugars, e.g., glucose, fructose, trehalose,



Figure 3.

Role of chitosan in removing (a) drought stress and (b) salinity stress.

sorbitol, mannose, and myoinositol that plants need to overcome drought [96]. These might enhance osmotic adjustment and maintain carbon balance in response to dehydration stress, which would improve drought resistance.

Additionally, drought stress hinders photosynthetic activity by reducing chlorophyll production. This could be caused by oxidative damage to chloroplast lipids, pigments, and proteins or by the loss of chlorophyll pigment complexes or lightharvesting protein complexes [50, 99]. Chitosan spraying resulted in an increase in chlorophyll and total carbohydrates which increased photosynthesis levels in soybean and maize [100], cowpea [50], and bean [62]. This might be the result of higher nitrogen and potassium levels in plant shoots, which aid in raising the number of chloroplasts per cell and boosting chlorophyll synthesis. Additionally, chitosan treatment's release of amino compounds with a higher availability level encourages the synthesis of chlorophyll [63].

Commercial antiperspirants are better than chitosan in raising the efficiency of the plant in retaining water, but they reduce the photosynthetic rates and carbon uptake as a result of reducing the internal CO_2 in leaves [71]. The efficiency of chitosan as an antiperspirant is due to its control of the mechanism of opening and closing the stomata, which allows the entry of carbon from the atmosphere into the plant. Thus, maintaining the efficiency of the photosynthesis process inside the plant, unlike the commercial antitranspirant, this acts as a thin antitranspirant membrane that covers the leaves' surface and blocks stomata, which prevents the entry of carbon needed for photosynthesis. Chitosan use as an antitranspirant substance would be more suitable for plants that experience occasional drought occurrences. Chitosan-treated plants

would enable their natural physiological system to quickly recover maximum carbon uptake while sustaining biomass and yield in these circumstances [53]. Histological examination of chickpea leaves revealed stomatal closure accompanied by decreases in stomatal conductance and transpiration rate in chitosan-treated seedlings during Fusarium infection, indicating the presence of stomatal immunity associated with chitosan [70].

4.2 Effect on salinity stress

Salinity can prevent plants from absorbing water and nutrients due to low external osmotic potential. In addition, the direct ionic effect results in excessive accumulation of Na and Cl ions, which causes toxic effects, closes the stomata, lowers internal CO_2 . and decreases the rate of photosynthesis. Salt-induced stress conditions were discovered to have lipid peroxidation brought on by the buildup of malondialdehyde [101]. A decrease in malondialdehyde content after chitosan treatment stabilized membrane damage and may have given salt stress tolerance [54]. Plants have developed their own inherent ROS scavenging mechanism by producing antioxidant enzymatic compounds, e.g., superoxide dismutase, peroxidase, and catalase; increased abundance of these enzymes denotes effective ROS detoxification (Figure 3b). These enzymes were shown to be elevated in chitosan-treated plants, and they are crucial for reducing salt stress since they are stronger antioxidant enzymes [54, 55]. Chitosan has the ability to scavenge superoxide anions due to the presence of hydroxyl and amino groups that react with ROS [102]. Chitosan treatment reduces malondialdehyde levels and increases antioxidant enzyme activities during salinity stress, which minimizes the negative effects of salt stress on maize [56]. The low concentration of chitosan treatment could counteract the harmful effects of salt stress. A small amount of chitosan applied to sunflower seeds can suppress enzyme activity and decrease the oxidative damage brought on by salt stress [54]. During salt stress, wheat seed treated with chitosan showed increased levels of the antioxidant enzymes (superoxide dismutase, peroxidase, and catalase), stomatal conductance, and photosynthetic rate [55]. When a plant is exposed to salt stress, its chlorophyll concentration drastically decreases because of the accumulation of chlorophyllase and the instability of protein complexes. In salt stress, proline levels were elevated, which might be the result of increased proline biosynthesis, decreased proline oxidation to glutamate, or decreased utilization of synthesized proline. Plants are mostly protected against osmotic stress by proline [103]. Proline levels were shown to increase as chitosan concentrations increased [54].

4.3 Effect on heat stress

There is a paucity of published data on the use of chitosan in heat stress. Heat stress is frequently viewed as a complex issue because it frequently occurs in conjunction with drought stress and is challenging to measure [57]. Foliar application of chitosan combined with humic acid and zinc is the best treatment that could be recommended for dry bean production to withstand heat stress [59]. Chitosan use could reduce the effects of high temperatures by promoting abscisic acid activity [53]. According to Choi et al. [104], abscisic acid can activate genes associated with heat shock, such as ABF3, whose overexpression may improve heat stress tolerance. Applying chitosan to cucumber leaves at low temperatures can improve their proline and soluble protein levels as well as the activity of antioxidant enzymes [60].

5. The role of chitosan in enhancing plant traits

When a pathogen attacks a plant, a coordinated signaling mechanism is induced which leads to the accumulation of several gene products. Once the pathogen gets recognized by plant receptors, rapid localized cell death will be developed, which is known as the hypersensitive response, causing necrosis at the site of infection. While in the sections of the plants that are not infected, a systemic expression of a broad spectrum of resistance will be induced to prevent additional pathogen infection. Then reactive oxygen species are produced, defense-related genes are activated, and the expression of genes that produce compounds, including terpenes, phytoalexins, defense enzymes, and pathogenesis-related proteins, is increased.

Chitosan can induce plant resistance and activate several defense processes in plant tissue [32]. These defense mechanisms include accumulation of hydrolytic enzymes, synthesis of proteinase inhibitors and pathogenesis-related (PR) proteins, enhancement of phytoalexins, formation of callose, promotion of lignification, and induction of reactive oxygen species (ROS) (**Figure 4**) [46, 105].

5.1 Plant growth promotion

Chitosan promotes plant growth in a variety of crops by significantly influencing the development rates of shoots, roots, flowering, and the number of flowers. Due to the great hydrophilicity of chitosan molecules, it adjusts the osmotic pressure in plant cells by increasing the absorption of water and important nutrients [61] and minimizing stress damage in plant cells. Chitosan increases the efficiency of plant nitrogen uptake, acts as a source of energy and an additional carbon source for the synthesis of carbo-hydrates, and acts as an activator for various metabolic processes [14, 51]. Chitosan encourages the proliferation of root cells by activating auxin and cytokinin, which further boosts nutrient uptake [52]. Chitosan assisted in triggering the hydrolytic enzymes required for the mobilization and degradation of reserve food components, including protein and starch [106]. Significant growth enhancements have been recorded after chitosan application by several studies in artichoke [107], cucumber [64], okra, [65], eggplant and tomato [66], strawberry [108], potato [109], chili [33], and watermelon [110]. Chitosan can enhance plant physiological mechanisms, e.g., nutrient absorption, cell elongation, cell division, enzymatic activation, and protein synthesis [111].

5.2 Increase photosynthetic activity

Chitosan improves the photosynthesis process by enhancing stomatal function and reducing the breakdown of chlorophyll [112]. Chitosan increased the chlorophyll levels in leaves by 13.4% compared to control plants [67]. The application of chitosan improved chlorophyll content and plant productivity in chili [113], peanut, and coffee plants [52]. The use of chitosan protects chlorophyll content in stressful circumstances; it increases the chlorophyll content in fenugreek under a salinity conditions [58]. Applying chitosan to cucumber leaves at low temperatures can improve their proline and soluble protein levels as well as the activity of antioxidant enzymes [60].

5.3 Up-regulate pathogenesis-related proteins

Plants produce proteins known as pathogenesis-related proteins (PR) in response to a pathogen invasion. They are induced as a part of systemic acquired resistance



Figure 4.

Mechanisms of chitosan in the enhancement of plant defense system and inhibition of phytopathogens.

where their corresponding genes are activated by infections. Chitosan has been characterized as an elicitor that causes plants to create a wide variety of pathogenesis-related proteins with antimicrobial action. Among these pathogenesis-related proteins are chitinase and 1,3-glucanase, two hydrolytic enzymes that destroy pathogen cell walls that contain chitin and/or β -D-glucans as major structural components [114].

Chitosan appeared to use a variety of mechanisms to enhance pathogenesis-related gene function, including activating membrane receptors and altering the DNA

structure of the plant. Chitosan of low molecular weight was more effective at inducing the defense-related genes β -1,3-glucanase and chitinase than the higher molecular weight [115]. Chitosan triggered the transcriptional up-regulation of defense-related genes β -1,3-glucanase and chitinase in rice seedlings [115]. Chitosan significantly increased the expression of general defense response genes in oat leaves [116]. Chitosan was able to promote resistance in pears and peaches by increasing chitinase and β -1,3-glucanase activities [74, 117]. Chitosan has been proven to be effective at triggering plant defense mechanisms by increasing the level of β -1,3 glucanase and chitinase enzymes in strawberry and pepper plants [118].

5.4 Stimulate defense-related enzymes

Chitosan acts as a physiologic elicitor, enhancing defense-related enzymes, e.g., peroxidase, catalase, superoxide dismutase, polyphenol oxidase, and phenylalanine ammonia-lyase [119]. Peroxidase helps oxidize phenolic and enodiolic compounds into quinones and hydrogen peroxide [120]. Peroxidase increases pathogen resistance in plants [121]. The chitosan treatment markedly boosted the peroxidase activity in the flesh surrounding the pear fruit wound [74]. Chitosan induced peroxidase expression activity in date palm roots when injected at three concentrations (0.1, 0.5, and 1 mg/ml) [30]. Fruit treated with chitosan maintained relatively higher peroxidase gene expression than control fruit [117]. Catalase is crucial for plant senescence and defense [122], it transforms H₂O₂ into H₂O and O₂. Catalase activity was increased in the chilling-sensitive and chilling-tolerant maize seedlings after chitosan treatment [61]. Superoxide dismutase destroys radicals and protects cells against the effects of oxidative stress. Superoxide dismutase activity has increased after chitosan treatment of Hydrilla verticillata [123]. Drought stress decreased the activities of the antioxidant enzymes catalase and superoxide dismutase in apple leaf tissues, but chitosan treatment enhanced their activities [49].

Polyphenol oxidase participates in plant defense by encouraging the production of lignin, which strengthens the cell wall structure and deters disease penetration [34]. Chitosan has significant antibacterial efficacy against rice leaf streak and leaf blight produced by *X. oryzae* pv. *oryzicola* and *X. oryzae* pv. *oryzae*, respectively, by increasing polyphenol oxidase activity [37]. Phenylalanine ammonia-lyase transforms L-phenylalanine to ammonia and trans-cinnamic acid [124]. It is induced in host tissues as a result of pathogen infection [69]. In grape berries, rice, and wheat, elicitation with chitosan resulted in an increase in phenylalanine ammonia-lyase [37]. When chitosan was injected into the roots of date palm, it enhanced the essential components of the host resistance against *F. oxysporum* f. sp. *albedinis*, increased the level of phenolic compounds, and stimulated date palm peroxidase and polyphenol oxidase activities [30].

5.5 Induce signal regulation

5.5.1 Activation of signal transduction

Chitosan can trigger the plant's defense mechanisms and functions as a regulatory molecule in signal transduction through several signaling pathways. When chitosan activates a particular receptor located on the cell membrane or intracellular, one or more second messengers relays the signal to the cell. This triggers a range of physiological responses as a single signal can be amplified and develop a complex signaling network. This is called signal transduction. The chitosan-mediated signal pathway includes reactive oxygen species (ROS), Ca²⁺, nitric oxide (NO), salicylic acid (SA), abscisic acid (ABA), jasmonic acid (JA), and ethylene (ET) [125].

One of the first reactions to a microbial pathogen attack is the oxidative burst, which has been demonstrated to occur upon chitosan elicitation. It is characterized by the quick and temporary creation of enormous levels of ROS (hydrogen peroxide (H_2O_2), hydroxyl radicals (OH⁻), singlet oxygen ($^{1}O_2$), and superoxide (O⁻²) (**Figure 5**) [126]. Chitosan induced the accumulation of H_2O_2 in tomatoes [127]. Upon pathogen infection, one of the quickest responses is an increase in cytosolic Ca²⁺ [128]. Within 20 min of being treated with chitosan, Glycine max suspension-cultured cells began to synthesize callose. However, in the absence of exogenous Ca²⁺, chitosan-induced callose production was not achievable [72]. Chitosan increased the amount of free cytosolic Ca²⁺ in Arabidopsis and stomatal closure [129].

NO, another messenger, implicated in the plant defense response against pathogens and involved in chitosan-induced resistance [114]. Downy mildew-infected pearl millet seedlings treated with chitosan exhibited increased NO buildup commencing 2 h after inoculation, as well as protection from downy mildew [114]. Chitosan induced the generation of NO and phosphatidic acid in tomato cell culture while the phospholipase-mediated signaling pathway was inhibited after using NO scavenger, indicating that the production of phosphatidic acid during the plant defensive response needed NO [130].

Plant hormones regulate growth processes in plants and play important roles in plant responses to biotic and abiotic stresses [131]. Salicylic acid, jasmonic acid, and ethylene play a crucial role as signaling molecules in modulating Arabidopsis' responses to biotic and abiotic stress. Jasmonic acid and ethylene are central signaling molecules



Figure 5. An overview of how chitosan helps plants withstand phytopathogens, abiotic stress, and ROS.

in the induced systemic resistance; on the other hand, salicylic acid is involved in systemic acquired resistance, which occurs when a pathogenic attack on one area of the plant results in resistance in other parts. It has been suggested that jasmonic acid is a component of a signal transduction pathway that controls the activation of genes involved in plant defensive responses to pathogen invasion. Chitosan significantly increased the jasmonic acid in wounded rice leaves [132]. ABA regulates the intensity and speed of callose deposition. Additionally, ABA-mediated signaling transduction is crucial for plants to respond to abiotic and biotic stresses [133]. Chitosan activated the defense signaling pathways in tomato plants against *A. solani* and *X. vesicatoria* [26].

5.5.2 Activation of symbiotic signaling

Beneficial microorganisms are able to form a symbiotic connection while living in close proximity to their plant hosts, which aids in the acquisition of nutrients necessary for plant growth. Additionally, the plant meets some of the requirements that helpful microorganisms need to complete their metabolic processes for growth and reproduction [134]. Chitosan triggered symbiotic signaling between plants and beneficial microbes [135]. Lipo-chitooligosaccharides, which are chitin oligosaccharides bound to a lipid moiety, have been discovered to be released by mycorrhizal fungi and rhizobium bacteria during the development of symbiotic interactions and have been recognized as key signaling molecules triggering the plant symbiotic response [136]. Chitosan undergoes enzymatic degradation without harming the beneficial rhizosphere biota in the soil and promotes symbiotic interactions between plants and microorganisms, causing changes in the rhizosphere's microbial balance and harming plant diseases [44].

5.6 Usage as bio-fertilizer

As a result of the repeated use of inorganic fertilizers that are difficult to decompose, the toxicity of the soil increased, which affected the beneficial microorganisms present in the soil and the properties of the soil. Therefore, the use of chitosan at low concentrations as a bio-fertilizer was a safe and effective alternative to avoid the risks of using inorganic fertilizers. Utilizing chitosan as a biofertilizer showed a significant decrease in late blight infestation of potato tubers and an increase in plant nutrient uptake [137]. The addition of chitosan to soil improved the phosphorous and nitrogen content in *Eustoma grandiflorum* [68].

Large amounts of inorganic fertilizers are lost in the soil due to the inability of plants to absorb them, causing farmers to over-apply them, resulting in their presence in the soil at higher than required rates, causing soil toxicity, water pollution, and damage to crops, particularly vegetables, which are severely affected by fertilizer toxicity [138]. However, when chitosan coating on fertilizer is added to the soil, it improves the absorption of inorganic fertilizer by plants, which minimizes the use of fertilizers, makes the soil less toxic, and reduces the production cost [139].

6. Recovery of contaminated agricultural wastewater and soil

The concentration of heavy metals and other contaminants in the environment is increasing rapidly as a result of multiple human activities. Heavy metals are dangerous because they are highly poisonous, do not biodegrade, and cause cancer and other disorders. Therefore, it is necessary to find an efficient way to remove them from the environment and dispose of them. From this perspective, bio adsorption is acknowledged as an affordable and effective solution [140]. Effective pollutant removal makes it possible to reuse valuable resources like cultivable soil and fresh water. The application of chitosan-based adsorbents in these areas has been extensively studied due to their low production costs, biocompatible and biodegradable nature, strong resistance to antimicrobial attack, and absence of the creation of potentially toxic secondary end products [140]. The chemical composition of chitosan makes it simple to combine with certain ions, molecules, and other compounds to produce complex structures for specific applications. Carboxylated graphene oxide-chitosan (GO-COOH/CS) spheres were used for the immobilization of Cu²⁺ from water and soil, as well as for reducing the bioaccumulation of Cu²⁺ from wheat plants [141]. Cesium-contaminated clay can be cleaned using ionized chitosan and magnetic microgels functionalized with Prussian blue analogs, 200 mg/g of ionized chitosan hydrochloride can achieve 87.6% cesium release from clay in about 2 h [142].

A combination of chitosan and duckweed was assessed for its ability to remove boron from water. Chitosan beads had the maximum boron absorption capacity of chitosan at 3.18 mg/g [143]. Chitosan can help reduce the environmental effects of industrial wastewater treatments and soil acidity by reducing CO₂ and SO₂ [144]. Columns packed with chitosan have the ability to remove arsenic from groundwater [145]. A composite consisting of chitosan and hyacinth extract effectively absorbs Cu, Pb, and Cd ions from water [146]. Cd removal from the soil and aquatic environments was examined using the magnesium oxide biochar-chitosan composite (MgO-BCR-W) [147].

The adsorbent made of chitosan/MnO₂ nanocomposite was employed to extract Cr (VI) from the aqueous solutions [148]. A magnetic chitosan/polyacrylic acid nanocomposite successfully adsorbed Pb (II) from an aqueous solution [149]. Iron chitosan microspheres were synthesized by ionotropic gelation for the removal of arsenic (V) from water [150]. Chitosan contains functional amino and hydroxyl groups, which enables it to form compounds with heavy metals. Foliar application of chitosan can reduce the harmful effects of cadmium in *Brassica rapa* L. [151]. Herbicides can be made more effective by adding chitosan to formulations, which lowers the amount utilized and the risk of hazard accumulation in the environment. A formulation consisting of chitosan and glyphosate exhibits lower phytotoxicity and higher herbicidal efficacy and releases active substances better than using glyphosate alone [152]. Chitosan and tripolyphosphate nanoparticles are efficient carriers to reduce soil sorption, cytotoxicity, and mutagenicity of paraquat and enhance their herbicide activity [153].

7. Chitosan future prospect

Despite extensive research on chitosan, the mechanism of action of chitosan in regulating plant immunity and suppressing the pathogen has not been sufficiently elucidated. It is thought that the mode of action of chitosan may be more complex and involve a series of overlapping details that need to be further studied in the future. Proteomics is one of the contemporary disciplines that has been successfully used to investigate the global variations in protein expression in biological organisms under a variety of environmental conditions [154]. There are a lot of proteomic studies that explain the chitosan mode of action in withstanding biotic

and abiotic stresses in plants. The inhibitory effect of chitosan on *P. expansum* was proteomic analyzed, and 26 proteins were identified and grouped according to their potential biological roles [155]. A comprehensive proteomic study of chitosan-responsive proteins explained the inhibitory mechanism of chitosan against *F. oxysporum* f. sp. *cucumerinum*. This led to the identification of 62 expressed proteins involved in the hindering of the Fusarium cell wall, disrupting DNA, and disrupting structural and functional protein biosynthesis and explained how chitosan influences metabolic pathways [31]. We wish plants and pathogens treated with chitosan would receive abundant proteomic studies in order to make maximum use of chitosan in sustainable agriculture.

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Conflict of interest

The authors declare no conflict of interest.

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Chapter 6

Chitinase from Basal Trypanosomatids and Its Relation to Marine Environment: New Insights on *Leishmania* Genus Evolutionary Theories

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Abstract

Leishmaniasis, an infectious disease that affects humans, domestic dogs, and wild animals, is caused by 20 of the 53 Leishmania genus species and is transmitted by sandflies. Despite its significant impact, the disease is often neglected. Leishmania genus, belong to Trypanosomatide Family and Kinetoplastida Order, are grouped in five subgroups according to biogeographic and evolution history of parasites and hosts. The GH18 Leishmania chitinase is encoded by a specie-specific single copy gene, conserved in basal groups of trypanosomatids, and is absent in the genus Trypanosoma. Preservation of the chitinase genomic locus in the aquatic free-living protozoan Bodo saltans, discloses a primitive common origin. Trypanosomatid chitinase amino acid sequence comparative analysis revealed high similarity with chitinase from sea living prokaryotes and protozoan microorganisms, indicating a probable marine origin. Amino acid sequence comparative analysis revealed that perhaps the trypanosomatid chitinase derived from a water living Kinetoplastida ancestor and its phylogenetic reconstruction corroborates the Supercontinent Origins theory for Leishmania. The chitinase-encoding gene was effective for differential molecular diagnosis among Leishmania clinical important species worldwide.

Keywords: leishmaniasis, molecular diagnosis, molecular evolution, chitinase, marine environment

1. Introduction

Leishmania genus protozoan parasites are the causative agent of leishmaniases, a complex of diseases that affect the tegument or the viscera. Leishmania parasites are transmitted among humans, domestic dogs, and wild animal hosts by insect

vectors of the Psycodidae Family (sandflies) as well as the *Phlebotomus* (Old World) and *Lutzomyia* genus (New World - the Americas) [1]. Up to now 53 *Leishmania* species were described which are divided into five groups, the subgenera *Leishmania, Viannia, Sauroleishmania, Mundini,* and *Paraleishmania*. Most of the *Leishmania* species are zoonotic and 20 are incriminated to cause disease in human [1].

The diseases caused by *Leishmania* parasites, named leishmaniases, can be divided into Tegumentar (TL) and Visceral (VL), depending on the species of infecting parasite and host immunity conditions. Leishmaniasis can range from mild tegumentar ulcerations to fatal visceral infection. Leishmania parasites are endemic in 98 countries distributed in all continents and its prevalence is estimated as 0.4 and 1.2 million cases of VL and TL, respectively [2]. In the Americas, Brazil accounts for the highest incidence of leishmaniasis with wide spreading of TL and VL in expansion [3].

The VL is in second place as to the highest impact on health population, just behind malaria, in India and in the Mediterranean countries being caused by *L. (Leishmania) donovani* and *L. (Leishmania) infantum*, respectively [4]. In the Americas VL is caused by *L. infantum*, where this species of parasite is not endemic and probably it entered the Americas by infected dogs brought by Mediterranean colonizers [5]. Seventy-five percent of TL new cases occur in Afghanistan, Algeria, Colombia, Brazil, Iran, Syria, Ethiopia, Sudan, Costa Rica, and Peru, causing morbidity and disfiguration in infected people [6]. In South America, TL is mainly caused by the most prevalent endemic *Leishmania* species, viz., *L. (Viannia) braziliensis, L. (Leishmania) amazonensis,* and *L. (Leishmania) mexicana*. In spite of much efforts, a precise diagnostic test and effective treatment for leishmaniasis are still unavailable [7]. Thus, a detailed understanding of all aspects of specific biology and host-parasite relationships are important prior to facilitating the formulation of innovative and effective drugs and diagnostic tests for developing adequate prevention and control strategies.

The Leishmania parasites to complete their life cycle, they must invade the digestive tract of sand-fly species, which requires the degradation of insect chitin by a specific parasite chitinase [8]. Chitinase catalyzes the β -1,4-glycoside bond hydrolysis reaction of N-acetylglucosamine of chitin and chitodextrins [9]. Amino acid sequence similarity analysis of Leishmania GH18 chitinase grouped these enzymes in the GH18 and GH19 glycosyl hydrolase families. Initial studies in *Leishmania* revealed that chitinase and N-acetylglucosaminidase activities were found in promastigote supernatant cultures of *L. (Leishmania) major*. Similar activity of both enzymes was observed in *L. donovani*, *L. infantum*, *L. braziliensis*, *Leptomonas seymouri*, *Crithidia fasciculate*, and *Trypanosoma lewisi*. The chitinolytic action was attributed to the parasite secretion and was not secreted through the sand fly gut [10, 11]. The gene encoding a GH18 chitinase was initially obtained for *L. donovani* (Ld Cht1) using molecular approach and biochemical characterization. Molecular genetic studies enabled the identification of a similar gene in several species of *Leishmania* genus (*L. major*, *L. infantum*, *L. donovani*, and *L. braziliensis*) [12].

The biological importance of the Leishmania GH18 chitinase in parasite life cycle was confirmed after homologous episomal overexpression of chitinase in both amastigotes and promastigotes of *L. Mexicana*. In the insect vector, overexpression of Leishmania chitinase resulted in an increase in transmission rate where in the vertebrate host ensued increased pathogenicity, thereby indicating that chitinase plays an important role in parasite development, survival, and transmission in mammalian hosts [13, 14]. Due to the biological properties and characteristics of *Leishmania* genus GH18 chitinase, including *locus* conservation, species-specific amino acid and

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nucleotide sequence, expression in all species of parasite developmental stages, and extracellular exportation, this investigation focused on the potential of the chitinaseencoding gene as a molecular diagnostic tool and a phylogenetic marker for studying basal trypanosomatids groups.

2. Chitinase genetic *locus* conservation among Kinetoplastida revealed its marine origin

Evaluation of the phylogenetic relationship of the GH18 family chitinase in Kinetoplastida was performed through comparative analysis by Basic Local Alignment Search Tool (BLAST) on chitinase amino acid and nucleotide sequences of trypanosomatids, available in public databanks (**Figure 1**) [15]. The results showed that the chitinase amino acid sequence is highly conserved among the species of the *Leishmania* genus, with identity variations ranging from 78 to 100%. Additionally, basal trypanosomatids such as *Leptomonas*, *Strigomonas*, and *Angomonas* also harbor a GH18 chitinase that is similar to that of *Leishmania*, with identity percentages of 60, 40, and 35%, respectively. This chitinase was found to be absent in parasites from *Trypanosoma* genus. A protein with identity of 32% with *Leishmania* genus chitinase was also present in the free-living acquatic protozoa from Kinetoplastida Order and Bodonidae Family, *Bodo saltans*, which is used as external group of Trypanosomatid in phylogenetic studies. These results suggest the occurrence of a homologous GH18 chitinase in a Kinetoplastida ancestor.

The TritrypDB genomic resource tools were utilized to analyze the locus of the chitinase-encoding gene in all available Kinetoplastida sequences, comprising *Bodo saltans*, *Leptomonas*, *Angomonas*, *Strigomonas*, and *Leishmania*. The results indicated a high degree of conservation, which further supports the hypothesis of a shared origin. In all organisms included in the analysis, the GH18 chitinase is a single copy gene (**Figure 1**).

BLAST analysis of the chitinase amino acid sequence from *Bodo saltans* showed identity of 38% with the chitinase of the marine microorganisms *Perkinsus marinus* and *Micromonas pusilla*, indicating that the GH18 chitinase of the trypanosomatids ancestor emerged from marine environment. These data are supported by the phylogenetic reconstruction of Kinetoplastida GH18 chitinase which grouped in a common cluster, separated from the GH18 chitinases of human and insects (**Figure 2**).

3. The phylogenetic relationship among basal trypanosomatids chitinase corroborated the supercontinent origin hypothesis for *Leishmania* genus

The barcode for phylogenic relationship among trypanosomatids corresponds to the variable region V4 (described also as V7 and V8) of the small ribosomal subunit (V4 rRNA SSU) [5, 16]. In order to investigate the potential of chitinase-encoding gene for trypanosomatids phylogenetic reconstruction, a partial 953 bp chitinaseencoding gene fragment and the corresponding amino acid sequence from trypanosomatids available in genomic databanks and generated by our research group was evaluated for phylogenetic reconstruction by *Neighbor-Joining* [17] method, using the MEGA4 software (**Figure 3**) (**Table 1**). The obtained trees were compared to the phylogenetic analysis performed by *Neighbor-Joining* method, with the trypanosomatid barcode of the same species (**Figure 4**). The phylogenetic reconstruction of *Leishmania* based on GH18 chitinase-encoding gene corroborated the thesis of the

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Figure 1.

Leishmania species CH18 chitinase locus obtained from available sequences in the Tritrypdb data base, a trypanosomatide genomic bank, using the software for genomic analysis (https://tritrypdb.org/a/jbrowse.jsp?loc=LinJ.16:276441..307814&data=/a/service/jbrowse/tracks/linfJPCM5&tracks=gene%2CSyntenic%20Sequences%20 and%20Genes%20 (Shaded%20by%20Orthology).

Supercontinent Origin for *Leishmania* genus [18] with higher accuracy, according to bootstrap results, when compared to the trypanosomatid barcode. The rRNA SSU V4 groups the subgenera *Mundinia* together with the subgenera *Viannia* (**Figure 4**), while the phylogenetic relationships based on the chitinase protein separate the species of all *Leishmania* subgenera and *Paraleishmania* group.

4. Conventional polymerase chain reaction (PCR) associated to restriction length polymorphism differentiated *Leishmania* subgenera of old and New World

In silico analysis of the Genbank *Leishmania* species chitinase, which belongs to the glycosil-hydrolase 18 (GH18) family, revealed that it is localized on chromosome 16 and encoded by a single copy gene. The analysis also indicated high inter-subgenera identity in all crucial putative domains and post-translational modifications [21, 22].

Detection and identification of *Leishmania* parasite subgenera was successfully obtained by employment of *Pst I* restriction analysis on the *Leishmania* species chitinase gene 953 bp fragment. Using *Dde I* restriction analysis was possible to separate viscerotropic and tegumentar species from Old World *Leishmania* subgenus and *Sauroleishmania* and *Viannia* subgenera (**Table 2**). Chitinase from Basal Trypanosomatids and Its Relation to Marine Environment: New Insights... DOI: http://dx.doi.org/10.5772/intechopen.111471



Figure 2.

Evolutionary relationships of GH18 chitinase amino acid complete sequence by neighbor-joining method. Phylogenetic reconstruction was conducted in MEGA4 and the figure corresponds to the optimal tree with the sum of branch length = 6.42769400. The bootstrap values (1000 replicates) are shown next to the branches. Evolutionary distances used to infer the phylogenetic tree are drawn to scale.

5. Discussion

Several regions, such as Brazil, are endemic to more than one species of Leishmania, with both VL (*L. infantum*) and TL (*L. braziliensis, L. amazonensis*). Additionally, there are parasites from the *Trypanosoma* genera (*T. cruzi, T. rangeli*) that show cross-reaction in various serological and molecular tests. Thus, the differential molecular diagnosis of *Leishmania* based on chitinase-encoding gene is highly sensitive and specific and becomes relevant in these multiple species of *Leishmania* and Trypanosoma endemic areas. Therefore, a diagnostic method capable of distinguishing between different *Leishmania* species in animal, human, and vector reservoirs will better guide leishmaniasis control.

Nucleic acid detection techniques in samples from people and/or animals infected with *Leishmania*, such as PCR, are used for detection and identification of the parasite since the 1980s. PCR includes the amplification of various fragments, including those of the gene that encodes the small ribosomal RNA subunit (SSU rDNA) [23], the transcribed internal ribosomal DNA spacer (ITS) [24], sequences that correspond to kinetoplast (kDNA) [25], and mini-exon [26], as well as the gene encoding the heat shock protein HSP70 [27], etc. Despite its high sensitivity and, depending on the molecular target, high specificity, PCR is more commonly utilized in epidemiological studies rather than as a routine diagnostic method. In addition, to achieve



Figure 3.

Evolutionary relationships of Leishmania genus and ancient trypanosomatides by neighbor-joining method using the 292 amino acid residues corresponding to the 949 bp chitinase-encoding gene fragment. Phylogenetic reconstruction was conducted in MEGA4 and the figure corresponds to the optimal tree with the sum of branch length = 4.73393897. The bootstrap values (1000 replicates) are shown next to the branches. Evolutionary distances used to infer the phylogenetic tree are drawn to scale. Leishmania subgenus and Paraleishmania parasites are described at right.

high sensitivity in the methodologies evaluated so far, PCR complementation with other techniques including nested PCR and hybridization is required. To identify *Leishmania* species, various methodologies are employed, including the analysis of restriction fragment sizes of PCR products. However, as most gene targets have multiple copies, interpretation of the results can increase the difficulty of using these techniques in clinical routine. In addition, false positives are possible due to contamination with other post-PCR amplified samples or DNA fragments and cross-reaction with other pathogens, including *Trypanosoma* [28].

The differential diagnosis of *Leishmania* subgenera based on chitinaseencoding gene offers certain advantages over other molecular methods. This is because it is encoded by a single copy gene, which is absent in the *Trypanosoma* genus, thereby allowing for specific detection of *Leishmania* parasites. Also, the sensitivity of the method, regarding the size of the amplified fragment, is high, supporting post-PCR analysis of a single reaction obtained directly from
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Species	Gene	Source	Observations
Angomonas deanei	Chitinase	ENA ¹ :EPY25377	
A. deanei	SSU	Genbank ² :HM593012	Isolate TCC080E
Blechomonas ayalai	Chitinase	Tritrypdb ³ :rna_Baya_138_0020-1	B08-376
B. ayalai	SSU	Genbank KF054116	Isolate B08-376
Bodo saltans	Chitinase	Genbank CYKH01001162	Sequência do <i>locus</i>
B. saltans	SSU	Genbank MF962814	Strain NG
Crithidia luciliae	SSU	Genbank KY364901	Isolate ATCC14765
Critidia fasciculate	Chitinase	Tritrypdb::CFAC1_120016000	Strain Cf-Cl
Endotrypanum monterogeii	Chitinase	Tritrypdb:EMOLV88_160012400.1	Strain LV88
E. monterogeii	SSU	Genbank X53911	
Fimbriiglobus ruber	Chitinase	Genbank OWK46432.1 Bacterium from Planctomycetale associated to Cu	
Homo sapiens	Chitinase	Genbank AAG10644.1	Chitotriosidase, Macrofago, CH1; GH18.
L. amazonensis	Chitinase	Genbank: MG869127	Strain IOCL 0575
L. amazonensis.	SSU	Genbank JX030083	Isolate LaPH8
L. braziliensis	Chitinase	Genbank LS997615	MHOM/BR/75/M2904
L. braziliensis	SSU	Genbank JX030135	Isolate LbrET
L. donovani	Chitinase	Genbank CP019523	Strain MHOM/ IN/1983/AG83
L. donovani	SSU	Genbank X07773	
L. enrietti	Chitinase	Tritrypdb:LENStrain LEM 3045LEM3045_160013500.1	
L. enrietti	SSU	Genbank KF041798	Isolate CBT47
L. gerbilli	Chitinase	Tritrypdb:LGELEM452_160013100.1 Strain LEM452	
L. guyanensis	Chitinase	ENA:CCM15041	
L. guyanensis	SSU	Genbank X53913	
L. gymnodactyli	SSU	Genbank KX790780 TCC722	
L. hoogstraali	SSU	Genbank KF041810	Isolate L7
L. infantum	Chitinase	Genbank:FR796448	Strain JPCM5
L. infantum chagasi	SSU	Genbank KT240047	Isolate CBT 153
L. lainsoni	SSU	Genbank KF041805	Isolate IOCL 1023
L. major	Chitinase	Genbank:FR796412.1 Strain Friedlin	
L. major	SSU	Genbank XR_002460813	
L. Mexicana	Chitinase	Genbank:AY572789	
L. Mexicana	SSU	Genbank GQ332360 Strain Friedlin	
L. naiffi	SSU	Genbank KF041807 Isolate IOCL 1365	

Species	ecies Gene Source		Observations	
L. panamensis	Chitinase	Genbank CP009385	Strain MHOM/PA/94/ PSC-1	
L. panamensis	SSU	Genbank JN003595	Strain UA946	
L. peruviana	Chitinase	Genbank LN609244	PAB 4377	
L. shawi	SSU	Genbank KF041808	Isolate IOCL 1545	
L. tarantolae	Chitinase	Tritrypdb:LtaP16.0770	Parrot-TarII	
L. tarentolae	SSU	Genbank X53916		
L. tropica	Chitinase	TritrypdbLTRL590_160013800.1	Strain L590	
L. turanica	Chitinase	Tritrypdb:LTULEM423_160013000.1	Strain LEM 423	
Leishmania sp	SSU	Genbank AF303938	MHOM/MQ/92/MAR1	
Leptomonas pyrrhocoris	Chitinase	Tritrypdb:rna_LpyrH10_15_0870	H10	
L. pyrrhocoris	SSU	Genbank JN036653	Isolate H10	
Leptomonas seymori	Chitinase	Tritrypdb:PCLsey_0068_0030	ATCC 30220	
L. seymouri	SSU	Genbank KP717894	Isolate Ld_39	
Lutzomyia longipalpis	Chitinase	Genbank AAN71763.1	GH18 chitinase	
M. pusilla	Chitinase	Genbank XP_003063458.1	Marine photosynthetic eukaryotic microorganism.	
Paratrypanosoma confusum	Chitinase	Tritrypdb:PCON_0062580	Isolate cul13	
P. confusum	SSU	Genbank KF963538	Isolate cul13	
P. marinus	Chitinase	Genbank XM_002788039.1	Eukaryotic microorganism, pathogen of oysters.	
Phlebotomus papatasi	Chitinase	Genbank AAV49322.1	GH18 chitinase	
Planktomyces sp	Chitinase	Genbank OAI56776.1	Bacterium from marine environment.	
Sergeia podlipaevi	SSU	Genbank DQ394362	Strain CER3	
Strigomonas culicis	Chitinase	ENA:EPY22137	522 amino acids	
S. culicis	Chitinase	ENA:EPY29957	311 amino acids	
S. culicis	SSU	Genbank HM593009	Isolate TCC012E	

¹ENA: European Nucleotide Archive (Available from: http://www.ebi.ac.uk/ena/data/view/<accession>). ²Genbank [19]. ³TritrypDB [20]: Kinetoplastid Genomics Resource.

Table 1. Nucleotide sequences information.

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Figure 4.

Evolutionary relationships of Leishmania genus and ancient trypanosomatides by neighbor-joining method using the 834 nucleotide residues corresponding to the small RNA subunit variable region 4 (V4 rRNASSU) encoding gene. Phylogenetic reconstruction was conducted in MEGA4 and the figure corresponds to the optimal tree with the sum of branch length = 0.80513269. The bootstrap values (1000 replicates) are shown next to the branches. Evolutionary distances used to infer the phylogenetic tree are drawn to scale. Leishmania subgenus and Paraleishmania parasites are described at right. Box: Lenrietti classified in subgenus Mundinia.

biological samples. Restriction analysis of the 953 bp *Leishmania* chitinase PCR fragment with *PstI* permitted the identification of medical important species in Latin America where three different *Leishmania* subgenera circulate in animal reservoirs, human, and sandflies [21]. Given the specificity of the *Leishmania* chitinase-encoding gene, the molecular diagnostic method can also be used to identify isolated parasites from biological samples, with high specificity, by restriction analysis and/or sequencing [29]. Also, using the restriction enzyme Dde *I* on the 953 bp chitinase PCR fragment, it is possible to differentiate *L. major* from all others Old World *Leishmania* subgenus species, which is of clinical importance in Oriental TL endemic countries (**Table 2**) [30].

Subgenera	Species	Geographic distribution	Dde I fragments (bp)	Pst I fragments (bp)
Leishmania	L.donovani	Old World	608, 157,188	390, 113, 258, 192
Leishmania	L. infantum	New and Old World	608, 157,188	390, 113, 258, 192
Leishmania	L. turanica	Old World	608, 157,188	390, 113, 258, 192
Leishmania	L. gerbili	Old World	608, 157,188	390, 113, 258, 192
Leishmania	L. major	Old World	58, 550, 79, 78, 188	390, 113, 258, 192
Leishmania	L. mexicana	New World	602, 6, 157, 188	390, 113, 450
Leishmania	L. amazonensis	New World	602, 6, 157, 188	390, 113, 450
Viannia	L. braziliensis	New World	325, 284, 344	503, 450
Sauroleishmania	L. tarantolae	Old World	765, 188	503, 450
Mundinia	L. enrietti	New World	112,292, 180, 25, 344	503, 258, 192

Table 2.

Restriction fragment sizes of the 953 bp chitinase PCR amplicon digested with Dde I and Pst I.

The origins of parasites from *Leishmania* genus and its evolutionary relationships are investigated through phylogenetic reconstructions associated to data on biogeographic dispersion and evolution of their vertebrates and sandflies corresponding hosts, being a matter of discussion on conflicting information [1]. The revision by Akhoundi, et al. 2016, details the three principal theories proposed for Leishmania origin, a Palearctic, Neotropical, and a Neotropical/African/Multiple Origins. The most widely accepted theory regarding the origin of *Leishmania* is the Supercontinent hypothesis, a variation of the Multiple Origins hypothesis. This theory suggests that the Viannia and Leishmania subgenera evolved independently during the separation of South America from Africa. The Supercontinent hypothesis proposes that *Leishmania* originated on Gondwana and evolved from monoxenous parasites [18]. This theory is supported by biogeographic data and animal host migration patterns, and was developed through phylogenetic reconstruction using a large multi-gene dataset (over 200,000 informative sites) [18]. In this work, the phylogenetic reconstruction of Leishmania genus with the 953 bp partial chitinase-encoding gene sequence corroborates the Supercontinent hypothesis, grouping species of each *Leishmania* subgenera and the Paraleishmania group (Figure 4).

The genomic locus of GH18 chitinase-encoding gene is conserved among basal trypanosomatids, including the *B. saltans*. Besides, amino acid sequence comparison studies among GH18 chitinases from trypanosomatids using public genome data-banks revealed 35% of identity of GH18 chitinases of marine protozoa and bacteria to the similar enzyme of *B. saltans*. These results strongly suggest that the GH18 chitinase from Kinetoplastida group derived from a common marine ancestor, harboring the primitive enzyme.

The phylogenetic position of subgenus *Sauroleishmania*, according to the Supercontinent hypothesis, indicates the switch of its *Leishmania* ancestors from mammalian to reptilian hosts [1]. In considering a probable marine environment emergence of the trypanosomatid GH18 chitinase, a highly conserved and unique gene in basal groups, including *Leishmania*, it is possible to explore that the *Sauroleishmania* subgenus could diverge from an ancestor before the rise of mammals, during the transition

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of animals from marine to the terrestrial environment. In this case, *Leishmania*-like parasites could be found in fish and amphibians. Considering the conservation of the chitinase-encoding gene in *Leishmania* group, the diagnostic method developed in this work can be used to investigate this hypothesis directly on biological samples, circumventing the isolation difficulties of unknown *Leishmania*-like parasites.

The barcode gene used for phylogenetic studies in trypanosomatids corresponds to the V4 rRNA SSU. However, for basal groups of trypanosomatids, the chitinase-encoding gene and amino acid sequence presented better results, as demonstrated through comparison between phylogenetic trees generated by Neighbor-joining method for both markers (**Figures 3** and **4**). Furthermore, the 953 bp chitinase fragment can be obtained by PCR directly from biological samples, while the V4 rRNA fragment is obtained preferentially from isolated parasites. These results indicate that the partial sequence of the gene encoding trypanosomatids chitinase can be used as a barcode to investigate the phylogenetic relationships among basal species of the group.

In silico prediction of the biochemical and molecular characteristics of Leishmania chitinase, such as high solubility and exportation of the native protein to the extracellular medium, associated to species-specific sequences, as well as the absence of a similar gene in the genus *Trypanosoma*, indicated its potential use as an antigen in accurate differential serological diagnosis for *Leishmania* species. Thus, in order to produce high amounts of *Leishmania* chitinase from the four different species representing the three major taxonomic groups from the subgenera Viannia (L. braziliensis) and Leishmania (L. amazonensis/L. mexicana and L. infantum), a conventional prokaryotic expression system was initially used. However, after tests under various expression conditions, the proteins remained insoluble, probably due to unfolding, the absence of the predicted N-glycosylation signal, and the formation of inclusion bodies [21]. The homologous chitinase of L. donovani was partially soluble in E. coli only when expressed in fusion with thioredoxin [31], thus indicating that even when using different strategies, the prokaryotic system is not appropriate for producing a correct folded Leishmania chitinase, essential in obtaining an effective antigen for serological testing, for example.

After observing the successful expression of L. major GP63, a glycosylated membrane protein, in insect cells using recombinant baculovirus [32], we obtained several baculovirus constructions of chitinase from the four Leishmania species by utilizing the Bac-to-Bac insect cell expression system. However, despite several attempts, it was found that this was not the case. This suggests that there may be intrinsic molecular characteristics associated with the insolubility of *Leishmania* species chitinase, such as post-translational signals that are incompatible with the organelle machinery of insect cells. Consequently, this may lead to the accumulation of unfolded recombinant proteins in cellular compartments. Investigation by way of *in situ* detection of recombinant proteins from constructions with N and C terminal histidine-tags, using Fluorescein Isothiocyanate (FITC) labeled anti-histidine antibody, revealed their accumulation in the plasma membrane [21]. This was confirmed by Thin Layer Chromatography (TLC) of insect cell membranes infected with L. infantum chitinase baculoviruses previously enriched with recombinant chitinase, and presenting lipid chemical signatures of plasma membrane. Recombinant forms of *Leishmania* species chitinase accumulation in plasma membrane, independent of differential physicochemical characteristics, presuppose the occurrence of a specific molecular feature necessary for precise Leishmania chitinase folding, such as chaperones.

Chaperones play an essential role in the regulation of biological functions by facilitating changes in the conformation of non-native protein. This is achieved

through several mechanisms, including ribosome nascent polypeptide folding, organelle and cellular protein-membrane addressing, and the disassembly of macromolecular aggregates. Biochemical modifications, such as phosphorylation networks and hydrophobic interactions, also contribute to this process [33, 34]. There are several eukaryotically conserved and divergent chaperone families in *Leishmania* parasites that could explain the need for specific chitinase-chaperone interaction in correct folding [35], such as cyclophilin, a specific *L. donovani* chaperone, involved in reversing ADP-dependent inactive aggregates of adenosine kinase under physiological conditions, and a key enzyme in the leishmanial purine salvage pathway [36]. Further evidence of specific chitinase-chaperone interaction comprises the highly conserved *Leishmania* genus chitinase C-terminal ATP-dependent kinase domain (Data in Brief co-publication) since biochemical regulation of several chaperone functions is often associated to ATP-dependent phosphorylation [37]. Further studies of native *Leishmania* chitinase are required to investigate this hypothesis.

Leishmania chitinase is specific to basal groups of trypanosomatids genera, probably derived from an ancestor living in a marine environment, and unique in the human pathogen group. There are no *Leishmania* chitinase or homologous proteins described with a molecular structure associated to biochemical function. Considering biological importance and *Leishmania* genus chitinase specificity, novel molecularbased studies of native protein should shed light on its biochemical function, thereby facilitating its use not only in diagnosis, but also in drug and vaccine designs for controlling and treating leishmaniasis.

6. Conclusions

- Phylogenetic analysis of kinetoplastida chitinase revealed an ancient origin;
- Chitinase phylogeny corroborates the Supercontinent Origins theory for *Leishmania*;
- Chitinase is conserved among basal trypanosomatids and absent in *Trypanosoma* genus;
- Diagnosis based on *Leishmania* chitinase differentiates Old and New World Subgenera;
- Amino acid comparison analysis supports a marine origin for Leishmania chitinase.

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Conflict of interest

The authors declare no conflict of interest.

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Polysaccharides are organic polymers in which the repeating unit consists of monosaccharides. They exist in plants, animals, algae, or microbial worlds. Chitosan is a cationic linear polysaccharide of glucosamine produced from chitin deacetylation in alkaline media. Chitin is obtained from the exoskeleton of shrimps, crabs, and squids. Therefore, chitosan is environmentally benign and biodegradable. This book summarizes different aspects of chitosan and its derived materials, addressing isolation, properties, and applications. This book is intended for academics, professionals, and scientists as well as graduate and undergraduate students without any geographical limitations.

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